



Aptamer-DNA concatamer-quantum dots based electrochemical biosensing strategy for green and ultrasensitive detection of tumor cells via mercury-free anodic stripping voltammetry

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ABSTRACT

An electrochemical biosensing strategy was developed for green and ultrasensitive detection of tumor cells by combining aptamer-DNA concatamer-CdTe quantum dots (QDs) signal amplification probe with mercury-free anodic stripping voltammetry (ASV). First, aptamer-DNA concatamer- CdTe QDs probes were designed by DNA hybridization and covalent assembling, which contained specific recognition of aptamer and signal amplification incorporating the DNA concatamer with QDs. Meanwhile, the capture electrode, glassy carbon electrode (GCE)/ Graphene oxide (GO)/Polyaniline (PANI) / Glutaraldehyde (GA) / concanavalin A (Con A) was fabricated by a layer-by-layer assembling technique. K562 cells, as model cancer cells were detected to demonstrate the feasibility of this sensing strategy. Then, novel composite, graphene (GR)- Poly diallyldimethylammonium chloride (PDDA)/L-Cysteine (L- Cys), was explored in ASV which replaced mercury electrodes using for determination of tumor cells. The proposed electrochemical biosensor showed high sensitivity with the detection limit of 60 cells mL⁻¹. More importantly, this novel design of signal amplification probes and the exploration of new composites in mercury-free ASV analysis would provide a promising method for ultrasensitive biosensor preparation and green electrochemical detection of tumor cells.

1. Introduction

Sensitive and specific determination of tumor cells is essential in clinical cancer screening and disease diagnosis (Loren et al., 2013; Mullighan et al., 2008, 2009). Numerous typical detection methods have been developed including flow cytometry, polymerase chain reaction, and immunohistochemistry (Li et al., 2002; Liu et al., 2013; Pitsillides et al., 2011; Zhou et al., 2016a, 2016b; Phillips et al., 2009; Schamhart et al., 2003; Singh et al., 2004). Electrochemical biosensing detection, due to moderate cost, instrumental simplicity, and high sensitivity, has attracted worldwide attention (Grieshaber et al., 2008; Saha et al., 2012; Shao et al., 2010). To achieve higher sensitivity and selectivity, various new materials and novel designs have been explored in electrochemical biosensors construction, such as nanomaterials (Liu et al., 2016; Tu et al., 2015; Yang et al., 2018), conducting polymers (Rowe et al., 2011), aptamers (H. Liu et al., 2013; W. Liu et al., 2013) and catalyst enzymes (Zhu et al., 2011).

Quantum dots (QDs), as a new nanomaterial, due to their

electroactivity (Gill et al., 2008; Liu et al., 2004), versatility in surface modification and unique optical properties (Wang et al., 2017; Cui et al., 2007; Qian et al., 2006; Zhang et al., 2006) have attracted more and more interests. Recently, QDs incorporating with a wide variety of DNA concatamer, as new amplifying tags, have been developed in electrochemical biosensor (Chen et al., 2011; Liu et al., 2013). Liu et al. (H. Liu et al., 2013; W. Liu et al., 2013) developed a versatile super-sandwich cytosensor for detecting cancer cells. They combined the signal amplification of the DNA concatamer, the specific binding ability of aptamer, the optical and electrical properties of QDs, with the sensitive detection of fluorescence and electrochemistry. Multiwall carbon nanotubes (MWCNTs) @ polydopamine (PDA)@Au QDs composites were assembled on the surface of the electrode as the immobilization matrices of concanavalin A (Con A) to capture tumor cells. Then, aptamer-DNA concatamer-CdTe QDs as signal probe bond specifically on the cells. These design strategies in biosensor exhibited good sensitivity, wide linear range (1.0×10^2 to 1.0×10^6 cells mL⁻¹), good reproducibility and acceptable precision (The detection limit is estimated to be

