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# Pyrazole[3,4-*d*]pyridazine derivatives: Molecular docking and explore of acetylcholinesterase and carbonic anhydrase enzymes inhibitors as anticholinergics potentials

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## ABSTRACT

Recently, the pyridazine nucleus has been widely studied in the field of particular and new medicinal factors as drugs acting on the cardiovascular system. Additionally, a number of thienopyridazines have been claimed to possess interacting biological macromolecules and pharmacological activities such as NAD(P)H oxidase inhibitor, anticancer, and identified as a novel allosteric modulator of the adenosine A1 receptor. The literature survey demonstrates that coumarin, 1,2-pyrazole benzothiazole, and 1,3- thiazole scaffolds are the most versatile class of molecules. In this study, a series of substituted pyrazole[3,4-*d*]pyridazine derivatives (**2a–n**) were prepared, and their structures were characterized by Mass analysis, NMR, and FT-IR. These obtained pyrazole [3,4-*d*]pyridazine compounds were very good inhibitors of the carbonic anhydrase (hCA I and II) isoenzymes and acetylcholinesterase (AChE) with  $K_i$  values in the range of  $9.03 \pm 3.81$ – $55.42 \pm 14.77$  nM for hCA I,  $18.04 \pm 4.55$ – $66.24 \pm 19.21$  nM for hCA II, and  $394.77 \pm 68.13$ – $952.93 \pm 182.72$  nM for AChE, respectively. The possible inhibition mechanism of the best-posed pyrazole[3,4-*d*]pyridazine and pyrazole-3-carboxylic acid derivatives and their interaction with catalytic active pocket residues were determined based on the calculations.

## 1. Introduction

The pyrazoles are essential parts of many heterocyclic compounds. They have extensive workspace due to biologic activities and medicinal properties [1,2]. These molecules have potential applications such as antibacterial, antifungal, antibiotics, insecticide, pesticide, biosensors, etc., [3–7]. Pyrazole molecules are known to exhibit specific properties such as antiviral, antitumor, antidepressant, anti-inflammatory and antioxidant activities [8–11]. In the medical sector, the pyrazoles are used in the structure of some drugs such as Viagra, Celebrex, and Zerbaxa [12–14]. Additionally, pyrazole-pyridazine derivatives were used in agricultural products and drug researches due to their diverse biological activities [15]. Furthermore, substituted pyrazoles were used in

the polymer structures in terms of optoelectronic properties [16]. The commonly used methods for synthesis of the pyrazole-pyridazine derivatives are reported the cyclocondensation of hydrazine hydrate or monosubstituted hydrazines and 1,3-dicarbonyl compounds [17,18]. In this study, the synthesis of pyrazole-pyridazine was performed using various hydrazines and substituted pyrazole. The synthesized molecules were characterized by IR, NMR, and Mass analysis.

Carbonic anhydrase (CA) isozymes catalyze the reversible hydration of carbon dioxide ( $\text{CO}_2$ ) to produce bicarbonate anion ( $\text{HCO}_3^-$ ) and a proton ( $\text{H}^+$ ) [19,20]. Hence, these enzyme superfamily members including sixteen isoenzymes are primer controllers of pH outside and within the cell. Diversity in subcellular localization and catalytic activity, carbonic anhydrase isoenzymes show the main range of

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biological aims with several therapeutic potentials [21,22]. In particular, inhibitor compounds of hCA I and II isozymes in human cells are utilized as factors for treating glaucoma and edema. Membrane-bound brain-associated hCA IV isoenzyme is involved in pH regulation in neuron cells and glial cells analogously to the tumor-associated CA IX isoform. This plays an important role in progression, acidification of tumors, and metastasis in various cancers. A new approach to therapy of obesity and epilepsy is evaluated by inhibition behavior of special isozymes of CA [23,24]. Indeed, selective inhibitor compounds of CA isoforms in pathogenic microorganisms is gaining enhancing consideration as a new method of therapy of infectious disease [25].

