



# Lipid microenvironment affects the ability of proteoliposomes harboring TNAP to induce mineralization without nucleators

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## Abstract

Tissue-nonspecific alkaline phosphatase (TNAP), a glycosylphosphatidylinositol-anchored ectoenzyme present on the membrane of matrix vesicles (MVs), hydrolyzes the mineralization inhibitor inorganic pyrophosphate as well as ATP to generate the inorganic phosphate needed for apatite formation. Herein, we used proteoliposomes harboring TNAP as MV biomimetics with or without nucleators of mineral formation (amorphous calcium phosphate and complexes with phosphatidylserine) to assess the role of the MVs' membrane lipid composition on TNAP activity by means of turbidity assay and FTIR analysis. We found that TNAP-proteoliposomes have the ability to induce mineralization even in the absence of mineral nucleators. We also found that the addition of cholesterol or sphingomyelin to TNAP-proteoliposomes composed of 1,2-dipalmitoyl-*sn*-glycero-3-phosphocholine reduced the ability of TNAP to induce biomineralization. Our results suggest that the lipid microenvironment is essential for the induction and propagation of minerals mediated by TNAP.

**Keywords** Biomineralization · Proteoliposome · Alkaline phosphatase · Matrix vesicles · Nucleational core

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## Introduction

During bone formation, osteoblasts or chondrocytes produce and mineralize the extracellular matrix (ECM) [1]. The formation of apatite is initiated in the lumen of matrix vesicles (MVs) and has been described to depend on the presence of a nucleator, the nucleational core (NC), in the lumen of MVs [2, 3]. The components of the NC, the amorphous calcium phosphate (ACP) in association with phosphatidylserine (PS), result in calcium–phosphate–lipid complexes (PS-CPLXs). Further accumulation of Ca<sup>2+</sup> and inorganic phosphate (P<sub>i</sub>) in MVs can lead to the growth of apatite crystals, which then propagate on the ECM and continue to grow onto collagenous fibers, forming bone tissues. Tissue-nonspecific alkaline phosphatase (TNAP, E.C. 3.1.3.1) [4, 5] is a glycosylphosphatidylinositol (GPI)-anchored protein present on the plasma membrane of osteoblasts and chondrocytes, and also on the surface of MVs [6]. TNAP provides P<sub>i</sub> needed for apatite crystallization by hydrolyzing phosphomonoesters [7]. Among various substrates, TNAP catalyzes the hydrolysis of pyrophosphate (PP<sub>i</sub>), a mineralization inhibitor [8], thus facilitating the formation of apatite crystals [5, 9–11]. Adequate local concentrations of extracellular PP<sub>i</sub>, originating from the hydrolysis

of sodium adenosine-5'-triphosphate (ATP) by nucleotide pyrophosphatases/phosphodiesterases (NPPs) or from transport of intracellular  $PP_i$  by the action of progressive ankylosis protein (ANK), and of extracellular  $P_i$  originating from the hydrolysis of  $PP_i$  or of other phosphomonoesters by TNAP are essential to allow normal mineralization [5, 12–14]. The catalytic properties of the GPI-anchored, polidocanol-solubilized and phosphatidylinositol-specific phospholipase C-released TNAP suggest that the location of TNAP on the membrane of MVs can control substrate selectivity in this micro-compartment, suggesting that assays of TNAP bound to MVs or bound to proteoliposomes might be more relevant than assays performed with solubilized enzymes, particularly when studying the hydrolysis of organophosphate substrates [15].

The MVs' membrane contains high content of cholesterol (Chol), sphingomyelin (SM) and PS, resembling the lipid content of rafts in the plasma membrane [16, 17]. The organization of the lipids influences the activity of the proteins present on the rafts, thus the study of the interactions between TNAP and the lipids of the MVs' membrane is essential to understand the initiation and propagation of biomineralization process [18–21].

The ability of proteoliposomes to mimic the function and structure of biomembranes [22–27] makes them a convenient model to determine the factors of MV-mediated mineralization. Other researchers also aim to work with therapeutic mimetic cell to provide a source of cell-like assemblies, which exhibits the core structural or functional properties of their natural counterparts with broad envisioned applications in biomedicine. One most recent example is the generation of microreactors loaded with liposomes harboring TNAP to biomineralization studies [28].

