Research article

Silicon alleviates the impairments of iron toxicity on the rice photosynthetic performance via alterations in leaf diffusive conductance with minimal impacts on carbon metabolism

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1. Introduction

Iron (Fe) is a micronutrient that plays a crucial role in plant metabolism, as denoted by its functions in redox reactions of photosynthesis, respiration and nitrogen assimilation (Ricachenksy et al., 2010; Vigani et al., 2013). However, when in excess, Fe is associated with increased production of reactive oxygen species, via Fenton and Haber-Weiss reaction (Onaga et al., 2016), which ultimately leads to morphophysiological disorders, including cell damages with degradation of cell membranes (Pereira et al., 2013), nutritional disorders (Müller et al., 2015), iron plaque formation in roots (Pereira et al., 2014; Pinto et al., 2016) and decreased photosynthetic performance, thus impairing growth and crop yields (Audebert and Fofana, 2009; Stein et al., 2009). Disruption of carbon metabolism is due mostly to feedback inhibition by ATP and NADPH which are not properly used in the Calvin-Benson cycle under Fe excess (Siedlecka et al., 1997).

Iron toxicity is a major mineral disorder in lowland rice production (Audebert and Fofana, 2009) due to excess ferrous iron (Fe²⁺) formation under acid and anoxic soil conditions. This toxicity can be direct, caused by excessive metal accumulation, or indirect, due to decreased uptake of nutrients such as P, K, Ca and Mg (Silveira et al., 2007; Stein et al., 2009; Müller et al., 2015). Regardless, Fe toxicity has been shown to disrupt rice photosynthetic activity coupled with decreased productivity, which depends on the tolerance of the cultivar, severity and time of stress imposition (Stein et al., 2009, 2014). Decreases in the net CO₂ assimilation rate (A) under Fe toxicity have been attributed to impairments in photochemical reactions as well as to stomatal and non-stomatal limitations of photosynthesis (Pereira et al., 2013; Pinto et al., 2016). In addition to stomatal conductance (gₛ), which is related to the diffusion of CO₂ from the surrounding atmosphere to the intercellular air spaces, the photosynthetic capacity is also influenced by mesophyll conductance (gₘₐₜ), which in turn is related to the diffusion of CO₂ from
Fig. 1. The effects of silicon, Si (0 or 2 mM: Si or + Si, respectively), and iron, Fe (25 µM or 5 mM: Fe and + Fe, respectively), on the root and leaf concentrations of Si (A–D) and Fe (E–H) of two rice cultivars [cv. ‘EPAGRI-109’ and cv. ‘BR-IRGA-409’, respectively tolerant and sensitive to Fe excess] grown in nutrient solutions. The measurements were performed at 42 days after applying Fe excess. n = 5 ± SE.
Table 1

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The results (significance: * = not significant, **P < 0.05, ***P < 0.01, ****P < 0.001) of ANOVA for the effects of silicon (Si), iron (Fe) and cultivar (Cv) and their interactions on root and leaf concentrations of Si and Fe, net CO2 assimilation rate (A), stomatal conductance (gs), mesophyll conductance (gms), stomatal (SL), mesophyll (ML) and biochemical (BL) limitations of photosynthesis, maximum rate of RuBisCO carboxylation on a chloroplast basis (Vmax), maximum rate of carboxylation limited by electron transport (Jmax), ratio of electron transport rate devoted to oxygenation/carboxylation (J/Jc), variable-to-maximum Chl fluorescence ratio (Fv/Fm), efficiency of the capture of excitation energy by open photosystem II reaction centres (Fv/Fl), and photochemical quenching coefficient (qP), nucleotide ratios, carbohydrates, malate, fumarate and proteins. The results are presented in the form of y/x, i.e. at 8 or 15 days after Fe additions. When a single result is presented, it represents the same significance for both days.

the intercellular spaces to the carboxylation sites in the chloroplast stroma (Flexas et al., 2012). So far, the effects of Fe toxicity on gms are virtually unknown, although the effects of other toxic metals on gms have been reported (Sagardoy et al., 2010; Velikova et al., 2011; Sanglard et al., 2014).

