



## Short Communication

Interface engineering gifts CsPbI<sub>2.25</sub>Br<sub>0.75</sub> solar cells high performance<sup>☆</sup>

Zhimin Fang<sup>a,b</sup>, Xianyi Meng<sup>a</sup>, Chuantian Zuo<sup>c</sup>, Dan Li<sup>a</sup>, Zuo Xiao<sup>a</sup>, Chenyi Yi<sup>d</sup>, Mingkui Wang<sup>e,\*</sup>, Zhiwen Jin<sup>f,\*</sup>, Shangfeng Yang<sup>b,\*</sup>, Liming Ding<sup>a,\*</sup>

<sup>a</sup> Center for Excellence in Nanoscience (CAS), Key Laboratory of Nanosystem and Hierarchical Fabrication (CAS), National Center for Nanoscience and Technology, Beijing 100190, China

<sup>b</sup> Hefei National Laboratory for Physical Sciences at Microscale, Key Laboratory of Materials for Energy Conversion (CAS), Department of Materials Science and Engineering, University of Science and Technology of China, Hefei 230026, China

<sup>c</sup> CSIRO Manufacturing, Bag 10, Clayton South, Victoria 3169, Australia

<sup>d</sup> Department of Electrical Engineering, Tsinghua University, Beijing 100084, China

<sup>e</sup> Wuhan National Laboratory for Optoelectronics, Huazhong University of Science and Technology, Wuhan 430074, China

<sup>f</sup> School of Physical Science and Technology, Lanzhou University, Lanzhou 730000, China

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Organic-inorganic halide perovskite (ABX<sub>3</sub>) solar cells (PSCs) have made great progress in recent years [1]. The power conversion efficiency (PCE) has increased up to 25.2% (NREL Best Research-Cell Efficiency Chart, <https://www.nrel.gov/pv/cell-efficiency.html>, Accessed August 2019). However, they suffer from poor thermal stability due to the volatile A-site organic cations. All-inorganic CsPbI<sub>3-x</sub>Br<sub>x</sub> perovskite materials can tolerate temperature exceeding 400 °C, while organic-inorganic halide PSCs rapidly deteriorated at greater than 200 °C [2]. The excellent thermal stability of inorganic perovskites makes them promising materials for PSCs [3]. Their relatively low PCE is due to two reasons. One is the wide bandgap, which leads to insufficient light absorption and low *J*<sub>sc</sub>. Developing tandem solar cells is an effective approach to absorb sunlight as much as possible [4]. Another is the large energy loss (*E*<sub>loss</sub>), leaving much room for increasing *V*<sub>oc</sub>. Interface engineering can reduce the *E*<sub>loss</sub> and increase the PCE. Wang et al. [5] reported a 17.06% PCE by using PTABr-treated CsPbI<sub>3</sub> as the absorber. The post-treatment with PTABr can realize gradient Br-doping and surface passivation, leading to enhanced *V*<sub>oc</sub> and *FF*. They also modified CsPbI<sub>3</sub> surface with choline iodine (CHI), which increased the charge-carrier lifetime and improved the energy level alignment between CsPbI<sub>3</sub> and charge-