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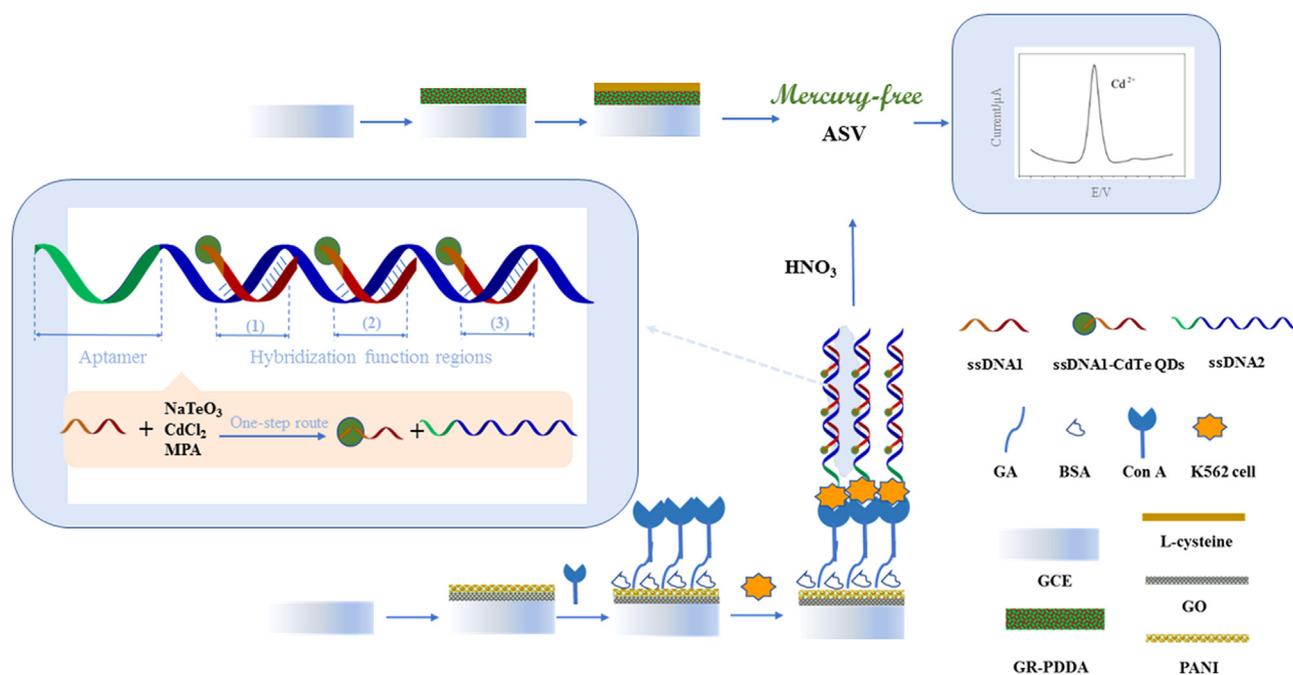
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Scheme 1. Procedures for the fabrication of supersandwich cytosensor using controlled aptamer-DNA concatamer-QDs and the detection of Cd²⁺ via a mercury-free ASV route.

were dispersed in PBS and stored in dark at 4 °C.

Synthesis of ssDNA3- CdTe QDs. The synthesis of ssDNA3 - CdTe QDs was also according to the procedures of ssDNA1- CdTe QDs. Two different domains were designed in ssDNA3, that is, phosphorothioates backbone served as a ligand structure for the synthesis of CdTe QDs and phosphates backbone in which the corresponding DNA sequence was in accord with aptamer DNA sequence.

Preparation of aptamer-DNA concatamer- CdTe QDs. Typically, 0.2 mmol cadmium chloride, 0.2 mmol MPA, 25 mL distilled water were loaded into a three-necked flask in sequence under vigorous stirring. 0.025 mmol sodium tellurite and 0.5 mmol sodium borohydride were mixed in the above solution under the stirring (pH = 9) at room temperature. Then, 10 μ L ssDNA1 (10 μ M) solution was added in 990 μ L above mixture solution and heated to 90 °C for 2.5 h. 10 μ L ssDNA2 (1 μ M) and 1 mL ssDNA1- CdTe QDs solution were mixed and heated to 95 °C for 10 min to make the folded ss- DNA chain spread out and then cooled down in ice-water bath. Then ssDNA1- CdTe QDs was assembled onto the ssDNA2 substrate strand and formed DNA superstructure. The mixture was then subjected to ultrafiltration using a 100 kDa MW filter. The aptamer-DNA concatamer-CdTe QDs were obtained and stored in dark at 4 °C.

Cell culture. Human leukemia K562 cells were cultured in a flask with a RPMI 1640 medium (Gibco, Grand Island, NY) supplemented with 10% fetal calf serum, ciprofloxacin (10 μ g mL⁻¹), in an incubator (5% CO₂, 37 °C). Cell counting was used by trypan dye exclusion method by a Neubauer's chamber (10 μ L of cell suspension mixed with 10 μ L of dye). The remaining cells in suspension were then centrifuged at 1100 rpm for 10 min and then suspended in sterile PBS buffer (pH = 7.4) (Gulati et al., 2018).