Although silicon (Si) is not recognized as an essential element for plant growth, there seems to be no doubts that it can be beneficial in protecting plants against stresses, biotic and abiotic, even though the underlying mechanisms affording such protection remain unresolved (Debona et al., 2017; Coskun et al., 2019). Some negative effects associated with metal/metalloid toxicity, including e.g. aluminium (Liang et al., 2001), arsenic (Sanglard et al., 2016), boron (Gunes et al., 2007), cadmium (Shi et al., 2010), manganese (Li et al., 2012) and zinc (Neumann and zur Nieden, 2001), have been alleviated or prevented by proper Si fertilization. In the case of Fe toxicity in particular, some information suggests that Si can mitigate the effects of toxicity on rice via decreased Fe concentrations in both leaf and root tissues, increased activity of the antioxidant system with concomitant decreases in lipid peroxidation, which ultimately lead to a lower impact of Fe excess on the growth of plants amended with Si (Chalmardi et al., 2014; Dufey et al., 2014). Si has also been reported to reduce the Fe plaque formation in roots of rice plants challenged with Fe excess (Bu et al., 2012). Whether, or not, Si could prevent Fe toxicity on the photosynthetic performance and carbon metabolism remains elusive, however.

We here hypothesized that Si mitigates the effects of Fe toxicity on the photosynthetic performance and carbon metabolism of rice leaves, particularly in cultivars susceptible to Fe toxicity. To test this hypothesis we designed a greenhouse experiment using two rice cultivars with contrasting abilities to tolerate Fe toxicity; we challenged these cultivars with Fe excess and amended them with Si. Specifically, the following questions were addressed: (i) could silicon fertilization attenuate the Fe toxicity on the rice photosynthetic machinery? (ii) would Si be able to favor a metabolic homeostasis under conditions of Fe excess?

2. Materials and methods

2.1. Plant material, growth conditions and experimental design

The experiments were carried out in Viçosa (20°45’S, 42°54’W, 650 m above sea level) in south-eastern Brazil. Two lowland rice cultivars, BR-IRGA-409 and EPAGRI-109, respectively sensitive and tolerant to Fe excess (Stein et al., 2009, 2014) were used. Details of seed germination and early seedling growth have been described elsewhere (Dallagnol et al., 2011). The plants were grown in plastic pots (5 L volume) in a greenhouse with controlled air temperature (30/25 ± 2°C (day/night)) and naturally fluctuating photosynthetic photon flux density (PPFD). The maximum PPFD values that were registered inside the greenhouse (approx. 70% of light transmittance) were c. 1500 μmol photons m−2 s−1. The plants were grown in non aerated culture solutions (pH adjusted daily to 5.0), prepared based on Hoagland and Arnon (1950), and included the following nutrients: 1.0 mM KNO3; 0.25 mM NH4H2PO4; 0.1 mM NH4Cl; 0.5 mM MgSO4·7H2O; 1.0 mM Ca(NO3)2; 0.30 μM CuSO4·5H2O; 0.33 μM ZnSO4·7H2O; 11.5 μM H2BO3; 3.5 μM MnCl2·4H2O; 0.1 μM (NH4)6Mo7O24·4H2O; 25 μM FeSO4·7H2O and 25 μM EDTA bisodic. Si was supplied (2 mM) or not (0 mM) over the entire experiment as monosilicic acid, which was prepared by passing potassium silicate through cation-exchange resin (Amberlite IR-1208, H+ form; Sigma Aldrich, São Paulo, Brazil). Additionally, plants were divided into two groups: one received regular Fe supply (25 μM in the form of Fe-EDTA (FeSO4)) over the course of the experiment (control plants), and in the other group Fe was also supplied with 25 μM of Fe-EDTA until 45 days after transplanting, and then Fe supply was suddenly increased to 5 mM (Fe excess). This high Fe concentration corresponds to typical environmental conditions of Fe excess in relevant Brazilian rice producing areas (Stein et al., 2014). All the samplings and measurements were performed at 8 or 15 days after Fe excess additions (DAA). For biochemical analyses, leaf tissues were collected at midday, flash frozen in liquid nitrogen, and subsequently stored at −80°C until required.

