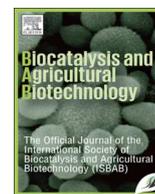




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Encapsulated enhanced silver nanoparticles biosynthesis by modified new route for nano-biocatalytic activity

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ABSTRACT

The study aims to investigate the green synthesis of silver nanoparticles (AgNPs) by *Phyllanthus acidus* extract (leaf and twig) as reducing and capping agents and evaluation of their bionanocatalytic activities for the first time. The synthesized AgNPs were characterized by UV-Visible spectrophotometry, Fourier Transform Infrared spectroscopy (FT-IR), Scanning Electron Microscopy (SEM), Dynamic Light Scattering (DLS) and Differential Scanning Calorimetry (DSC). The antioxidant ability of plant extract and AgNPs was analyzed using DPPH[•], NO[•] and OH[•] scavenging assays. SEM and DLS analysis confirmed the morphology of the nanoparticles to be spherical and the average size to be 48.36 and 164.30 nm. The most efficient scavenger was LAgNPs with EC₅₀ of 58.83 ± 1.65 µg/ml in DPPH[•] and TAGNPs in NO[•] (EC₅₀ 60.75 ± 1.59 µg/ml) and OH[•] (EC₅₀ 43.07 ± 1.62 µg/ml) scavenging assays. Further, they have enhanced the enzyme activity of α-amylase, cellulase, and xylanase over 2–6 folds. The results suggest that the Phytofabricated AgNPs from *P. acidus* can be exploited for industrial and biomedical applications.

1. Introduction

Nanoparticles have significant role in pharmaceutical, industrial and biotechnological applications. In recent years, nanoparticles and nanomaterials are used for different purposes, which includes targeted drug delivery systems, tumor imaging, diagnosis, cosmetics and biosensor (Mubarak Ali et al., 2011; Jeyaraj et al., 2013; Gopinath et al., 2016). Physical, chemical and biological approaches are employed for the synthesis of metal nanoparticles. The synthesis of nanoparticles through chemical and physical methods requires high pressure, energy, temperature and toxic chemicals. In this regard, green synthesis of metallic nanoparticles is of interest, primarily due to their ease of synthesis, eco-friendly approach and cost-effectiveness. In recent years, synthesis of AgNPs facilitated by phytocompounds is gaining significance due to their availability and eco-friendliness (Rout, 2012). Biosynthesis of AgNPs using plant sources such as *Eucalyptus chapmaniana* (Sulaiman et al., 2013), *Momordica charantia* (Ajitha et al., 2015), *Terminalia bellirica* (Patil et al., 2017), *Cochlospermum religiosum* (Sasikala et al., 2015) and many more have been reported. Silver nanoparticles (AgNPs) are proven to have potential antioxidant, antibacterial, antifungal, antiplasmodial and larvicidal properties (Abdel

Aziz et al., 2014; Bharathi et al., 2018a; Saxena et al., 2010; Elumalai et al., 2010). Antimicrobial activity of green synthesized AgNPs allows them to use in water filtration process, textiles and food industries (Van Dong et al., 2012; Balashanmugam and Kalaichelvan 2015; Sathiyavimal et al., 2018; Vasantharaj et al., 2019). AgNPs as amperometric biosensor using fungal enzyme can be one of the application for healthcare applications (Subrahmanyam et al., 2001). On the other hand, enhanced catalytic activity of NPs is already in use in different fields, such as urea sensing (Manikandan et al., 2017), glucose sensing, biodiesel production (Xie and Ma., 2010) and controlling blood coagulation (Sanfins et al., 2014). In healthcare, AgNPs can act as potential drug delivery systems as such (Kolanthai et al., 2017) or with appropriate polymeric nano carriers (Ranganathan et al., 2018) and for the treatment of cancer (Khargonekar et al., 2017; Nandagopal et al., 2016).