Confocal Microscopy Imaging. K562 cells (1 \times 10⁵ cells mL⁻¹) were seeded onto a well plate. After 24 h, cells were incubated with aptamer-DNA concatamer- CdTe QDs probe for 1.5 h. To remove the unbound probes, the cells were washed three times with PBS. Then, the well was placed above a 20 \times objective on the confocal microscope.

2.4. Fabrication of electrochemical biosensor

The fabrication and detection procedure of electrochemical

biosensor were indicated in Scheme 1. The capture electrode GCE/GO/PANI/GA/Con A was prepared according to our recent work (Wang et al., 2018a, 2018b). More information about the protocol were described in the Supporting information. The capture electrode GCE/GO/PANI/GA/Con A was immersed in K562 cells solution with different concentration at 37 °C for 30 min. After carefully washing with the buffer to remove non-captured cells, 5 μ L aptamer- DNA concatamer- CdTe QDs probe solution was dropped onto the capture electrode and incubated at 37 °C for 1.5 h. The supersandwich biosensor were obtained (denoted as GCE/GO/PANI/GA/Con A/K562 cells/ aptamer-DNA concatamer-CdTe QDs). After rinsing with PBS, the biosensor was stored at dark stored in dark at 4 °C.

2.5. Construction of detecting electrode GCE/ GR- PDDA/ L-Cys and electrochemical analysis

Construction of detecting electrode GCE/ GR- PDDA/ L-Cys. Prior to surface modification, the GCE (d = 3 mm) was polished (0.05 mm alumina) and followed by successive sonication in acetone, HNO₃ (1:1 v/v), and distilled water, respectively. 5 μ L mixture solution containing GR (100 μ L, 1 mg mL⁻¹) and PDDA (6 μ L, 20%) with different molar ratio was dropped on bare GCE and dried at room temperature for 2 h. The modified electrode, GCE/ GR-PDDA, was performed in 10 mmol L⁻¹ PBS (pH = 7.2) containing 1 mmol L⁻¹ L-Cys by CV scanning between -0.6 and 2.5 V for about 20 segments (scan rate, 100 mV s⁻¹).

Electrochemical analysis. To release Cd²⁺, the obtained sensing system, GCE/GO/PANI/GA/Con A/K562 cells/ aptamer-DNA concatamer-CdTe QDs, was immersed in HNO₃ solution (200 μ L, 0.1 M) for 2 h to dissolve the residual QDs. Then, 4.8 mL Hac-NaAc buffer (0.2 M, pH = 5.2) was added as the reaction supporting solution of ASV. The anodic stripping voltammetry detection involved electrodeposition at -1.2 V for 30 min, and stripping from -1.1 to -0.4 V using a square-wave voltammetry, with 4 mV potential steps, 25 Hz frequency and 25 mV amplitude. The oxidation peak of Cd²⁺ was achieved at -0.86 V in ASV and the intensity of oxidation peak currents of Cd²⁺ was proportional to the amount of K562 cells.

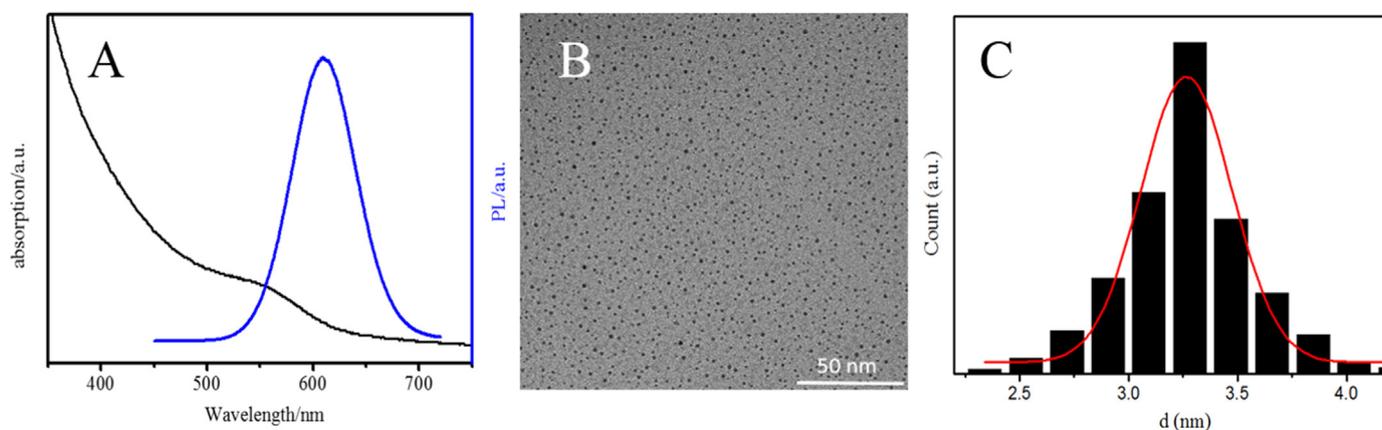


Fig. 1. UV-Vis and PL spectrum of DNA-CdTe QDs (A), TEM images (B) and particle size distribution histogram (C).