2.2. Fe and Si contents

Root and leaf tissues were previously washed in deionized water. Root materials were subsequently washed using a dithio-citrate-bicarbonate solution (DCB) for removing the Fe that was adsorbed to the tissues. This solution (1/4 strength), made with 40 mL of 0.3 M sodium citrate, 5 mL of 1 M sodium bicarbonate and 3 g of sodium dithionite (Taylor and Crowder, 1983), was prepared with chemicals (minimum purity of 99.5%) that were obtained from Sigma-Aldrich or Vetec (São Paulo, Brazil). Root tissues were maintained under constant shaking in DCB solutions during 3 h. Both root and leaf tissues were then oven-dried at 60°C until constant mass and finely ground from which Fe was extracted by nitropersilchior digestion at 95°C; Fe contents were subsequently determined using a Spectra AA 220 Fast Sequential AAS atomic absorption spectrometer (Varian, St. Clara, CA, USA) following Malavolta et al. (1989).

For Si analyses, plant tissues were washed in deionized water and then oven-dried at 60°C until constant mass, after which they were ground to pass through a 40-mesh screen with a Thomas Wiley mill (Thomas Scientific, Swedishos, NJ, USA). Si contents were colorimetrically determined on 0.1 g of dried and alkali-digested tissues (Dallagnol et al., 2011).
2.3. Photosynthetic measurements

The leaf gas exchange parameters were assessed in situ in combination with measurements of Chl a fluorescence emission kinetics using two cross-calibrated open-flow gas exchange analysers (LI-6400XT, LI-COR, Lincoln, NE, USA) furnished with integrated fluorescence chamber heads (LI-6400-40, LI-COR Inc.). The net CO₂ assimilation rate (A), stomatal conductance to water vapor (gₛ) and internal CO₂ concentration (Cᵢ) were measured on attached, completely expanded leaves from 10:00 to 14:00 h (solar time), under artificial PPFD, i.e., 1000 μmol m⁻² s⁻¹ at the leaf level and 400 μmol CO₂ mol⁻¹ air. All measurements were performed at 25 °C, and the vapor pressure deficit was maintained at c. 1.0 kPa; the amount of blue light was fixed to 10% of PPFD to optimize the stomatal aperture.

Segments from attached leaves were darkened for at least 30 min and then illuminated with weak modulated measuring beams (0.03 μmol m⁻² s⁻¹) to obtain the initial fluorescence (F₀). Saturating white light pulses of 8000 μmol photons m⁻² s⁻¹ were employed for 0.8 s to warrant maximum Chl a fluorescence emissions (Fₘ). From which the variable-to-maximum Chl a fluorescence ratio, Fᵥ/Fₘ = [Fₘ – F₀]/Fₘ, was calculated. In light-adapted leaves, the steady-state fluorescence yield (Fₛ) was measured after computing the gas exchange variables. Subsequently, a saturating white light pulse (8000 μmol m⁻² s⁻¹; 0.8 s) was applied to achieve the light-adapted maximum fluorescence (Fₘ'). The actinic light was then turned off, and far-red illumination was applied (2 μmol m⁻² s⁻¹) to measure the light-adapted initial fluorescence (F₀'). The values of these variables were used to estimate the photochemical quenching coefficient (qₚ) and the capture efficiency of excitation energy by open photosystem (PS) II reaction centres (Fᵥ'/Fₘ') (Logan et al., 2007). The actual PSII photochemical efficiency (φₚₛₛ) was estimated as φₚₛₛ = (Fₘ' – Fₛ)/Fₘ', as detailed elsewhere (Genty et al., 1989). The apparent electron transport rate (J) was estimated as J = φₚₛₛ * PPFD * β * α where β is a parameter associated with the partitioning of absorbed quanta between PS II and
PS I and α is the leaf absorptance (Genty et al., 1989). The product of β and α was computed following the procedures of Valentini et al. (1995) under non-photorespiratory conditions.