Acetylcholine (ACh) is the main player as far as Alzheimer's disease (AD) is interesting which is synthesized by choline acetyltransferase (CAT). This area of the brain cell is affected in the pathophysiological phenomenon of the AD [26,27]. This reasons, finally, a marked difference in ACh and CAT synthesis. CAT amounts are decreased 58–90% in the AD, which raises the severity of dementia [28]. Acetylcholinesterase (AChE) is the enzyme accountable for the degradation of the ACh, which is markedly decreased in the AD. Therefore, these enzymes cholinergic synapses waiting for ACh and choline acetate is a key enzyme responsible for the hydrolysis. AChE enzyme, both three-dimensional structure of the active site amino acid sequence of the enzyme is related to many insects, and nerve transmission in the backbone is highly conserved [29]. Organic compounds with the potential to inhibit AChE enzyme or AChE inhibitors (AChEIs) are also very noteworthy due to their significance in the therapy of multiple neurodegenerative diseases. It is expected that AD, an extensive shape of dementia, will increase its prevalence globally and will be a significant problem for human health in the coming periods [30].

Herein, the facile synthesis to condensation reaction of pyrazole-3-carboxylic acid compounds (1a,b) and pyrazole[3,4-d]pyridazine (2a–n) derivatives have been reported, and their inhibition potential of hCA I, and II isoenzymes were investigated. The most suitable and potent AChE inhibition effects of these molecules are also studied.

## 2. Experimental

### 2.1. Materials and equipment

All chemicals used in this study were purchased from Sigma (Steinheim, Germany) and Merck (Darmstadt, Germany) companies. Bruker DRX-400 MHz FT-NMR spectrometer was employed to obtain the structure of molecules at <sup>1</sup>H (400 MHz) and <sup>13</sup>C (100 MHz) NMR spectra. The infrared spectra were recorded on a Perkin Elmer Precisely Spectrum. Samples were used as pellet form, and the measuring range was from 4,000 to 450 cm<sup>-1</sup>. An LC/MS spectrometer (Thermo Scientific TSQ-Quantum Access) was used to measure the mass spectrum. Melting points of these molecules were determined on an Electrothermal Gallenkamp apparatus and are uncorrected. The possible impurities and reaction steps of resulting compounds were tested by thin layer chromatography (TLC) using an aluminum TLC plate, silica gel coated with a fluorescent indicator of F254, Merck Millipore. Camag TLC devices envisaged these synthesized compounds. The synthesis of molecules has been described in previous studies [1].

### 2.2. Biological evaluation

#### 2.2.1. hCA isoenzyme purification and inhibition studies

Both hCA isoenzymes from human erythrocytes were purified via a simple single-step by method Sepharose-4B-L-Tyrosine-sulphanilamide affinity gel chromatography. For this purpose, the human erythrocyte samples were centrifuged at 13000 rpm for 25 min. Then, the solution was filtered to remove precipitate. hCA isoenzymes were isolated from the serum, which its pH was adjusted to 8.7 by adding solid Tris. Affinity column was equilibrated by buffer solution (25 mM Tris-HCl/0.1 M Na<sub>2</sub>SO<sub>4</sub>) at the pH 8.7. The serum was loaded to affinity gel and

washed with buffer solution (25 mM Tris-HCl/22 mM Na<sub>2</sub>SO<sub>4</sub> at the pH 8.7. hCA isoenzymes were taken from the column in fractions of 2 mL. The hCA isoenzymes activity were measured by following the change at absorbance a specific (348 nm) of p-nitrophenylacetate (PNA) to p-nitrophenolate ion over a period of 3 min at room temperature (25 °C) using a spectrophotometer (Thermo Scientific, UV-Vis Spectrophotometer). There was 0.4 mL of 0.05 M Tris-SO<sub>4</sub> buffer (pH 7.4), 0.3 mL of 3 mM PNA, 0.2 mL of H<sub>2</sub>O, and 0.1 mL of enzyme solution in a test tube content of this reaction. Esterase activity assays were identified from a series of experiments at three different pyrazole derivatives. CA inhibitory effects of pyrazole-3-carboxylic acid compounds (1a,b) and pyrazole[3,4-d]pyridazine derivatives (2a–n) were studied by Verpoorte et al. [31] and previous studies [32–35]. Enzyme activity was measured at 348 nm by UV-Vis spectrophotometer with p-nitrophenyl acetate (PNA) substrate.

#### 2.2.2. AChE inhibition study

AChE inhibitory effects of pyrazole-3-carboxylic acid compounds (1a,b) and pyrazole[3,4-d]pyridazine derivatives (2a–n) were studied by per Ellman et al. [36] and previous studies [37–39]. It was studied at 412 nm spectrophotometrically using acetylthiocholine iodide as a substrate for the enzymatic reaction. 5,5'-Dithio-bis(2-nitro-benzoic) acid was utilized for the measurement of the AChE activity [40].