3. Results and discussion

3.1. Characterization of ssDNA1-CdTe QDs

ssDNA1-CdTe QDs were synthesized via a facile one-step route. The UV-vis absorption and PL spectra of ssDNA1-CdTe QDs were shown in Fig. 1A. A strong exciton absorption peak of ssDNA1-CdTe QDs appeared at 556 nm. The sharp PL emission peak was at about 609 nm and the full width at half maximum (FWHM) was about 26 nm, which represented CdTe QDs with a narrow size distribution. Furthermore, about 53 nm Nonresonant Stokes Shift (NRSS), which was the difference between positions of the band maxima of the absorption and emission spectra, indicated the CdTe QDs with the well-defined structure and minimal surface defects. According to the Peng's empirical formula (Yu et al., 2004): $D = (9.8127 \times 10^{-7})\lambda^3 - (1.7147 \times 10^{-3})\lambda^2 + (1.0064)\lambda - (194.84)$, the size of CdTe QDs was about 3.3 nm by the calculation, D (nm) is the diameter of QDs and λ (nm) is the excitonic absorption peak position. The TEM picture and the particle size distribution histogram support this conclusion as shown in Fig. 1B and C.

3.2. Characterization of aptamer-DNA concatamer-CdTe QDs probe

To investigate the effect of both the structure and number of QDs on signal intensity of aptamer-DNA concatamer-CdTe QDs probe, ssDNA3-CdTe QDs and aptamer-DNA concatamer-CdTe QDs probes were synthesized. A multifunctional ssDNA substrate (ssDNA2) was designed, which was comprised of the aptamer region to recognize K562 cell and three target regions to hybridize ssDNA1-CdTe QDs. After three ssDNA1-CdTe QDs had hybridized with DNA substrate (ssDNA2), the controlled aptamer-DNA concatamer-CdTe QDs probes were formed. The gel electrophoresis was also used to verify the formation of aptamer-DNA concatamer-CdTe QDs. In Fig. S1., ssDNA1, and ssDNA2 produced the bands at about 75 bp and 130 bp, respectively. Aptamer-DNA concatamer-CdTe QDs produced one band at above 300 bp. The results showed that aptamer-DNA concatamer-CdTe QDs had long DNA strands, and the hybridization reaction was implemented by ssDNA1, and ssDNA2. In order to ensure that the connection between tumor cells and probes, the two sample cells were incubated with the aptamer-DNA concatamer-CdTe QDs probe and ssDNA3-CdTe QDs probe at 37 °C for 2 h, respectively. The extracellular ssDNA3-CdTe QDs probes exhibited fluorescence image compared with the image under visible light irradiation, which suggested that the probes had bonded with K562 cells. The ssDNA3-CdTe QDs probe and aptamer-DNA concatamer-CdTe QDs probes experiments were carried out (Fig. 2A, Fig. 2B, Fig. 2C, Fig. 2D) and much stronger fluorescence intensity than ssDNA3-CdTe QDs probe at the equal reaction conditions which illustrated that aptamer-DNA concatamer-CdTe QDs probes

were more effective in tumor cell detection. Thus, from what had discussed as above, we could control number and precise structure in probes by the design of target region in DNA substrate strand, and then regulated the signal intensity in electrochemical detection. The morphology and size of the probes were characterized by TEM as shown in Fig. 2E and Fig. 2F. The ssDNA3-CdTe QDs was nearly monodisperse and spherical, whose size was about 3 nm. And the aptamer-DNA concatamer-CdTe QDs probe showed a dendrimer-like structure about 10 nm.