The mitochondrial respiration rate in the light (R_l) was determined according to Martins et al. (2013) as the value that forces the intercept of the plot A versus C_i–C_o (chloroplastic CO₂ concentration) to be zero using points in the C_i range below 300 μmol mol⁻¹ air (RuBisCO limitation).

The partitioning of electrons between photosynthesis (J_c) and photosynthesis (J_n) was accomplished using the values of J_c, A and R_n, employing the equations outlined by Valentini et al. (1995) and Long and Bernacchi (2003).

To obtain in situ A/C_i curves, 4–5 plants per treatment were used.

These curves were initiated at an ambient [CO₂] (C_a) of 400 μmol mol⁻¹ air under a saturating PPFD of 1000 μmol m⁻² s⁻¹. After reaching a steady state condition, C_a was progressively decreased to 50 μmol mol⁻¹ air after which C_a was returned to 400 μmol mol⁻¹ air to reestablish the original A. Next, C_a was increased stepwise to 1600 μmol mol⁻¹ air. All of the gas exchange data were corrected for the leakage of CO₂ into and out of the leaf chamber of the LI-6400 as detailed in Rodeghiero et al. (2007).

Calculations of both C_a and mesophyll conductance to CO₂ (g_m) were performed using gas exchange and Chl a fluorescence variables according to the procedures outlined by Harley et al. (1992), and detailed in Detmann et al. (2012). The g_m values were used to convert A–C_i into A–C_a curves. These curves were then used to calculate the maximum carboxylation rate on a C_a basis (V_cmax) and the maximum carboxylation rate limited by electron transport (J_cmax) by fitting the mechanistic model of CO₂ assimilation as proposed by Farquhar et al. (1980) and detailed elsewhere (Martins et al., 2014); for these purposes, the C_a-based temperature dependence of the RuBisCO kinetic parameters were used (Bernacchi et al., 2002). The photosynthetic variables V_cmax, J_cmax and g_m were thereupon normalized to 25 °C following the temperature response equations described in Sharkey et al. (2007).

The overall photosynthetic limitations were fractionated into their functional components (stomatal, mesophyll and biochemical constraints) following the approach proposed by Grassi and Magnani (2005) and described in detail in Sanglard et al. (2014).

2.4. Metabolites

A 10-mg sample of lyophilized leaf tissue was finely ground and added to pure methanol, and the mixture was kept at 70 °C for 30 min. After an initial centrifugation (13,000 × g, 5 min), the pellets were extracted further two times using pure methanol. Supernatants were retained, combined, and stored at −20 °C for soluble sugar determinations. The concentrations of glucose, fructose and sucrose in the supernatants were measured through the reduction of NAD+ by glucose-6-P dehydrogenase following the sequential addition of hexokinase, phosphoglucose isomerase and invertase, according to Trethewey et al. (1998); all of these chemicals were obtained from Sigma-Aldrich (São Paulo, Brazil), Reduction of NAD+ was continuously registered at 340 nm using an ELISA reader (Tunable Microplate Reader, VERSAmax, Sunnyvale, USA). The pellet was suspended in KOH, incubated at 95 °C for 1 h, neutralized with acetic acid and centrifuged at 16,000 × g for 10 min. Starch was then hydrolyzed in 100 mM citrate buffer at pH 4.6 containing amyloglucosidase and α-amylase (Trethewey et al., 1998), and the released glucose was assessed as described above. Proteins were also quantified from the methanol-insoluble pellet using bovine serum albumin as a standard (Bradford method). Further details have described elsewhere (Praxedes et al., 2006; Ronchi et al., 2006). The concentrations of malate and fumarate (from the methanol-soluble phase) were exactly determined as described elsewhere (Nunes-Nesi et al., 2007), using the above quoted ELISA reader.