In the present study, for the biosynthesis of AgNPs, the leaves and twigs of *Phyllanthus acidus* (*P. acidus*) belonging to the family of Euphorbiaceae is used. The plant is abundantly found in Asian countries and their leaves are used to treat psoriasis (Burkill and Hanif, 2002), cough, asthma, bronchitis, soles (Caius, 1986), rheumatism, skin disorders (Morton and Miami 1987), hypertension, respiratory illness

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(Sousa et al., 2007), diabetics (Banik et al., 2010) and for hepatoprotection (Lee et al., 2006). In traditional medicine, the leaves, bark and the root of the tree are used to treat fever (Hadi and Bremner, 2001; Jain and Singhai, 2011), as antimicrobial (Jagessar et al., 2008) and nephroprotective (Vidya et al., 2013) agents. However, no reports exist related to synthesis of AgNPs from different parts of *P. acidus*. We are reporting for the first time on the biosynthesis of AgNPs from leaf and twig extract of *P. acidus*. We have demonstrated the phytochemical screening of aqueous leaf and twig extract, biosynthesis of AgNPs, characterization of the NPs and studied their free radical scavenging and bionanocatalytic activities.

2. Materials and methods

2.1. Collection of plant materials

The leaves and twigs of *P. acidus* were collected from CSIR-Central Leather Research Institute, Chennai, Tamil Nadu, India. The collected samples were washed with double distilled water to remove dust particles and air dried under shade condition.

2.2. Preparation of aqueous extract

5 g of powdered leaves was extracted with 100 ml of water at 60–70 °C using Soxhlet extractor, to obtain the aqueous extract of *P. acidus* (leaves and twigs) until the extract was clear. The extract was further filtered using WhatmanNo.1, and the filtrate was refrigerated (4 °C) until further use.

2.3. Qualitative phytochemical screening

Phytochemical screening of leaves and twigs extracts was performed for the detection of different phytochemicals such as tannins, flavonoids, saponins, alkaloids, quinones, coumarins, terpenoids and cardiac glycosides. All tests were performed according to the standard methods described by Al-Owaisi et al. (2014) and Sheel et al. (2014).

2.4. Synthesis of AgNPs

To synthesize AgNPs from *P. acidus*, 10 ml of aqueous extract was added to 90 ml of 1 mM AgNO₃ solution at room temperature and maintained in the dark. The silver nanoparticles using leaf extract (LAgNPs) and the twig extract (TAgNPs) were primarily checked for their plasmon resonance by UV-Visible spectrophotometer (Shimadzu UV 2450).

2.5. Antioxidant activity

The antioxidant activity of plant extracts and AgNPs was evaluated by 2,2-Diphenyl-1-picryl hydrazyl (DPPH) (Udaya Prakash et al., 2014), nitric oxide (NO) (Jagetia et al., 2004) and hydroxyl (OH) radical scavenging assays (Saeed et al., 2012). Different concentrations (50–250 µg/ml) of the NPs were studied against the radicals. The percent inhibition and effective concentration-50 (EC₅₀) were recorded.

2.6. Characterization of the synthesized AgNPs

The phytosynthesized AgNPs were characterized by different analytical methods, viz., Fourier transform infrared spectroscopy (FTIR), Scanning electron microscopy (SEM) (Phenom Pro), (JASCO FT-IR 4700), Dynamic light scattering analysis (DLS) (Malvern Instruments Ltd., Malvern, UK) and Differential scanning calorimetry (DSC, TA Instruments, model Q200).