Our research group has standardized a methodology to fabricate biomimetic models of MVs containing TNAP incorporated into lipid microdomains, allowing biochemical and biophysical studies of the biomineralization driven by MVs [22, 29, 30]. Here, we prepared proteoliposomes harboring TNAP composed by 1,2-dipalmitoyl-*sn*-glycero-3-phosphocholine (DPPC), DPPC:Chol (9:1), DPPC:SM (9:1) and DPPC:Chol:SM (8:1:1) (molar ratios) to evaluate how the lipid composition affects the ability of proteoliposomes to form apatite minerals through the activity of TNAP for the biological substrate ATP as the source of  $P_i$  when incubated in a synthetic cartilage lymph (SCL) buffer either in the presence or absence of nucleators.

## Materials and methods

### Production of TNAP

Human TNAP was expressed as previously described [24, 31] and used immediately after detergent removal to avoid aggregation.

### Liposome preparation

DPPC, DPPC:Chol (9:1), DPPC:SM (9:1), and DPPC:Chol:SM (8:1:1) (molar ratios) were prepared in 50 mM Tris-HCl buffer, pH 7.5, containing 2 mM  $MgCl_2$ , to yield a final solution with 10 mg/mL of lipids, as previously described [30].

### Proteoliposome preparation

TNAP (0.02 mg/mL) was incorporated into the liposomes dispersed in a 50 mM Tris-HCl buffer, pH 7.5, containing 2 mM  $MgCl_2$  by direct insertion in a 1:10,000 protein:lipid (molar ratio). The mixture was incubated at 25 °C for 1 h and ultracentrifuged at 100,000×g for 1 h, at 4 °C. The pellet (proteoliposomes) was resuspended to the original volume in the same buffer. The activity of TNAP in both pellet and supernatant was assayed to determine the percentage of TNAP incorporation into liposomes [29]. The protein concentrations were estimated as described by Hartree [32] in the presence of 2% (0.2 g/mL) SDS. Bovine serum albumin was used as a standard.

### Dynamic light scattering measurement

The determination of liposomes and proteoliposomes' size distribution was performed by dynamic light scattering (DLS) as previously described [26]. Average values ( $n=5$ ) of the vesicles' diameters was obtained at 25 °C by unimodal distribution, previously filtered (0.8 mm). DLS measurements were performed throughout the period of use of the samples (7 days) and both diameter and intensity varied less than 5%, when stored at 4 °C. Data were reported as the mean of five measurements of three different vesicles' preparations.

### Enzymatic activity measurements

For ATP hydrolysis, phosphomonoesterase activities were assayed discontinuously by measuring the amount of released  $P_i$  at 37 °C in 50 mM Tris-HCl buffer containing 2 mM  $MgCl_2$ , pH 7.4, as previously described [24, 26]. One enzyme unit (1 U/mg) is defined as the amount of enzyme

hydrolyzing 1.0 nmol of substrate per minute per milligram of protein. Maximum velocity ( $V_{\max}$ ), apparent dissociation constant ( $K_{0.5}$ ), and Hill coefficient ( $n$ ) obtained from substrate hydrolysis were calculated as described [33]. Data were reported as the mean of triplicate measurements of three different enzyme preparations. Statistically significant differences were defined as  $p < 0.05$ .

### Synthesis of ACP and PS-CPLX nucleators

ACP and PS-CPLX nucleators were synthesized as previously described [3, 34, 35].