3.3. Characterization of capture electrode GCE/GO/PANI/GA/Con A

The electrochemical impedance spectroscopy (EIS) and cyclic voltammetry (CV), as the effective tools, were adopted to investigate the assembly process of biosensor as shown in Fig. S2. The semicircle diameter of EIS in high frequency region is correlation to electron-transfer resistance (R_{et}). A change in diameter reflects the blocking behavior in the modification processes of electrode which associated with a change of R_{et} . (Pan et al., 2010). The R_{et} of bare GCE showed a small value (Fig. S2A curve a) due to the free electron transfer process, and the peak current of CV was distinct (Fig. S2B curve a). while the GO modified electrode has a much higher R_{et} in the ESI (Fig. S2A curve b), and the peak current of CV decreased (Fig. S2B curve b). But when the PANI modified electrode had a lower resistance (Fig. S2A curve c) and a higher the peak current of CV (Fig. S2B curve c), which proved that PANI was an excellent conducting material. Subsequently, the immobilization of GA and Con A onto the GCE/GO/PANI electrode, the resistance increased (Fig. S2A curve d,e), and the peak current of CV decreased continually (Fig. S2B curve d,e) which suggested that GA and Con A were immobilized on the electrode and blocked electron transfer between the redox probe and electrode. After capture of the K562 cells, the resistance showed great increase due to the dielectric behavior of cells. As expected, the peak current of CV further decreased (Fig. S2B curves f). Additionally, after incubation with aptamer DNA concatamer-QDs probes, the electrode was further hindered (Fig. S2A curve g and Fig. S2B curve g), which also suggested the capture electrode, GCE/GO/PANI/GA/Con A could capture the tumor cells and signal probes with high efficiency.

3.4. Characterization of detecting electrode and optimization of experimental conditions

The detecting electrode was needed to establish for achieving a novel non-toxic determination strategy, in which a GR-PDDA/L-Cys composites were explored. Abundant sulfur, nitrogen and oxygen elements in L-Cys could produce extremely strong adsorption for heavy metal ions such as Cd^{2+} and Pb^{2+} in the solution which could take the place of the amalgam in ASV process. To investigate the modified