### 2.3. Computational studies

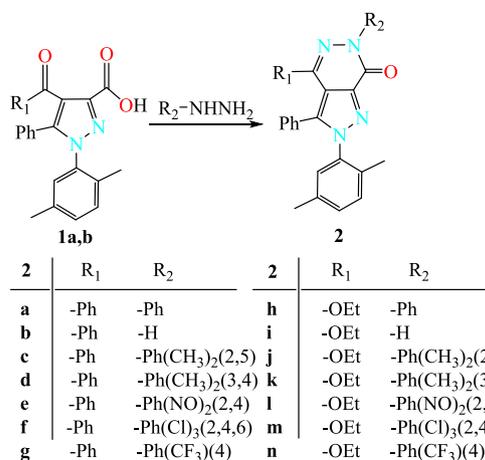
Computational studies were achieved using Small Drug Discovery Suites software (Schrödinger 2017-2, LLC, USA). 2D structures of pyrazole-3-carboxylic acid compounds were drawn, and 3-Dimensional structures of the compounds were achieved using Maestro 11.4. The 3-Dimensional structures were prepared by obtaining accurate protonation state, and molecular geometries at physiological pH with the LigPrep module [15]. It was reached to X-Ray crystal structures of targeted receptors (h CA I PDB code: 4WR7, h CA II PDB code: 5AML, and AChE PDB code: 4M0E) from RCSB Protein Data Bank. The receptors were chosen for molecular docking process due to their low resolution (1.5 Å, 1.36 Å, and 2 Å, respectively). Afterward, the receptor structures prepared with the protein preparation wizard as reported on our previous studies. The bond orders and charges were primarily allocated. Missing side chains and hydrogen atoms were secondarily appended to the structure of the receptor. Amino acids were thirdly ionized at physiological pH with Propka module. Water molecules that were formed less than three contacts with the receptor were fourthly fended off. Lastly, energy minimization was performed with OPLC force field. Prepared receptors were used at detection of the binding pocket and molecular docking process.

In order to designate active catalytic pocket of the receptors and druggability of catalytic active pocket, top-ranked potential protein binding pocket of the receptors were identified with SiteMap module. Following designation catalytic active pocket, for verifying the reliability of molecular docking process, docking validation studies were carried out as reported on our previous studies. Following docking validation, prepared compounds, ligands, were docket into the catalytic active pocket of the receptor using Glide module as reported at our previous studies. Grid box was generated by selecting co-crystallized ligand into the predicted catalytic active pocket. The molecular docking calculation was performed on Standard Precision (SP) and Extra Precision (XP) settings [15].

## 3. Results and discussion

### 3.1. Chemistry

The pyrazole[3,4-d]pyridazine compounds are the most commonly studied subject in heterocyclic chemistry due to their biologic activities [17]. Therefore, pyrazole[3,4-d]pyridazine compounds were prepared,



Scheme 1. Synthesis of pyrazole[3,4-d]pyridazines (2a–n).

and spectroscopic methods confirmed their structures. Initially, pyrazole-3-carboxylic acid (1a,b) compounds were prepared by our research group. Pyrazole[3,4-d]pyridazine (2a–n) derivatives were directly obtained by cyclization reaction synthesized 1a and b compounds and various hydrazines as shown Scheme 1. The yields of 2a–n compounds were between 25 and 65%. All of the synthesized 2a–n compounds had two or three important rings. The structures of obtained compounds were characterized via spectroscopic methods.

### 3.2. Biochemistry results

Patients with AD have decreased cerebral generation of ACh and impaired cortical cholinergic function. Thus, reinforcement of cholinergic activity therapeutically can reduce cognitive decline in dementia [41,42]. AChEIs including tacrine, donepezil, rivastigmine, and galantamine, which increase ACh levels in the brain cells via the inhibition of AChE enzyme, are considered first-line pharmacotherapy for dementia. Evidence obtained is that ChEIs affect the development of the cognitive concept. However, the majority of these surveys were conducted in North America and European Caucasians [43,44]. In this work, effects of pyrazole-3-carboxylic acids (1a and b) and pyrazole [3,4-d]pyridazine (2a–n) derivatives against hCA I, hCA II, and AChE enzymes were studied. The compounds in hCA I, hCA II and AChE enzymes were studied as *in vitro* and these effects are evaluated with the data in Table 1.