### Mineralization assays with proteoliposomes

TNAP-proteoliposomes were incubated in SCL buffer in the presence or absence of ACP or PS-CPLX at pH 7.5. SCL contained 2 mM  $\text{Ca}^{2+}$ , 104.5 mM  $\text{Na}^+$ , 133.5 mM  $\text{Cl}^-$ , 63.5 mM sucrose, 16.5 mM Tris, 12.7 mM  $\text{K}^+$ , 5.55 mM glucose, 1.83 mM  $\text{HCO}_3^-$ , and 0.57 mM  $\text{MgSO}_4$  [34]. The assay was accomplished with a saturating ATP concentration (as the  $\text{P}_i$  source) for TNAP activity present in each distinct proteoliposome. Thus, ATP concentrations of 6 mM, 10 mM, 5 mM, and 9 mM were used for the proteoliposomes composed of DPPC, DPPC:Chol (9:1), DPPC:SM (9:1), and DPPC:Chol:SM (8:1:1), respectively. Enzyme-devoid liposomes were used as control. Mineral formation/propagation was measured by turbidity at 340 nm ( $A_{340}$ ) using a multi-well microplate assay as described before [34]. Triplicate samples (280  $\mu\text{L}$ ) were successively distributed into wells of a 96-well microplate. Turbidity measurements were made after brief agitation (10 s) followed by 48 h of incubation at 37 °C, using a Molecular Devices M3 microplate reader (Sunnyvale, CA). The results were normalized according to the protein concentration of each proteoliposome, since different efficacies of TNAP incorporation were obtained depending on the lipid composition of the proteoliposome.

### Fourier-transform infrared spectroscopy

The proteoliposomes were incubated in SCL buffer, as described above, in the absence of nucleators. Then, the samples were placed on the germanium crystal of an attenuated total reflectance (ATR) accessory to assess the chemical groups in the formed minerals by means of FTIR spectroscopy (Shimadzu-IRPrestige-21). The efficacy of mineralization was assessed by calculating the ratio between the areas of the internal reference band of the ester phospholipid ( $\text{C}=\text{O}$ ) at 1740  $\text{cm}^{-1}$  and the band corresponding to the asymmetrical stretching of the  $\text{PO}_4^{3-}$  group at 1032  $\text{cm}^{-1}$ .

### Statistical analysis

All quantitative measurements were carried out 3–5 times and values are reported as mean  $\pm$  standard error of the mean (SEM). Groups were compared with the one-way ANOVA or a two-tailed Student's  $t$  test. A  $p$  value of  $< 0.05$  was considered significant.

### Results

Biomineralization induced by MVs is a complex process in which  $\text{Ca}^{2+}$  and  $\text{P}_i$  accumulate inside MVs and induce nucleation and formation of apatitic minerals [5, 7]. Previously, we have characterized the incorporation of TNAP into proteoliposomes made of DPPC:Chol, DPPC:SM, or DPPC:Chol:SM by differential scanning calorimetry (DSC) [30]. Here, the biophysical characterization of vesicles by DLS (Table 1) showed that the average diameters of the liposomes and proteoliposomes did not alter significantly when compared with the diameter obtained for DPPC-vesicles. Comparing the sizes distribution of liposomes with proteoliposomes harboring TNAP, in the presence of TNAP, no more than 10% in the increase in the vesicles diameter was observed (DPPC-vesicles 6%, DPPC:Chol-vesicles 10%, DPPC:SM-vesicles 6% and DPPC:Chol:SM-vesicles 4%) (Table 1). Comparing the lipid compositions in the presence of cholesterol and sphingomyelin with neat DPPC-vesicles, the diameter of vesicles formed did not alter significantly (maximum variation of 15%), despite the diverse lipid compositions studied. Low polydispersity index (PI) values were obtained for the different lipids compositions, indicating that the samples were nearly monodisperse in size (Table 1).

The kinetic behavior of membrane-anchored TNAP depends on the lipid composition of the liposomes [26]. Thus, we have investigated the effect of the liposome composition on the kinetic parameters of the reconstituted TNAP

**Table 1** Biophysical characterization by DLS of liposomes and TNAP-proteoliposomes composed by different lipid compositions

Liposome/proteoliposome composition		Diameter (nm)	Polydispersity index (PI)
Lipid (molar ratio)	TNAP		
DPPC	–	115.0 $\pm$ 0.4	0.08 $\pm$ 0.01
DPPC	+	121.1 $\pm$ 5.9	0.15 $\pm$ 0.05
DPPC:Chol (9:1)	–	121.9 $\pm$ 0.5	0.09 $\pm$ 0.03
DPPC:Chol (9:1)	+	133.8 $\pm$ 8.8	0.12 $\pm$ 0.05
DPPC:SM (9:1)	–	127.1 $\pm$ 1.0	0.09 $\pm$ 0.03
DPPC:SM (9:1)	+	134.5 $\pm$ 9.1	0.09 $\pm$ 0.06
DPPC:Chol:SM (8:1:1)	–	133.4 $\pm$ 0.8	0.04 $\pm$ 0.03
DPPC:Chol:SM (8:1:1)	+	138.8 $\pm$ 21.9	0.12 $\pm$ 0.11