2.7. Bio-nanocatalysis

The bionanocatalytic activity of α-amylase [EC 3.2.1.1], cellulase [EC 3.2.1.4] and xylanase [EC 253-439-7] was studied by DNS method using the substrates starch, cellulose and xylan respectively (Dastjerdi et al., 2015; Worthington and Worthington, 2011). Briefly, to 200 µg of the enzyme, equal concentration of the plant extract and the NPs were added. The substrate concentration was varied from 100 to 500 µg/ml. The substrate, enzyme and the NPs were allowed to react for 10 min at room temperature, after which 25 µl of DNS reagent was added and maintained at 40 °C for 10 min. The volume was made up to 300 µl using distilled water and the absorbance was measured at 540 nm. The enzyme activity was measured in Units/µg from the formula:

$$\text{Units}/\mu\text{g} = \frac{\text{Monosaccharide released } (\mu\text{g})}{\text{Enzyme added } (\mu\text{g}) \times \text{Incubation time}}$$

3. Results and discussion

3.1. Qualitative phytochemical screening

The experimental outcomes of the qualitative phytochemical screening of leaf and twig extracts of *P. acidus* are shown in Table 1. Phytochemical screening showed the presence of tannins, flavonoids, saponins, alkaloids, coumarins, terpenoids and cardiac glycosides in leaf extracts and flavonoids and alkaloids in twig extracts. Plant phenolics are known to possess high reducing potential (Pietta, 2000), which can be utilized for the reduction process in synthesis of AgNPs (Martinez-Castanon et al., 2008).

3.2. Biosynthesis of AgNPs

The aqueous extract of *P. acidus* (leaf and twig) was added to the AgNO₃ solution, and the color of the reaction mixture turned from colorless to dark brown which indicated the formation of AgNPs (Vanaja et al., 2013). The formation of AgNPs was confirmed by the surface plasmon resonance and the peaks were observed at 433 and 465 nm respectively for the AgNPs synthesized from the leaf extract (LAgNPs) and twig extract (TAgNPs) (Fig. 1). The mechanism of synthesis of AgNPs is attributed to the presence of the phytoconstituents, which is known to play a significant role in the formation, capping, and stabilization of AgNPs (Bharathi et al., 2018b).

3.3. Materials characterization

The synthesized AgNPs showed several peaks at 3440 cm⁻¹, 1637 cm⁻¹, 1370 cm⁻¹ and 598 cm⁻¹ were assigned to stretch vibrations of O–H of alcohol, N–H of primary amines, C–H of alkanes and C–Br of alkylhalides which indicates the complex nature of the phytochemicals present in the plant extract (Venugopal et al., 2017a,b) (Fig. 2). The shape and morphology of the AgNPs were analyzed by

Table 1
Phytochemical screening of *P. acidus* leaf and twig extract.

Phytochemical	<i>P. acidus</i>	
	Leaves	Twigs
Tannins	+	–
Flavonoids	+	+
Saponins	+	–
Alkaloids	+	+
Quinones	–	–
Coumarins	+	–
Terpenoids	+	–
Cardiac glycosides	+	–

+ Presence - Absence.