CA I isoenzyme, which is seen as a symptom of hemolytic anemia, is evaluated as a result of the presence of abnormal levels in blood [45]. The slow cytosolic isoform of hCA I was inhibited by the pyrazole-3-carboxylic acids (1a and b) and pyrazolyl[3,4-d]pyridazine (2a–n) derivatives, with  $K_i$  values ranging between  $9.03 \pm 3.81$  and  $55.42 \pm 14.77$  nM. Furthermore, (2g) and (2f) have  $K_i$  values of  $9.03 \pm 3.81$  and  $10.32 \pm 2.64$ , respectively. These values correspond to the most powerful properties of hCA I isoenzyme inhibition. The  $K_i$  values of acetazolamide (AZA), a drug used standard and clinically, are  $178.11 \pm 40.03$  nM (Table 1).

In addition, CA II isozyme is often related to various diseases for instance osteoporosis, renal tubular acidosis, and glaucoma [46]. The hCA II isozyme was also efficiently inhibited by the pyrazole-3-carboxylic acids (1a and b) and pyrazole[3,4-d]pyridazine (2a–n) derivatives evaluated here. These molecules appeared to be able to inhibit hCA II, with  $K_i$  values ranging from  $18.04 \pm 4.55$  nM to  $66.24 \pm 19.21$  nM. The results obtained in this study indicates better values compared to clinically used drug acetazolamide ( $K_i$  of  $194.12 \pm 68.83$  nM). All the investigated pyrazole-3-carboxylic acids (1a and b) and pyrazole[3,4-d]pyridazine (2a–n) derivatives demonstrated marked inhibition against hCA II. However, the compounds of (2g) and (2m) exhibited very good inhibition properties against

cytosolic hCA II with  $K_i$  values of  $18.04 \pm 4.55$  and  $19.24 \pm 5.23$  nM, respectively (Table 1).

The inhibitory effects of the synthesized pyrazole-3-carboxylic acids (1a and b) and pyrazole[3,4-d]pyridazine (2a–n) derivatives on AChE are shown in Table 2. The AChE inhibition profiles of the compounds evaluated here were quite interesting. Overall, the pyrazole-3-carboxylic acids (1a and b) and pyrazole[3,4-d]pyridazine (2a–n) derivatives had excellent inhibitory activity with  $K_i$  values ranging from  $394.77 \pm 68.13$  nM to  $952.93 \pm 182.72$  nM. Moreover, Tacrine, as a standard AChE inhibitor, was used. The  $K_i$  values of  $1005.02 \pm 198.66$  nM toward AChE were obtained. According to experimental data, the inhibition of AChE of pyrazole-3-carboxylic acids (1a and b) and pyrazole[3,4-d]pyridazine (2a–n) derivatives is much better than standard drug. The compounds of (2i) and (2l) showed excellent inhibitions profile against AChE with  $K_i$  values of  $394.77 \pm 68.13$  and  $463.86 \pm 93.07$  nM, respectively (Table 1).

$\text{HCO}_3^-$  and  $\text{CO}_2$  conducts the pH buffering among intracellular and extracellular spaces. The adjustment of both parts is under the control of CAs, which catalyzes the reversible conversion of  $\text{H}_2\text{O}$  and  $\text{CO}_2$  into  $\text{HCO}_3^-$  and  $\text{H}^+$  [47]. Hence, with this vital evidence, brain CA records as a significant aim for management of epileptic seizure syndromes. Diverse isoenzymes of CAs are available in the mammalian brain; hCA II isozyme is extensively expressed in myelin sheaths, and oligodendrocytes as well as choroid plexus, which is engaged in several neurological disturbances like epileptogenesis [48]. In this paper, evaluation of the effects of pyrazole-3-carboxylic acids (1a,b) and pyrazole[3,4-d]pyridazine (2a–n) derivatives on AChE, and hCA I and II isoenzymes was an important purpose, and inhibition results are present in Table 1. Compounds for the isoforms of HCA I and II are (2g), (2f) and (2j). However, when these compounds are compared, it is seen that the best inhibitor is (2g) and (2f). (2j) shows a weaker inhibitory property compared to others. This was also the case for AChE and gave the best inhibitory property (2i) and (2l). Similarly, the compound showing a weaker inhibitory property is (2a). In addition, for hCA II, (2g) and (2m) is the best inhibitor (Fig. 4).