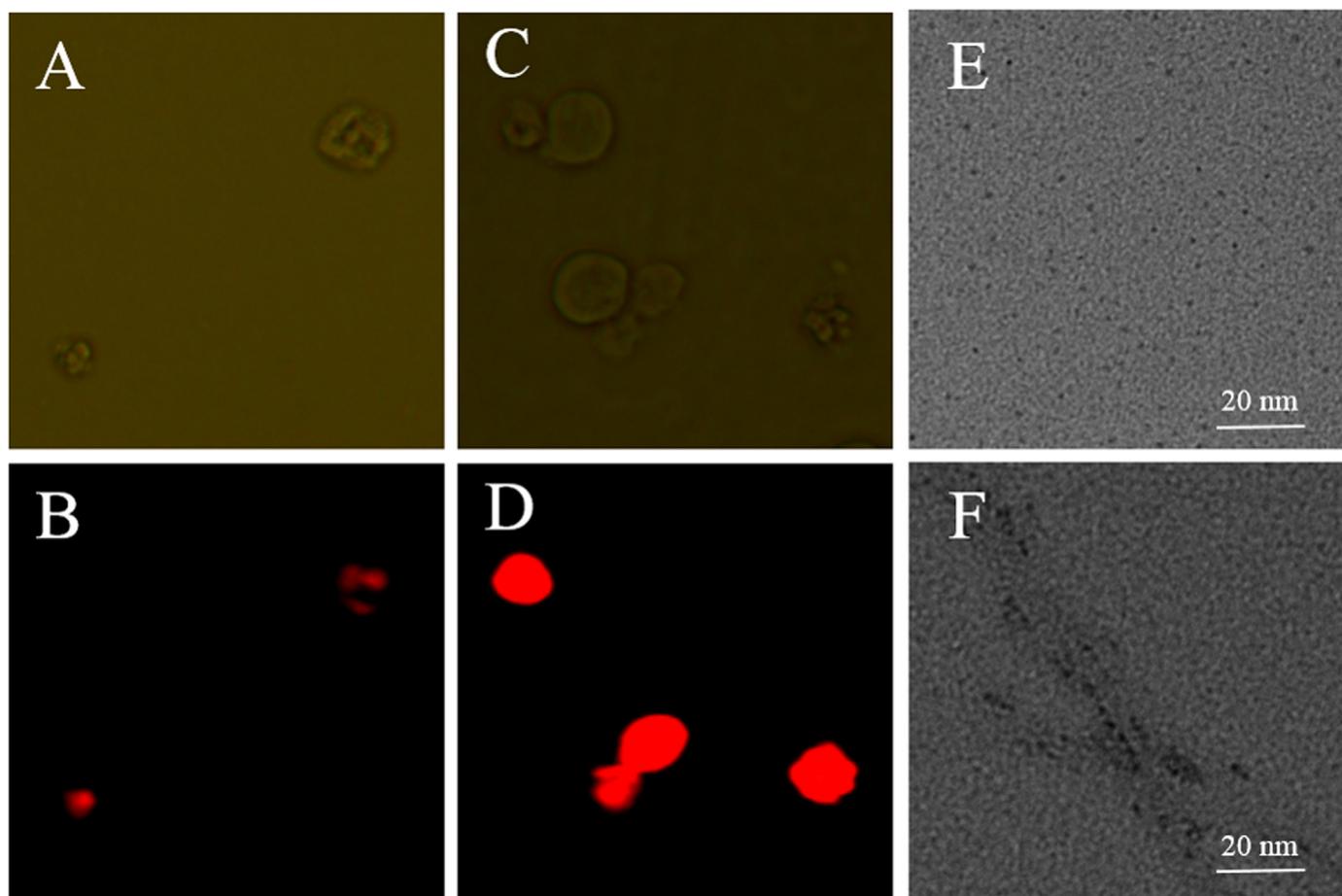


Fig. 2. Fluorescence microscopic images of single probe bonded K562 cells under visible light irradiation (A), and ultraviolet light irradiation (B). Fluorescence microscopic images of triploid probe bonded K562 cells under visible light irradiation (C), and ultraviolet light irradiation (D); TEM of aptamer-single DNA-QDs probes (ssDNA3-CdTe QDs) (E) and aptamer-DNA concatamer-CdTe QDs (F).

electrodes, EIS was used at the potential of 0.2 V under the participation of the redox reaction of $K_3Fe(CN)_6 / K_4Fe(CN)_6$ as shown in Fig. 3A. The semi-circle of EIS at high frequency region of bare GCE, GCE/ GR-PDDA and GCE/ GR- PDDA/ L- Cys increased which indicated the electronic transmission resistance of electrode interface grew gradually. This series of changes indicated that L- Cys and GR- PDDA were successfully modified on the bare GCE. Moreover, SEM images of the GCE/ GR- PDDA and GCE/ GR- PDDA/ L- Cys were shown in Fig. S3C. When GR- PDDA composites were assembled onto the GCE (Fig. S3C(A)), many particles appeared on the smooth surface of GCE, which illustrated that GR- PDDA had connected with GCE. After L-Cys being assembled, the SEM image of L-Cys coated GCE / GR- PDDA, became blurring and smooth, due to the form of the membrane resulted from the condensation reaction between L-Cys and GR- PDDA. The investigation of TEM also verified the assembly process of detecting electrode.

To optimize detecting electrode, GCE/ GR- PDDA/ L- Cys, electrochemical response, the effect of GR and PDDA with different ratio by R_{et} and oxidation peak current was discussed. Like the adhesive, PDDA with positively charge made the completely incompatible GR load on the GCE surface. Different proportions between GR and PDDA impacted electronic transmission capability of electrode interface enormously. R_{et} enhanced with the ratio value of GR: PDDA increasing (as Fig. 3B). Meanwhile, the ASV oxidation peak current increased with the ratio between GR and PDDA up from 1:1 to 1:3, and then down as shown in Fig. 3C and Fig. 3D. With the addition of PDDA, though a little PDDA could increase the specific surface area of the electrode by capturing much more signal recognition material, excessive PPDA also could

make the electron transfer efficiency weak and signal intensity low. Thus, 1:3 was chosen as the optimized composite ratio between GR and PDDA.

3.5. Detection of K562 tumor cells

To prepare electrochemical biosensor, Con A was introduced to GCE, which was modified by GO and PANI composite through the crosslinking of GA as Scheme 1. Then, K562 cells were immobilized on the capture electrode, GCE/GO/PANI/GA/Con A, after BSA blocking the non-specific binding sites. Subsequently, aptamer- DNA concatamer-CdTe QDs probe (or ssDNA3-CdTe QDs probe) was captured on the surface of K562 cells and the supersandwich electrochemical biosensor was obtained. After GCE/GO/PANI/GA/Con A/K562 cells/aptamer-DNA concatamer-QDs probes were dissolved in HNO_3 solution, Cd^{2+} could be released in the solution and detected by ASV. Due to much O, N, S of L- Cys resulted from GCE surface, the Cd^{2+} ions were captured specifically at about -0.86 V. Compared with the conventional ASV peak position at -0.72 V (Zhou et al., 2016a, 2016b), the peak position migration occurred toward the negative potential direction because some materials such as GR, PDDA and L- Cys were adopted on the surface of GCE. The ASV response signal of this biosensor with different concentrations of K562 cells were shown in Fig. 4A. The intensity of peak currents of Cd^{2+} showed a good linear relationship with the logarithm of cells concentration in range from 1.0×10^2 to 1.0×10^7 cells mL^{-1} (Fig. 4B). The linear regression equation was $I = -4.82451 + 4.27868 \lg C$, with a correlation coefficient of 0.9986 ($n = 5$). The limit of detection was 60 cells mL^{-1} (S/N = 3).