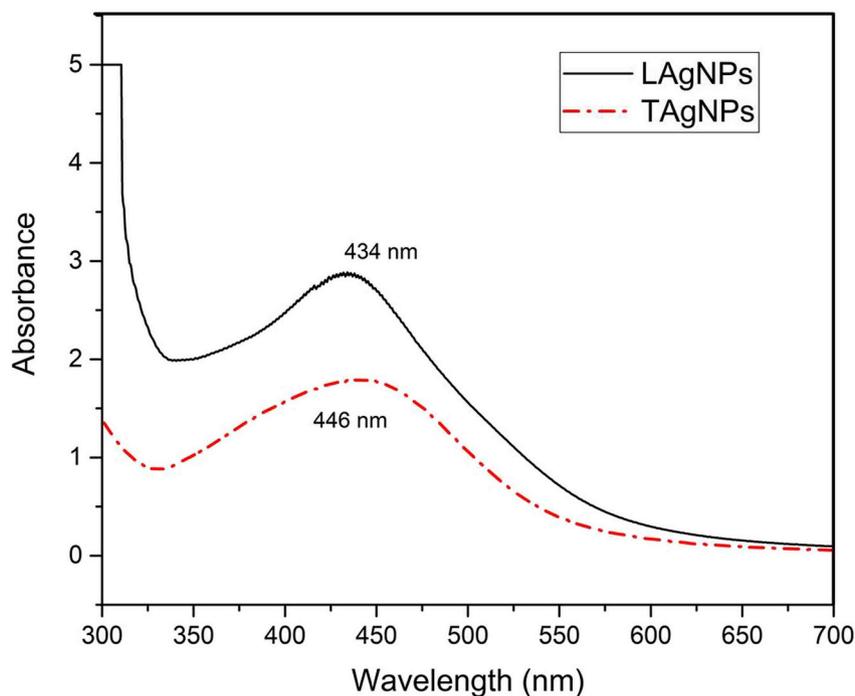


Fig. 1. UV-Visible spectrum of AgNPs synthesized using leaf and twig extracts of *P. acidus*.

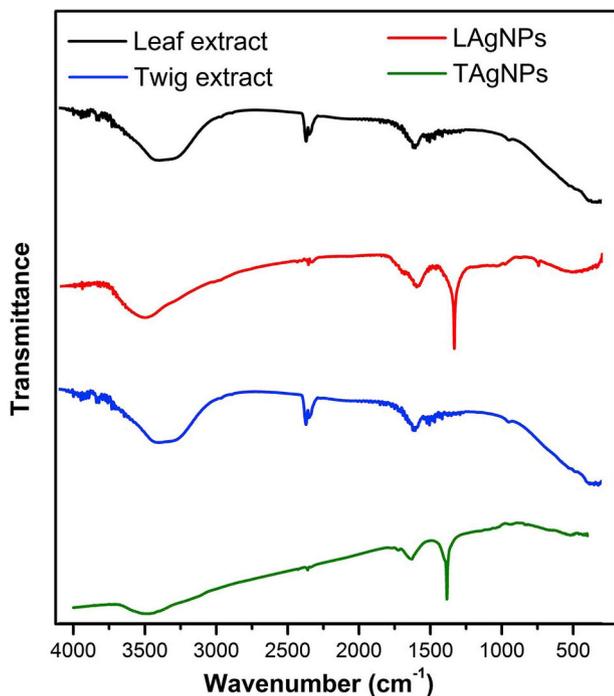


Fig. 2. FTIR spectrum of AgNPs synthesized using leaf and twig extracts of *P. acidus*.

SEM and the particles are in the shape of disc like spheres in the range of 75 and 112 nm in size (Fig. 3). The average size and distribution of AgNPs using DLS analysis were 48.36 and 164.30 nm (Fig. 4) with a negative charge of zeta potential as -7.97 and -22.4 respectively (Fig. 5). According to the DSC curve, the AgNPs synthesized showed an endothermic peak at 101.53 and 104.94 °C (Fig. 6). This suggests that phytochemicals responsible for the reduction of nanoparticles possess low thermal stability. Nanoparticles synthesized using the aqueous extract of *R. tuberosa* has also shown similar results (Vasantharaj et al., 2018).

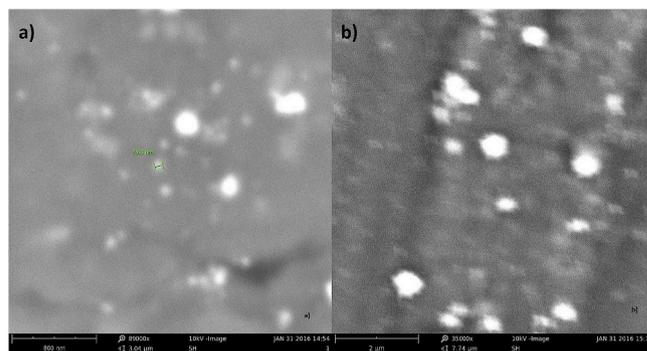


Fig. 3. SEM images of AgNPs synthesized using. (a) Leaf and (b) twig extracts of *P. acidus*.