Nucleotide extraction was performed by grinding lyophilized leaf materials in liquid nitrogen after which the appropriate extraction buffers were promptly added. First, 10 mg of lyophilized material were divided into two approximately equal samples to which 0.1 M HClO₄ or 0.1 M KOH was added. Two extracts were then obtained, one acidic (that was used for extracting oxidized NAD(P)⁺), and another alkaline (that was used for extracting reduced NAD(P)H). Extracts were further neutralized by adding 0.1 M KOH 0.2 M Tris (pH = 8.4) on the acidic extract or 0.1 M HClO₄ 0.2 M Tris (pH = 8.4) in the alkaline extract. NAD(H) content was assessed enzymatically in a medium containing 12 mM NaN₂EDTA, 0.3 mM PES, 1.5 mM ethanol and alcohol dehydrogenase. NAD(P)H was enzymatically quantified in a reaction medium containing 12 mM NaN₂EDTA, 0.3 mM PMS, 1.8 mM MTT, 9 mM glucose-6-phosphate and the enzyme glucose-6-phosphate dehydrogenase. NADP(H) was enzymatically quantified in a reaction medium containing 12 mM NaN₂EDTA, 0.3 mM PMS, 1.8 mM MTT, 9 mM glucose-6-phosphate and the enzyme glucose-6-phosphate dehydrogenase.
dehydrogenase. Further details are given in Gibon et al. (2004).

2.5. Enzyme activities

Enzyme extracts were obtained from frozen leaf tissues as exactly described by Nunes-Nesi et al. (2007). Activities of citrate synthase, pyruvate kinase (PK), phosphoglycerate kinase (PGK) and ATP-dependent phosphofructokinase (ATP-PFK) were determined as described by Gibon et al. (2004); NAD-dependent malate dehydrogenase (NAD-MDH) and NADP-dependent isocitrate dehydrogenase (NADP-IDH) as by Jenner et al. (2001); aldolase, enolase, NAD-dependent glyceraldehyde-3-phosphate dehydrogenase (NAD-GAPDH) and triose-phosphate isomerase (TPI) as by Fernie et al. (2001); and acid and alkaline invertases and sucrose synthase (Susy) as reported in Praxedes et al. (2006). RuBisCO activity was assayed as described in Ronchi et al. (2006).

2.6. Statistical analysis

The experiment was arranged in a completely randomized design following a 2 × 2 × 2 factorial scheme (two genotypes x two Si levels x two Fe levels), with six plants in individual pots (48 in total) per each treatment serving as conditional replicates. The data were submitted to a three-way ANOVA with all main factors computed as fixed factors. The MIXED procedure of SAS (version 9.1) was used at an α = 0.05. When any interaction was significant, the MIXED Slice statement was employed to analyse the dependency effect of the factors. Regression analysis was used to evaluate relationships between some selected traits.
3. Results

3.1. Si supply increases Si levels and decreases Fe levels, especially in the shoot

Regardless of cultivar, Si levels were higher in +Si plants, particularly in leaves relative to roots (Fig. 1A–D; Table 1). As expected, +Fe plants displayed remarkably higher Fe levels than their −Fe counterparts, as particularly noted in the Fe-sensitive cultivar (Fig. 1E–H; Table 1). In both cultivars, Fe contents increased over time. Presence of Si led to significantly decreased Fe levels in both leaves and roots, independently of cultivar and sampling dates (Fig. 1E–H; Table 1), and these decreases were more marked in the Fe-sensitive cultivar. Notably, the mitigating effects of Si resulted in greater reductions of Fe content in leaves than in roots (Fig. 1).