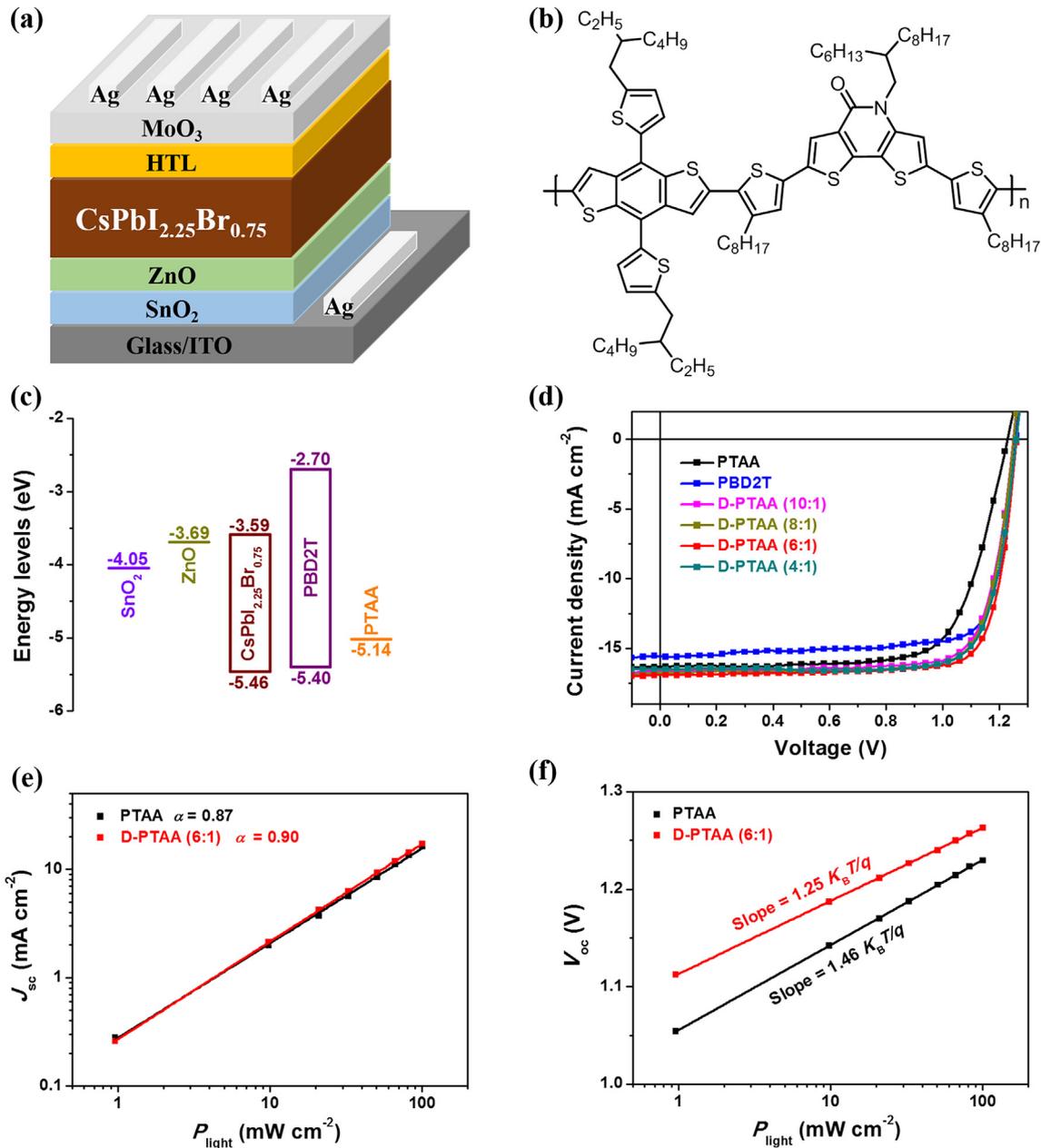
transport layers. CHI-CsPbI<sub>3</sub> solar cells gave a PCE of 18.4% [6]. Besides, choosing electron-transport layers (ETLs) and hole-transport layers (HTLs) with suitable energy levels is also effective for improving *V*<sub>oc</sub> [7]. In our previous work, we applied DPPA-modified ZnO as ETL for CsPbI<sub>2.25</sub>Br<sub>0.75</sub> solar cells, obtaining a 15.98% PCE with an enhanced *V*<sub>oc</sub> [8]. Yan et al. [9] used SnO<sub>2</sub>/ZnO bilayer as ETL to get high *V*<sub>oc</sub>. They also used PN4N-modified SnO<sub>2</sub> as ETL and PDCBT as HTL in CsPbI<sub>2</sub>Br solar cells, obtaining a 16.2% PCE with a 1.30 V *V*<sub>oc</sub> [10]. In this work, we made inorganic PSCs with a structure of ITO/SnO<sub>2</sub>/ZnO/CsPbI<sub>2.25</sub>Br<sub>0.75</sub>/HTL/MoO<sub>3</sub>/Ag (Fig. 1a). PTAA doped with polymer donor PBD2T [11] was used as HTL (D-PTAA). The deep highest occupied molecular orbital (HOMO) level of PBD2T matches HOMO of CsPbI<sub>2.25</sub>Br<sub>0.75</sub> well. Meanwhile, PBD2T can passivate the trap states on perovskite surface and suppress interfacial charge recombination. The solar cells with D-PTAA delivered a 17.37% PCE, which is the highest efficiency for Br-doped inorganic PSCs.