3.4. Antioxidant activity

In the present study, the free radical scavenging ability of the synthesized NPs was better than the extracts as assessed by DPPH, NO and OH radical scavenging assays. The most efficient scavenger was LAgNPs with EC_{50} of 58.83 ± 1.65 $\mu\text{g/ml}$ in DPPH and TAgNPs in NO (EC_{50} 60.75 ± 1.59 $\mu\text{g/ml}$) and OH (EC_{50} 43.07 ± 1.62 $\mu\text{g/ml}$) radical scavenging assays. The EC_{50} values recorded for the DPPH, NO and OH radical scavenging assays are represented in Table 2. Compounds capable of inhibiting radicals are known to have beneficial effects on some aspect of inflammation and tissue damage. The NPs are found to possess high activity in scavenging DPPH, NO and OH radicals which might be due to the presence of antioxidant ligands of the plant extracts, on the NPs.

3.5. Bio-nanocatalytic activity

The phytofabricated AgNPs considerably enhanced the activity of α -amylase, cellulase and xylanase (Table 3). Among the three enzymes studied, the combination of α -amylase with AgNPs showed enhanced activity (1.0 U/ μg LAgNPs, 0.992 U/ μg TAgNPs and enzyme alone 0.18 U/ μg) (Fig. 7). The dramatic increase of enzyme activity may be due to

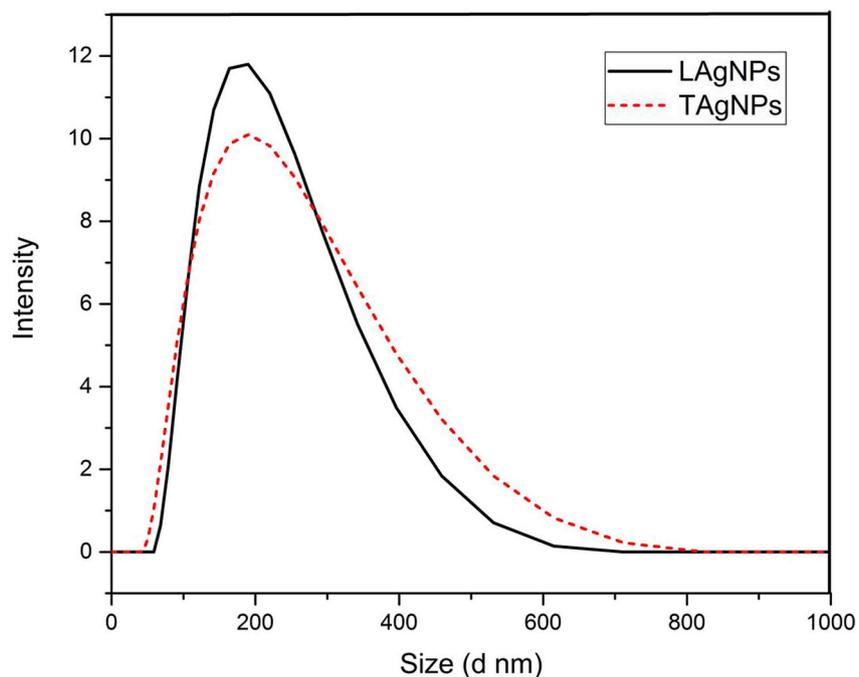


Fig. 4. Particle size distribution AgNPs synthesized using leaf and twig extracts of *P. acidus*.