3.2. Si partially reverses the impairments of Fe excess on the photosynthetic rates by fundamentally affecting the leaf diffusive conductance with minimal changes in the photochemical and biochemical capacity

Fe toxicity led to reductions in $A$ in accordance with decreases in both $g_s$ and $g_m$ (Fig. 2; Table 1). These reductions were more evident in the −Si plants, particularly in the Fe-sensitive cultivar. In fact, $A$, as well as $g_s$ and $g_m$, correlated significantly ($r^2 \geq 0.83$) with leaf Fe contents (Fig. 3). The decreases in $A$ due to Fe stress were partially reversed by Si, as observed at 8 or 15 DAA in the Fe-sensitive cultivar and also at 8 DAA in the Fe-tolerant cultivar (Fig. 2A and B; significant Si x Fe interaction, Table 1). Similar results were observed for $g_s$ (Fig. 2C and D) and $g_m$ (Fig. 2E and F). Interestingly, $A$ correlated significantly with $g_s$ ($r^2 = 0.96$) and $g_m$ ($r^2 = 0.83$); also, $g_s$ and $g_m$ were linearly correlated to each other ($r^2 = 0.82$) (Supplementary Fig. 1).

The applied treatments affected minimally, if at all, the photochemical parameters ($F_v/F_m$, $F_s/F_m$, and $q_p$) (Supplementary Fig. 2;
NADPH/NADP⁺ (B) of two rice cultivars [cv. ‘EPAGRI-109’ and cv. ‘BR-IRGA-409’, respectively tolerant and sensitive to Fe excess] grown in nutrient solutions. n = 5 ± SE.

Table 1) as well as $V_{\text{max}}$ (Fig. 4A and B) and $J_{\text{max}}$ (Fig. 4C-D). In contrast, there were effects of Si and Fe (significant Fe x Si interaction; Table 1) on the $J_{\text{v}}/J_{e}$ ratio which increased more markedly in the Fe-sensitive cultivar under Fe excess; however, these increases were less pronounced in presence of Si (Fig. 4E and F). Taken together, these data are consistent with diffusion limitations of photosynthesis in response to the applied treatments, and support the linear relationship between $A$ and $C_{i}$ ($r^2 = 0.89$, Supplementary Fig. 1D).

3.3. Limitations to photosynthesis are impacted by Si/Fe

We next evaluated the functional components of the global limitations of photosynthesis. In Fe plants, biochemical (BL) limitations accounted for most of the total photosynthetic constraints (67%) against approximately 16% for stomatal limitations (SL) and 17% for mesophyll limitations (ML), independently of cultivar (Fig. 5). In + Fe plants, SL were reduced regardless of cultivar and Fe stress duration with no effect of Si (Table 1, Fig. 5A and B). By contrast, Fe toxicity resulted in increased ML which were partially reversed by the addition of Si at 8 DAA in the Fe tolerant cultivar and at both 8 and 15 DAA in the sensitive cultivar (significant Si x Fe and Fe x Cv interactions, Table 1; Fig. 5C and D). The BL, in turn, were reduced in response to Fe toxicity (Fig. 5E and F), especially in –Si plants, independently of cultivar and stress duration (significant Si x Fe and Fe x Cv interactions). In summary, as can be deduced from Fig. 5, differential limitations (SL + ML) were responsible for approximately 60% of the total limitation of photosynthesis in + Fe plants, in contrast to their –Fe counterparts in which differential limitations did not exceed 40% of total photosynthetic limitations independently of Si application.

3.4. Carbon metabolism was minimally altered by the applied treatments

The activities of enzymes associated with carbon metabolism, including Rubisco, responded little, if at all, to the treatments (Supplementary Fig. 3). Of the 16 enzymes analyzed, only three displayed a reduction in their catalytic activity in + Fe plants. However, these changes were relatively small and thus we suggest that Fe (and Si) had no major impact on central metabolism under the present experimental conditions. Overall, we were also unable to detect a mitigating effect of Si on the enzyme activities under Fe excess (Supplementary Fig. 3).