Fig. 1b presents the chemical structure of PBD2T. D-PTAA films were made by adjusting PTAA/PBD2T ratio (*w/w*). The UV-vis absorption spectra of CsPbI<sub>2.25</sub>Br<sub>0.75</sub> and CsPbI<sub>2.25</sub>Br<sub>0.75</sub>/D-PTAA films are shown in Fig. S1 (online). Fig. 1c shows the energy level diagram. The conduction band of SnO<sub>2</sub> and ZnO are -4.05 eV [12] and -3.69 eV [8], respectively. The cascade energy level alignment between CsPbI<sub>2.25</sub>Br<sub>0.75</sub>, ZnO and SnO<sub>2</sub> can facilitate electron extraction and reduce the *E*<sub>loss</sub> [9]. The HOMO levels for PBD2T and PTAA are -5.40 eV [11] and -5.14 eV [13], respectively. The deeper HOMO of PBD2T provides a better energy level matching with CsPbI<sub>2.25</sub>Br<sub>0.75</sub>. When adding PBD2T into PTAA, a cascade

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\* Corresponding authors.

E-mail addresses: [mingkui.wang@mail.hust.edu.cn](mailto:mingkui.wang@mail.hust.edu.cn) (M. Wang), [jinzw@lzu.edu.cn](mailto:jinzw@lzu.edu.cn) (Z. Jin), [sfyang@ustc.edu.cn](mailto:sfyang@ustc.edu.cn) (S. Yang), [ding@nanocr.cn](mailto:ding@nanocr.cn) (L. Ding).



**Fig. 1.** Interface engineering for CsPbI<sub>2.25</sub>Br<sub>0.75</sub> solar cells. (a) Device structure; (b) the chemical structure of PBD2T; (c) the energy level diagram; (d) *J*-*V* curves for CsPbI<sub>2.25</sub>Br<sub>0.75</sub> solar cells with different HTLs; (e) *J*<sub>sc</sub>-*P*<sub>light</sub> plots; (f) *V*<sub>oc</sub>-*P*<sub>light</sub> plots.

energy level alignment between CsPbI<sub>2.25</sub>Br<sub>0.75</sub>, PBD2T and PTAA was built, thus improving the hole extraction and reducing the *E*<sub>loss</sub>. In addition, sulfur (S) atoms in PBD2T could form S-Cs and S-Pb bonds and effectively passivate the trap states on perovskite surface, thus reducing the interfacial charge recombination [10,14,15].

Solar cells with a structure of ITO/SnO<sub>2</sub>/ZnO/CsPbI<sub>2.25</sub>Br<sub>0.75</sub>/HTL/MoO<sub>3</sub>/Ag were made and PBD2T content was optimized. *J*-*V* curves for the solar cells with different HTLs are shown in Fig. 1d, and the corresponding performance data are listed in Table S1 (online). The optimal PTAA:PBD2T ratio is 6:1 (wt). Compared with the control device, the solar cells with D-PTAA HTLs offered enhanced PCEs. The increase of *V*<sub>oc</sub> results from the energy level matching between CsPbI<sub>2.25</sub>Br<sub>0.75</sub> and PBD2T. *J*<sub>sc</sub> increases first and then decreases as PBD2T content increases. The energy level matching between CsPbI<sub>2.25</sub>Br<sub>0.75</sub> and D-PTAA HTLs could improve the hole extraction. When PTAA:PBD2T ratio decreases below 6:1 (PBD2T content

increases), *J*<sub>sc</sub> decreases because of the lower hole mobility of PBD2T ( $1.3 \times 10^{-4} \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ ) [9] than PTAA ( $4 \times 10^{-3} \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ ) [11]. Solar cells with PBD2T HTL gave the lowest *J*<sub>sc</sub>. The improvement of *FF* was due to efficient hole extraction and reduced interfacial charge recombination. The thickness (Table S2 online) and annealing temperature (Fig. S2, Table S3, online) for HTLs were optimized. Solar cells with PTAA HTL gave a *V*<sub>oc</sub> of 1.23 V, a *J*<sub>sc</sub> of 16.34 mA cm<sup>-2</sup>, an *FF* of 71.63% and a PCE of 14.39%. The best cell with D-PTAA (6:1) HTL gave enhanced *V*<sub>oc</sub> (1.26 V), *J*<sub>sc</sub> (17.16 mA cm<sup>-2</sup>), and *FF* (80.14%), and a higher PCE of 17.37% with a lower *E*<sub>loss</sub> of 0.61 eV. The integrated photocurrents from EQE spectra for the cells with PTAA and D-PTAA (6:1) HTLs are 15.64 and 16.28 mA cm<sup>-2</sup>, respectively (Fig. S3a online). *J*<sub>sc</sub> stability for the best cell was studied (Fig. S3b online). The hysteresis for *J*-*V* curves was checked, and the D-PTAA cells presented less hysteresis than the control device (Fig. S4, Table S4, online).