1:10,000 protein:lipid (molar ratio), to determine the saturating ATP concentration for each lipidic composition, as shown by the arrows in Fig. 1. For DPPC-proteoliposomes harboring TNAP,  $V_{\max}$  values were observed to follow the order DPPC:Chol (9:1) > DPPC:SM (9:1) > DPPC:Chol:SM (8:1:1) proteoliposomes (Table 2). Only positive cooperativity ( $n$ ) was observed for ATP hydrolysis for all tested compositions (Table 2). The catalytic efficiencies of DPPC, DPPC:Chol (9:1), and DPPC:SM (9:1) showed the same order of magnitude (from  $5.7 \times 10^2$  to  $3.5 \times 10^2 \text{ M}^{-1} \text{ s}^{-1}$ ) and a lower value was obtained for the ternary composition ( $1.1 \times 10^1 \text{ M}^{-1} \text{ s}^{-1}$ ) (Table 2).

We then assessed the ability of these proteoliposomes to induce/propagate biomineralization in a SCL buffer with a saturating ATP concentration (as a source of  $\text{P}_i$ ) for TNAP activity present in each distinct proteoliposome, and in the presence or absence of ACP or PS-CPLX as nucleators. Thus, ATP concentrations of 6 mM, 10 mM, 5 mM, and 9 mM were used for the proteoliposomes composed of DPPC, DPPC:Chol (9:1), DPPC:SM (9:1), and

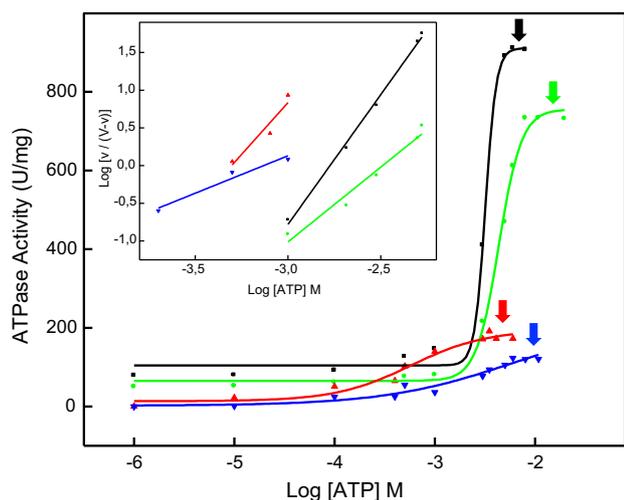
DPPC:Chol:SM (8:1:1), respectively, as determined in Fig. 1.

We found that DPPC-proteoliposomes induced mineral formation, as probed by turbidity assay, when incubated for 48 h in SCL buffer containing saturating concentration of ATP as TNAP substrate (Fig. 2). DPPC:Chol (9:1), DPPC:SM (9:1), and DPPC:Chol:SM (8:1:1) (Fig. 2) proteoliposomes harboring TNAP produced less mineral than DPPC-proteoliposomes. DPPC:Chol:SM-proteoliposomes induced slightly less minerals than those produced by DPPC:SM-proteoliposomes, however, higher than those produced by DPPC:Chol-proteoliposomes.

Notably, the presence or absence of a nucleator (ACP or PS-CPLX) did not significantly affect the production of minerals in any of the proteoliposomes.