The NADH/NAD⁺ and NADPH/NADP⁺ ratios were decreased by 24% and 30% in –Si + Fe plants from the Fe-sensitive cultivar, and by 23% and 49% in the Fe-tolerant cultivar, respectively, when compared with their + Si + Fe counterparts (with no significant Si x Fe interaction) (Table 1, Fig. 6). In turn, glucose and fructose pools increased significantly in + Fe plants of both cultivars, but more markedly in –Si plants (Table 1, Fig. 7A and B). On the other hand, the concentrations of sucrose, starch and proteins were unaffected by the treatments (Table 1, Fig. 7C and D,G). Malate and fumarate concentrations increased under Fe toxicity, particularly in the Fe-sensitive cultivar; Si weakened these increases regardless of cultivar (significant Si x Fe interaction) (Table 1, Fig. 7E and F).

4. Discussion

Irrespective of cultivar, the levels of Fe in both leaves and roots of + Fe plants were above those that are considered critical for Fe toxicity (≥ 300–500 mg kg⁻¹ dry mass) (Dobermann and Fairhurst, 2000). The Fe stress impaired the photosynthetic performance, particularly in the Fe-sensitive cultivar which displayed the highest leaf Fe levels. Notably, Si fertilization was associated with decreased Fe content and, as such, contributed to partially reverse the effects of Fe toxicity, in good agreement with our working hypothesis. It is also noted that Si per se did not affect A in –Fe plants, as also observed in other studies conducted with rice plants in their vegetative growth stage (Dettmann et al., 2012; Sanglard et al., 2014).

Decreases in A in rice cultivars under Fe excess have been largely attributed to decreases in $g_s$ (Pereira et al., 2013; Müller et al., 2015; Pinto et al., 2016). Here, we provided novel information that not only $g_s$ but also $g_m$ limits CO₂ flux towards the chloroplast stroma in response to Fe excess. Our results suggest an intrinsic correlation between $g_s$ and $g_m$ as proposed by Flexas et al. (2008) and Flexas et al. (2012). Similar results have been obtained in rice in response to arsenic (Sanglard et al., 2014) and in sugar beet in response to zinc (Sagaradoy et al., 2010). Given that the photochemical and biochemical abilities for CO₂ fixation were preserved under Fe excess, it is suggested that the reductions in A were mostly associated with diffusive limitations of photosynthesis.

The reduction of leaf conductance ($g_s + g_m$) in response to Fe excess (Fig. 2) was partially reversed by fertilization with Si. The reversal of these impacts depended on the time and exposure to Fe excess, being more evident at 8 than at 15 DAA, due possibly to the fact that Fe levels, more than the Si levels, increased over time irrespective of cultivar, as can be deduced from Fig. 1. Taken together, these results suggest that the mitigating effects of Si on Fe excess could be fundamentally associated with the reduction of Fe levels in + Si plants, in good agreement with the sharp relationships between $A$, $g_s$ and $g_m$ with leaf Fe concentrations. In any case we cannot dismiss some direct effect of Si on $g_s$ and/or $g_m$ (significant Si effects, Table 1). Although the physiological mechanisms associated with these effects have not been established,
some evidence suggests that Si may directly affect photosynthesis via modulating $g_s$ (Lavinsky et al., 2016) or $g_m$ (Detmann et al., 2012) in rice plants in the reproductive phase under unstressed conditions.