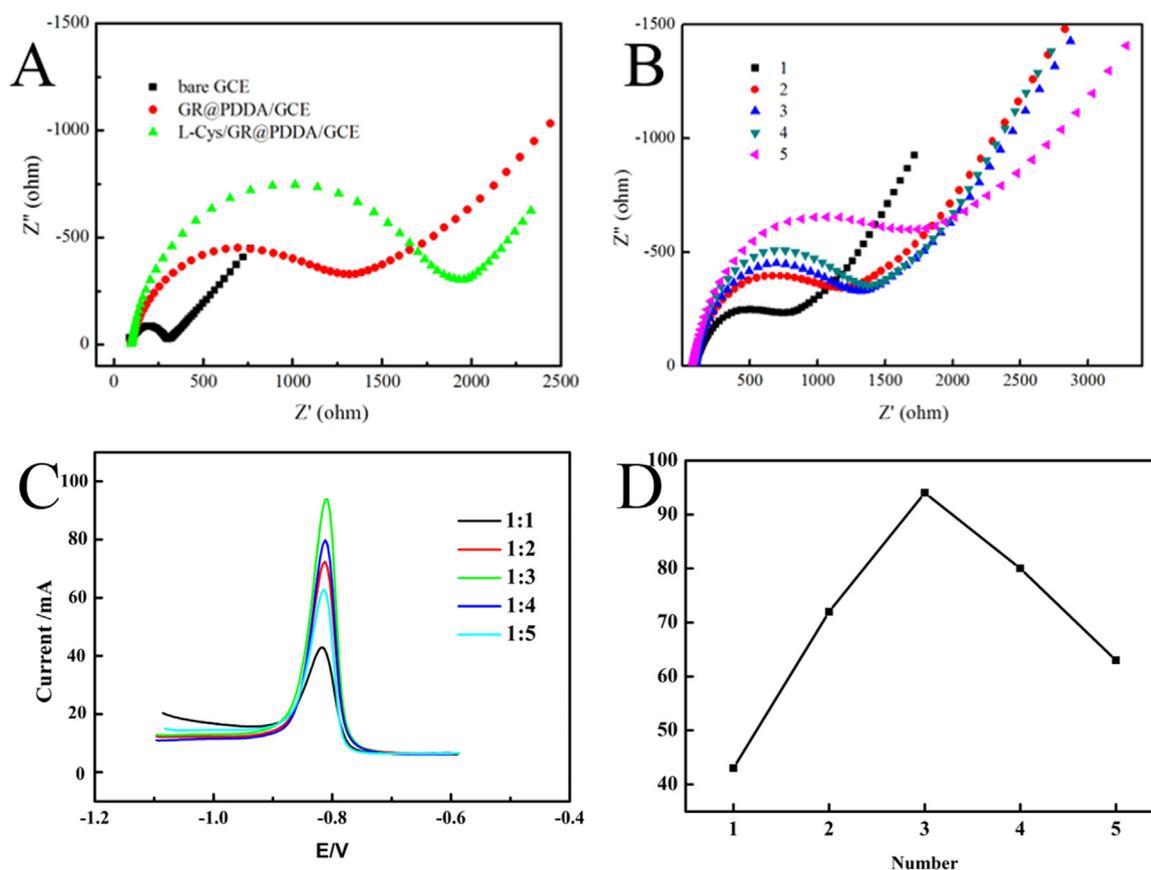


Fig. 3. EIS of bare GCE, GCE/GR-PDDA, GCE/GR-PDDA/L-Cys electrodes(A), and the EIS (B) and ASV (C), (D) of different proportion between GR and PDDA (Sample 1–5 were 1:1, 1:2, 1:3, 1:4, 1:5).

Furthermore, the single probe, ssDNA3-CdTe QDs, was also applied in the same experiment conditions (Fig. 4C) and the sensitivity of aptamer- DNA concatemer- CdTe QDs probe exhibited the superiority in analytical performance.