The differences in mineralization efficiency among the distinct types of proteoliposomes were further assessed by means of ATR-FTIR spectroscopy (Fig. 3). Indeed, turbidity did not provide the nature of mineral formed, therefore we used IR spectroscopy to determine the type of minerals. This analysis was performed in the absence of nucleator. All the recorded ATR-FTIR spectra showed broad bands at  $3470\text{--}3410 \text{ cm}^{-1}$  that were assigned to the O–H stretching vibrations of water [36]. The sharp and strong absorption bands at  $2922$  and  $2852 \text{ cm}^{-1}$  were assigned to the asymmetric and symmetric, respectively, C–H stretching vibrations of the  $\text{CH}_2$  groups of acyl chains. The band of the ester carbonyl C=O group of phospholipids was located at  $1740 \text{ cm}^{-1}$ , whereas the C–H scissoring band was at  $1470 \text{ cm}^{-1}$ . The C–O vibration characteristic of fatty acid esters generated a band at  $1234 \text{ cm}^{-1}$ , whereas in the  $1090\text{--}960 \text{ cm}^{-1}$  regions were located the broad absorption bands arising from the P–O stretches of both organic and inorganic phosphate [37]. The latter bands confirmed that a phosphate mineral phase arose after incubating the vesicles in SCL buffer with ATP, validating that the proteoliposomes were able to induce mineralization even in the absence of a nucleator.

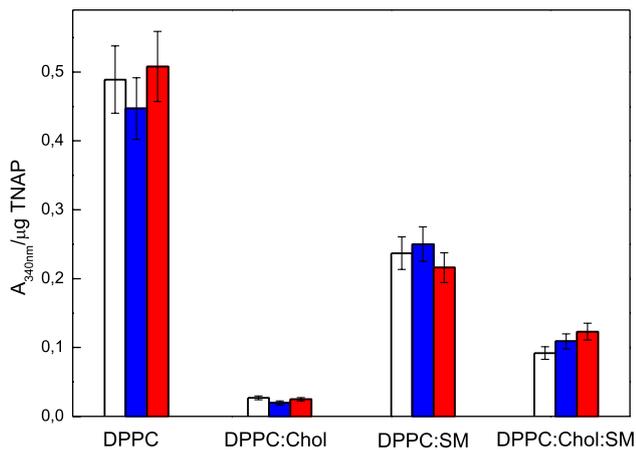
To get more quantitative information about the difference in the apatite minerals produced by the different types of proteoliposomes, we calculated the ratios between the area of the FTIR band relative to the apatite minerals at  $\sim 1032 \text{ cm}^{-1}$  and that of the band relative to carbonyl group of the phospholipids at  $\sim 1740 \text{ cm}^{-1}$  (Table 3). These



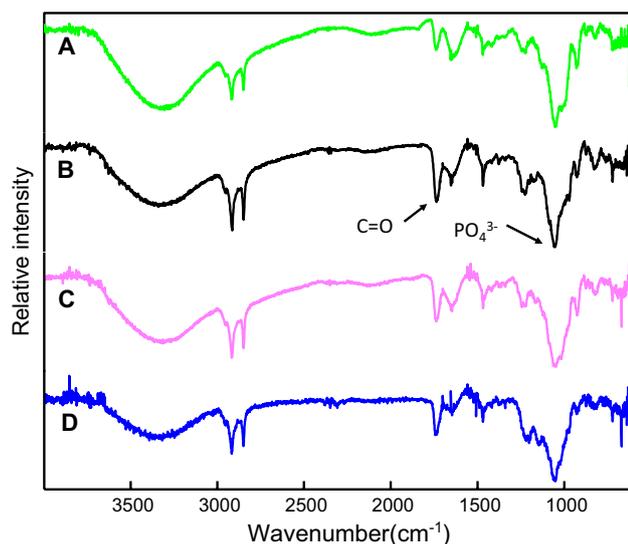
**Fig. 1** Achievement of  $V_{\max}$  for the hydrolysis of ATP by TNAP-proteoliposomes composed of different lipid compositions: (black) DPPC, (green) DPPC:Chol (9:1), (red) DPPC:SM (9:1) and (blue) DPPC:Chol:SM (8:1:1) (molar ratios). Arrows indicate the saturating ATP concentration for each distinct proteoliposome. Inset: representation to determine Hill coefficient ( $n$ ). Data were reported as the mean of triplicate measurements of three different enzyme preparations

**Table 2** Kinetic parameters for the hydrolysis of ATP by TNAP reconstituted in liposomes 1:10,000 protein:lipid (molar ratio) of different lipid compositions