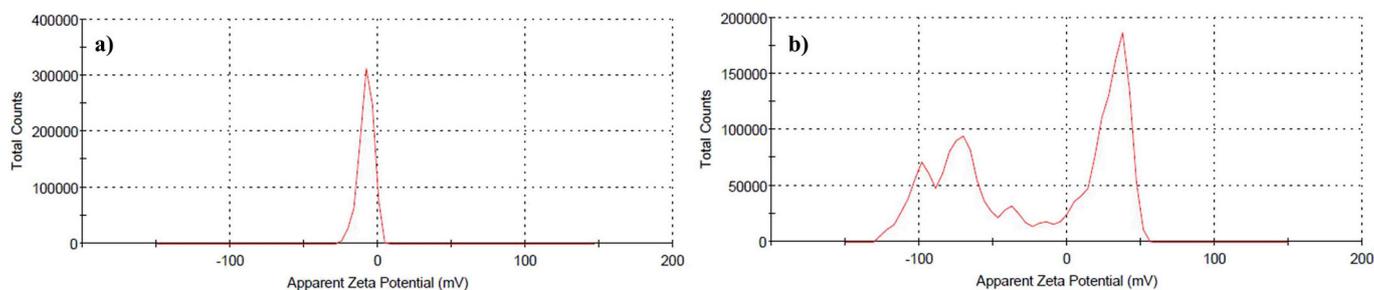


Fig. 5. Zeta potential of AgNPs synthesized using (a) leaf and (b) twig extracts of *P. acidus*.

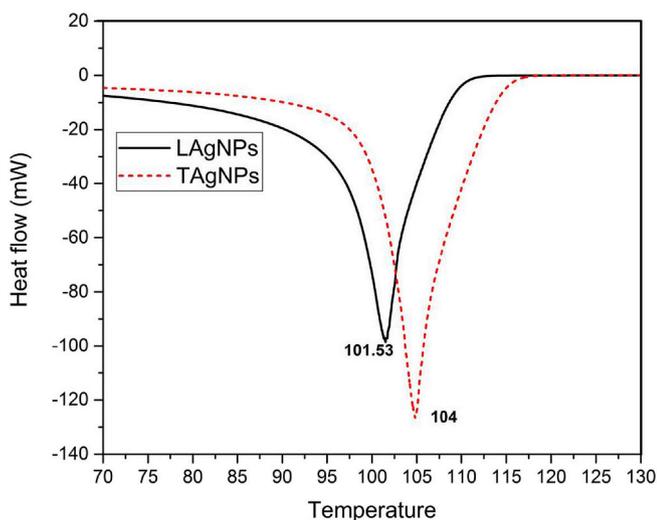


Fig. 6. DSC curve of AgNPs synthesized using leaf and twig extracts of *P. acidus*.

the conjugation of multiple sites of the enzymes with the NPs. Previously, the enhanced biocatalytic activity was observed in protease, α -amylase and lipase using various nano-composites (Jin et al., 2012; Wang et al., 2013; Jia et al., 2003; Wu et al., 2008, 2011; Dumri and Hung Anh, 2014). The enhanced activity of enzyme-NP conjugate in the

Table 2
Antioxidant activity of plant extract and AgNPs.

	DPPH*	NO*	OH*
L extract	128.09 \pm 1.87	86.03 \pm 1.87	188.65 \pm 1.97
LAgNPs	58.83 \pm 1.65	81.59 \pm 1.65	80.66 \pm 1.71
T extract	127.2 \pm 1.59	76.43 \pm 1.5	79.94 \pm 1.81
TAgNPs	82.6 \pm 1.5	60.75 \pm 1.59	43.07 \pm 1.62
Standard	BHT	Ascorbic acid	BHA
	107.72 \pm 1.62	127.34 \pm 1.54	74.87 \pm 1.79

Table 3
Effect of nanoparticles on α -amylase, cellulase and xylanase activity.