The dependence of  $J_{sc}$  and  $V_{oc}$  on the incident light intensity ( $P_{light}$ ) was studied (Fig. 1e and f).  $J_{sc}$  exhibits a power law dependence on  $P_{light}$ :  $J_{sc} \propto P_{light}^\alpha$  ( $\alpha \leq 1$ ). The  $\alpha$  values for PTAA and D-PTAA (6:1) solar cells are 0.87 and 0.90, respectively. The larger  $\alpha$  suggests that the energy level matching between  $\text{CsPbI}_{2.25}\text{Br}_{0.75}$  and D-PTAA is better and the holes can be extracted efficiently. The slope for  $V_{oc}$  vs.  $\text{Log}P_{light}$  plots for D-PTAA (6:1) cell ( $1.25 K_B T/q$ ) is smaller than that for PTAA cell ( $1.46 K_B T/q$ ). Generally, a slope larger than 1 indicates the presence of trap-assisted charge recombination. The smaller slope suggested that the trap-assisted charge recombination is effectively suppressed by PBD2T [7,8].

In summary, a polymer donor PBD2T was added into PTAA to make HTL for  $\text{CsPbI}_{2.25}\text{Br}_{0.75}$  solar cells. The cascade energy level alignment between  $\text{CsPbI}_{2.25}\text{Br}_{0.75}$ , PBD2T and PTAA can improve the hole extraction and reduce the  $E_{loss}$ . Meanwhile, the formation of S-Cs and S-Pb bonds can passivate perovskite surface effectively, suppressing the interfacial charge recombination.  $\text{CsPbI}_{2.25}\text{Br}_{0.75}$  solar cells with D-PTAA HTL deliver a 17.37% PCE, which is the highest efficiency for Br-doped inorganic PSCs. This work provides an effective approach for developing high-performance inorganic PSCs.

### Conflict of interest

The authors declare that they have no conflict of interest.

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### Author contributions

Zhimin Fang and Xianyi Meng performed the experiments. Dan Li synthesized PBD2T. Chuantian Zuo, Zuo Xiao, Chenyi Yi, Mingkui Wang, Zhiwen Jin and Shangfeng Yang participated in the discussion on experimental results. Liming Ding directed this project.

### Appendix A. Supplementary materials

Supplementary materials to this article can be found online at <https://doi.org/10.1016/j.scib.2019.09.023>.

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Zhimin Fang got his B.S. degree from Sichuan University in 2015. Now he is a Ph.D. student at University of Science and Technology of China under the supervision of Prof. Shangfeng Yang. Since September 2017, he has been working in Liming Ding Group at National Center for Nanoscience and Technology as a visiting student. His research work focuses on perovskite solar cells.



Mingkui Wang received his Ph.D. degree in physical chemistry in 2004 from Chinese Academy of Sciences. In 2005 he worked as an Alexander von Humboldt Fellow at Technische Universität München. He then joined Michael Graetzel Group at the Swiss Federal Institute of Technology Lausanne as a postdoctor. Since 2010 he has been a professor at Huazhong University of Science and Technology. Now his research is on next-generation solar cells.



Zhiwen Jin received his B.S. degree from Lanzhou University in 2011 and Ph.D. degree from Institute of Chemistry, Chinese Academy of Sciences in 2016. He joined Lanzhou University in 2018, and now he is a professor in School of Physical Science and Technology. His research interests include inorganic semiconductors, photoelectric devices and device physics, particularly inorganic perovskite solar cells.



Shangfeng Yang got his Ph.D. degree from Hong Kong University of Science and Technology in 2003. He then joined Leibniz Institute for Solid State and Materials Research, Dresden as an Alexander von Humboldt Fellow. In December 2007, he joined University of Science and Technology of China as a full professor. His research includes the synthesis of fullerene-based nanocarbons toward applications in energy devices.



Liming Ding got his Ph.D. degree from University of Science and Technology of China. He started his research on OSCs and PLEDs in Olle Inganäs Lab in 1998. Later on, he worked with Frank Karasz and Tom Russell at PSE, UMASS Amherst. He joined Konarka as a Senior Scientist in 2008. In 2010, he joined National Center for Nanoscience and Technology as a Full Professor. His research includes perovskite solar cells, organic solar cells and photodetectors.