3.6. Selectivity and stability of the fabricated electrochemical biosensor

To examine the selectivity of this electrochemical biosensing system, HeLa cells and colorectal cancer cells were determined in the same detection conditions with 1.0×10^5 cells mL^{-1} . The electrochemical signals of HeLa cells and Hct116 cells were far lower than K562 tumor cells (Fig. 5), which illustrated our biosensor was with high specificity. To assess the reproducibility of electrochemical biosensor,

three different concentrations of K562 cells including 1.0×10^2 , 1.0×10^3 and 1.0×10^4 cells mL^{-1} were investigated. The relative standard deviation (RSD) of these three assays were 7.22%, 5.26% and 5.76%, respectively, which meant that electrochemical biosensor with good reproducibility.

4. Conclusions

In summary, a novel electrochemical biosensor for detecting tumor cells was proposed using the aptamer- DNA concatemer- CdTe QDs signal amplification probes and new electrode modification composites, GR- PDDA /L- Cys, in mercury-free ASV for detecting Cd^{2+} . This green and controllably amplified electrochemical biosensor displayed good

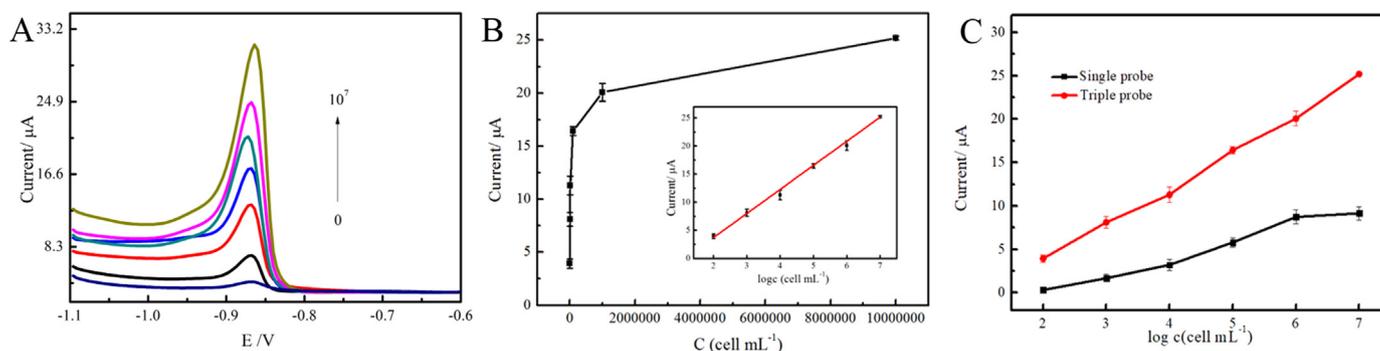


Fig. 4. (A) ASV of the electrochemical biosensor incubated with different concentrations of K562 cells: $0, 10^2, 10^3, 10^4, 10^5, 10^6$ and 10^7 cells mL^{-1} . (B) The relationship of electrochemical signal and cell concentration. The range of K562 cell concentration is from 10^2 to 10^7 cells mL^{-1} . The illustrated error bars represent the standard error of three independent measurements. (C) The ASV performance comparison of aptamer-DNA concatemer-CdTe QDs probe and single probe (ssDNA3-CdTe QDs).

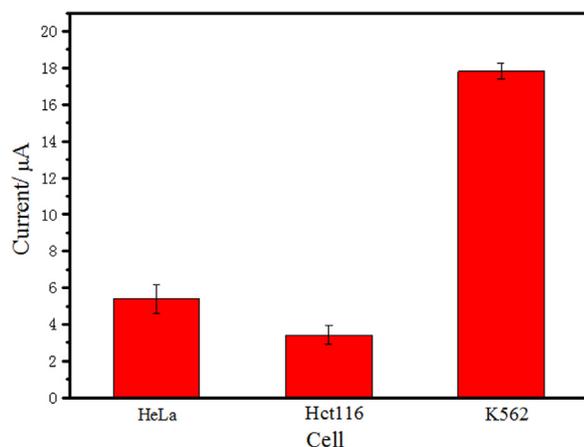


Fig. 5. Electrochemical signal of different tumor cells.

sensitivity, selectivity, wide linear range (1.0×10^2 to 1.0×10^7 cells mL^{-1}) and acceptable precision (LOD was 60 cells mL^{-1} ($S/N = 3$) for K562 cells). This design of signal amplification probes and the exploration of new composites would provide a promising method using for ultrasensitive biosensor preparation and green electrochemical detection of tumor cells.

Acknowledgments

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Appendix A. Supporting information

Supplementary data associated with this article can be found in the online version at doi:10.1016/j.bios.2018.09.076.

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