Kinetic parameters	TNAP-proteoliposomes			
	DPPC	DPPC:Chol	DPPC:SM	DPPC:Chol:SM
$V_{\max}$ (U/mg)	$910.0 \pm 10.6$	$732.8 \pm 15.3$	$191.2 \pm 12.3$	$122.8 \pm 10.3$
$K_{0.5}$ (mM)	$3.18 \pm 0.02$	$2.75 \pm 0.06$	$0.32 \pm 0.01$	$2.22 \pm 0.04$
$n$	$3.42 \pm 0.01$	$1.91 \pm 0.07$	$2.74 \pm 0.02$	$0.99 \pm 0.09$
$k_{\text{cat}}/K_{0.5}$ ( $\text{M}^{-1} \text{ s}^{-1}$ )	$5.7 \times 10^2$	$5.3 \times 10^2$	$3.5 \times 10^2$	$1.1 \times 10^1$



**Fig. 2** Effect of TNAP-proteoliposomes composed of DPPC, DPPC:Chol (9:1), DPPC:SM (9:1) and DPPC:Chol:SM (8:1:1) (molar ratios), on mineral propagation in the absence of nucleator (blank) and in the presence of ACP (blue) or PS-CPLX-seeded SCL (red), at pH 7.5. The assay was accomplished in SCL buffer with a saturating ATP concentration for TNAP activity for each distinct proteoliposome. ATP concentrations of 6 mM, 10 mM, 5 mM, and 9 mM were used for the proteoliposomes composed of DPPC, DPPC:Chol, DPPC:SM and DPPC:Chol:SM, respectively, as indicated by arrows in Fig. 1. Enzyme-devoid liposomes were used as control and bars show the increment in absorbencies after 48 h of incubation at 37 °C. All results are expressed as mean  $\pm$  SEM.  $P < 0.05$



**Fig. 3** FTIR spectra of the minerals obtained from the mineralization assays under incubation of TNAP-proteoliposomes composed of (green) DPPC, (black) DPPC:Chol (9:1), (pink) DPPC:SM (9:1) and (blue) DPPC:Chol:SM (8:1:1) (molar ratios), in SCL, at 37 °C, at pH 7.5, in the absence of nucleators. ATP concentrations of 6 mM, 10 mM, 5 mM, and 9 mM were used for the proteoliposomes composed of DPPC, DPPC:Chol, DPPC:SM and DPPC:Chol:SM, respectively, as indicated by arrows in Fig. 1. Mineralization was followed by the differences in the ratio between the areas of the internal reference band of the phospholipid (C=O) at 1740  $\text{cm}^{-1}$  and the band corresponding to the asymmetrical stretching of the  $\text{PO}_4^{3-}$  group at 1032  $\text{cm}^{-1}$

**Table 3** Ratios between the areas of the bands at 1032 ( $\text{PO}_4^{3-}$ ) and 1740 (C=O)  $\text{cm}^{-1}$  obtained by ATR-FTIR for the different proteoliposomes samples

Proteoliposomes composition	$\text{PO}_4^{3-}/\text{C=O}$
DPPC	$3.2 \pm 0.1$
DPPC:Chol (9:1)	$1.8 \pm 0.1$
DPPC:SM (9:1)	$2.2 \pm 0.1$
DPPC:Chol:SM (8:1:1)	$1.9 \pm 0.1$

data showed that DPPC-proteoliposomes harboring TNAP without Chol induced the formation of minerals more effectively than the vesicles containing this sterol and that the ternary system produced more minerals than the binary systems, thus validating the results obtained through the turbidity assay (Fig. 3b).