We obtained compelling evidence suggesting that the effects of Fe stress on $A$ were, to a great extent, unrelated to photochemical or biochemical constraints of photosynthesis. First, our analysis of Chl $a$ fluorescence demonstrated minimal variations, if any, in both the maximum PSII photochemical efficiency ($F_v/F_m$) and the photochemical quenching coefficient ($q_P$), suggesting stability of photochemical reactions in $+$Fe plants; these results contrast with those from other studies (e.g., Kampfenkel et al., 1995; Suh et al., 2002; Onyango et al., 2019) which have reported a higher susceptibility of PSII reaction centres to photoinhibition under Fe stress. Second, $J_{\text{max}}$ was kept at high values, even when $A$ was severely impaired, suggesting that the synthesis of both ATP and NADPH required to drive $CO_2$ fixation was largely uncompromised in $+$Fe plants. Third, the lack of treatment effect on $V_{\text{cmax}}$ implies that the RuBisCO capacity to fix $CO_2$ was preserved, in good agreement with unresponsiveness of RuBisCO activity to Fe excess. In any case, because both $F_v/F_m$ and $F_v'/F_m'$ (a proxy for thermal dissipation; Logan et al., 2007) were relatively invariant to the treatments
we applied, we contend that modification/inactivation of the PSII reaction centers and photoprotection by xanthophylls engaged in sustained thermal energy dissipation may have been unlikely (Logan et al., 2007) under our experimental conditions. Given that the J/e ratio tended to increase under Fe stress (particularly in -Si plants), we suggest that photoprotection should have played a key role for dissipating excess energy. Other energy dissipation mechanisms, such as the water-water cycle (Asada, 2000) or cyclic flow of electrons within PSII (Heber et al., 2001), could also have played a role in dissipating the excess reducing power under Fe stress, which seems to be consistent with the observed reductions in NADH/NAD+ and NADPH/NADP+ ratios.

Accumulation of metabolites such as hexoses (glucose and fructose) is commonly observed under conditions of abiotic stress, although it varies with the time and extent of stress imposed (Radomiljac et al., 2013; Sanglard et al., 2016), as observed here under Fe toxicity. In Arabidopsis thaliana some studies have reported that sugar levels increase remarkably under abiotic stress, which have usually been associated with osmeregulation, membrane stabilization and protection, and altered carbon balance conditions (Rizhsky et al., 2004; Kaplan et al., 2007; Hummel et al., 2010). In addition, accumulation of organic acids (malate and fumarate) in both cultivars under Fe excess may be related to the non-cyclical functioning of the mitochondrial tricarboxylic acid cycle, or alterations in cell redox balance (Hummel et al., 2010; Igamberdiev and Eprinets, 2016). Nonetheless, the levels of hexoses were much lower than those of sucrose + starch (which remained unaltered regardless of treatments), thus concurring for an invariant concentration of non-structural carbohydrates. This fact may be associated with the lower demand for assimilates in the +Fe plants, which accumulated less biomass than their -Fe counterparts (data not shown).

In conclusion, we here demonstrated that Si fertilization was able to partially reverse the toxic effects of Fe on the photosynthesis by fundamentally reducing the accumulation of Fe in the leaf tissues. This seems to largely explain the lower Fe content in +Si plants associated with higher g i and g s values, thus contributing to mitigate the negative impacts of Fe excess on A in +Si + Fe plants. Notably, Fe toxicity and/or Si fertilization affected minimally the photochemistry and biochemistry of photosynthesis; these facts, coupled with little, if at all, alterations of enzyme activities associated with the central carbon metabolism, suggest a remarkable metabolic homeostasis under Fe toxicity. It remains to be investigated the underlying mechanisms by which Fe at excess levels impairs leaf conductance and thus the magnitude of photosynthetic rates.

Author contributions

M. Santos performed the experiments and data analyses, and prepared the graphs and tables. L. Sanglard, M. Barbosa, D. Melo and W. Gonçalves performed the experiments. S. Martins helped in data analyses and helped to write the paper. F. DaMatta designed the experiment and wrote the paper.

Conflicts of interest

The authors hereby declare no conflicts of interests.

Acknowledgements

This research was supported by the Foundation for Research Assistance of Minas Gerais State, Brazil (FAPEMIG), Grant (CRA-APQ-02132-15) and by the National Council for Scientific and Technological Development, Brazil (CNPq), Grant (308652/2014-2) granted to FMD. This study was also financed in part by the Brazilian Federal Agency for Support and Evaluation of Graduate, Brazil (CAPES), Finance Code (001). Thanks are due also to the Instituto Riograndense do Arroz (IRGA) and Empresa de Pesquisa Agropecuária e Extensão Rural de Santa Catarina (EPAGRI) which kindly supplied the rice seeds.

Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.plaphy.2019.09.011.

References