Substrate concentration (μ g/ml)	100	200	300	400	500
α -amylase (200 μ g/ml)	0.045	0.074	0.095	0.103	0.180
LAgNPs	0.449	0.599	0.777	0.872	1.005
TAgNPs	0.397	0.517	0.622	0.743	0.992
Cellulase (200 μ g/ml)	0.046	0.05	0.067	0.098	0.125
LAgNPs	0.177	0.209	0.271	0.296	0.316
TAgNPs	0.176	0.222	0.297	0.359	0.440
Xylanase (200 μ g/ml)	0.070	0.081	0.098	0.119	0.141
LAgNPs	0.459	0.527	0.549	0.588	0.641
TAgNPs	0.514	0.539	0.566	0.610	0.710

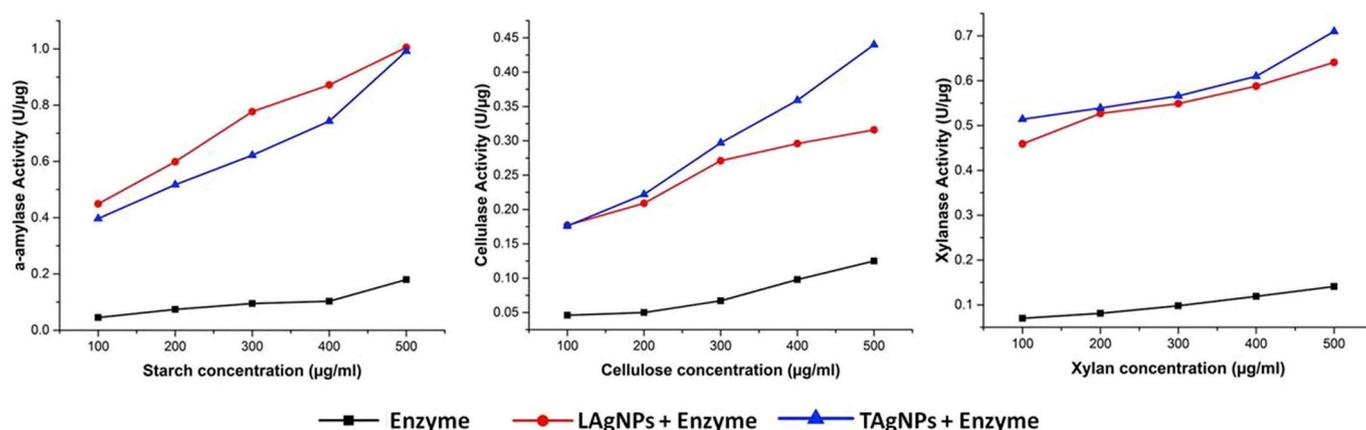


Fig. 7. Bio-nanocatalytic activity of AgNPs on α-amylase, cellulase and xylanase.

present study is still not fully evaluated, but specific characteristics of enzyme density, mass transport, NP morphology, NP surface chemistry, enzyme orientation (Venugopal et al., 2017a,b; Ardao et al., 2012; Ding et al., 2015) and trapping the ions on the cell surface (Zhang et al., 2011) are associated with the enhanced activity. Enzymes such as amylase, lipase and xylanase are widely employed in industries (Fujinami and Fujisawa., 2010). Most of the previous reports deal with the immobilization of enzymes to enhance enzyme activity (Talbert and Goddard., 2012). The present study suggests the use of nanoparticles to enhance the enzyme activity in various industrial sectors.

4. Conclusion

In summary, the study reveals the use of phytochemicals from the leaves and twigs of *P. acidus* as efficient capping and reducing agents for the synthesis of AgNPs. The most efficient scavenger was LAGNPs against DPPH* and TAGNPs against NO* and OH*, with their respective EC₅₀ values being 58.83 ± 1.65, 60.75 ± 1.59 and 43.07 ± 1.62 μg/ml. The biosynthesized AgNPs have shown excellent bio-nanocatalytic activities on α-amylase, cellulase and xylanase, as evidenced by an increase of enzyme activity by 2–6 folds when compared with the pure enzymes. The results indicate that the synthesized AgNPs has the potential to act as enhancers for various enzymes in biomedical and industrial applications.

Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.bcab.2019.101045>.

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