## Discussion

The lipid composition of MVs is different from that of the plasma membrane and from the total lipid composition in the whole cell [38, 39]. The main differences are a greater amount of PS (3.2–4.5 fold), SM (2.2–2.5 fold), and total lysophospholipids (1.8–2.2 fold), and a smaller amount of PC (0.6–0.7 fold) in MVs with respect to whole cells. Additionally, MVs are enriched in Chol and depleted in triacylglycerols compared to the plasma membrane and whole cells [20, 39]. The activity of membrane enzymes is affected by the lipid microenvironment [23, 30, 40–42]. We have previously described that an increase in the complexity of the proteoliposomes decreased the activity of GPI-anchored TNAP incorporated in the membrane [30]. Herein, we investigated the effects of the lipid composition on the ability of proteoliposomes to produce apatite minerals by means of turbidity assay. We found that proteoliposomes produced minerals in the order DPPC > DPPC:SM (9:1) > DPPC:Chol:SM (8:1:1) > DPPC:Chol (9:1) proteoliposomes, probably due to the lipid phase behavior of the membrane microenvironments [30, 40], which can influence differently the  $V_{\text{max}}$  of ATP hydrolysis and catalytic efficiency of TNAP, as observed in Table 2. Studies regarding biomineralization comparing other sources of  $\text{P}_i$  for TNAP hydrolysis when incorporated in proteoliposomes are need to be performed. However, the scenario with  $\text{PP}_i$ , for example, is even more complex than ATP. Beyond substrate,  $\text{PP}_i$  is a potent inhibitor of TNAP activity and can also suppress hydroxyapatite crystal formation and propagation acting as a potent calcification inhibitor in biologic fluids [4].

Chol alters the physical properties and lateral organization of the plasma membrane by increasing the thickness and decreasing the permeability of the phospholipid bilayer

[29, 30, 40, 43–45]. Addition of Chol decreases the area occupied per molecule in lipid bilayers, which increases lipid/lipid interactions and induces a separation of membrane lipids into a fluid liquid crystalline phase ( $L\alpha$ ) and a liquid ordered phase ( $L_o$ ) enriched in Chol and sphingolipids [17, 30, 40, 43]. The presence of  $L_o$  microdomains in proteoliposomes may explain the lower levels of minerals generated by Chol-containing proteoliposomes compared to Chol-devoid ones (Fig. 2).

The DPPC:SM membranes form distinct types of clusters than DPPC:Chol membranes [44, 45]. In particular, DPPC:SM vesicles did not display lateral phase segregation on the membrane as observed on DPPC liposomes [30], which may explain the higher levels of mineral produced by proteoliposomes made of DPPC and DPPC:SM (9:1) when compared to those made in the presence of Chol (DPPC:Chol and DPPC:Chol:SM) (Fig. 2). Finally, the insertion of SM in the DPPC:Chol system stabilizes the membrane through hydrogen bridges between the hydrocarbon chains, which may explain the increased levels of mineral formation observed in DPPC:Chol:SM (8:1:1) proteoliposomes compared to DPPC:Chol (9:1) ones (Fig. 2).

The infrared spectra of all proteoliposomes displayed the band relative to the apatite minerals centered at  $\sim 1032\text{ cm}^{-1}$ , which confirmed that minerals were produced by all the lipidic systems. Additionally, the values of the ratio between the area of the apatite band at  $\sim 1032\text{ cm}^{-1}$  and that of the phospholipid-specific band at  $\sim 1740\text{ cm}^{-1}$  (ester carbonyl C=O) for the tested proteoliposomes was higher for pure DPPC-TNAP compared to the proteoliposomes containing Chol and SM [46]. This result further validated that the proteoliposomes without Chol induced mineral formation more efficiently than those containing Chol.

In conclusion, we found that proteoliposomes harboring TNAP have the ability to induce mineralization after 48 h of incubation under saturating ATP concentrations as the source of  $P_i$ , even in absence of a nucleator. Additionally, we found that the presence of ordered domains due to the addition of Chol to DPPC or DPPC:SM-proteoliposomes decreased their ability to produce minerals. Our results indicated that in addition to the essential components required for the mineralization, such as enzymes, substrates, ions and lipids, the degree of packaging of lipid bilayers can affect the phosphohydrolytic activity and catalytic efficiency of TNAP in the bilayers, as well as affecting the supersaturation conditions necessary to precipitate apatite crystals. The diameter of vesicles formed did not alter significantly, despite the diverse lipid compositions studied. Thus, the physical properties and the lateral-phase organization of lipids in proteoliposomes are relevant to regulate the apatite propagation mediated by TNAP function during mineralization.

In the future, these proteoliposomes may act as nanodelivery systems mimicking the MVs' function and thus

facilitate the bone fractures' regeneration, help the dental implant fixation or decrease the time for patient recovery.

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## Compliance with ethical standards

**Conflict of interest** The authors declare that they have no conflict of interest.

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