### 3.3. Computational studies

Computational tools have become essential to understand drug-receptor interaction and inhibition mechanism. Therefore, interactions and a binding score of the best-posed compounds were analyzed with a computational method. The results obtained on the AChE, hCA I and II receptors were examined using pyrazole-3-carboxylic acid and pyrazole [3,4-d]pyridazine derivatives to compare with TAC and AZA reference inhibitors. The reference inhibitors, pyrazole-3-carboxylic acid and pyrazole[3,4-d]pyridazine derivatives were similarly docked into the catalytic active pocket of receptors. The exposure viewer module tested the interactions between the receptors and the best-exposed ligands. The studies on molecular docking revealed that all the pyrazole[3,4-d]pyridazine and pyrazole-3-carboxylic acid derivatives exhibited higher binding affinity towards hCA I, hCA II, and AChE, and also most of the compounds exactly fit into the active catalytic pocket. The glide docking results are presented in Table 2. The glide docking scores were well suited on a large scale with results of *in vitro* studies, but the theoretical values were not compared with the experimental affinity values for each compound. Herewith, the elucidating for binding mechanisms and interactions between receptors and ligands were the main objectives of this section of the present study. In this study, four parameters including Glide score, docking score, non-bonded interactions, and H-bonds were considered for docking analysis. The reliability of the docking process presented in Fig. 1. The figure shown that docking process carried out at high accuracy due to re-docked and co-crystallized ligands are well located in the active catalytic pocket.

Computational studies have shown that the compounds could easily fit into the active catalytic pocket of hCA I. However, it has been shown only the most active compound 2g as seen in Fig. 2a. The compound

**Table 1**

Human carbonic anhydrase isoenzymes (CA I, and II), and AChE enzymes inhibition study values of pyrazole-3-carboxylic acids and pyrazole[3,4-*d*] pyridazine derivatives (**1a,b**, and **2a-n**).

Compounds	IC <sub>50</sub> (nM)				K <sub>i</sub> (nM)					
	hCA I	r <sup>2</sup>	hCA II	r <sup>2</sup>	AChE	r <sup>2</sup>	hCA I	hCA II	AChE	
1a	24.84	0.9724	34.12	0.9816	894.12	0.9602	28.42 ± 5.08	40.16 ± 11.27	674.03 ± 105.73	
1b	36.03	0.9919	47.03	0.9704	1022.16	0.9811	42.66 ± 11.45	52.63 ± 15.62	904.72 ± 205.84	
2a	20.88	0.9504	29.15	0.9815	1105.94	0.9915	24.63 ± 9.23	38.53 ± 8.43	952.93 ± 182.72	
2b	41.02	0.9815	49.14	0.9618	791.05	0.9699	40.28 ± 7.33	54.84 ± 15.82	643.84 ± 97.36	
2c	45.92	0.9581	53.82	0.9861	704.17	0.9851	51.37 ± 13.70	66.24 ± 19.21	583.61 ± 137.41	
2d	33.14	0.9632	36.04	0.9982	916.06	0.9506	37.60 ± 11.36	44.36 ± 9.25	730.73 ± 193.61	
2e	38.15	0.9811	44.26	0.9655	1001.51	0.9821	43.95 ± 13.55	42.18 ± 14.62	817.32 ± 103.83	
2f	8.84	0.9409	15.34	0.9738	816.94	0.9593	10.32 ± 2.64	23.11 ± 7.35	721.54 ± 137.83	
2g	9.41	0.9714	12.94	0.9903	834.25	0.9924	9.03 ± 3.81	18.04 ± 4.55	703.47 ± 176.07	
2h	24.66	0.9395	29.80	0.9525	702.93	0.9592	29.64 ± 10.44	34.58 ± 10.24	563.85 ± 96.90	
2i	30.18	0.9916	36.26	0.9705	506.62	0.9905	43.71 ± 9.56	50.23 ± 14.06	394.77 ± 68.13	
2j	51.73	0.9791	59.93	0.9625	806.44	0.9958	55.42 ± 14.77	53.34 ± 12.66	638.05 ± 136.07	
2k	29.15	0.9814	35.83	0.9816	1082.48	0.9641	30.25 ± 6.82	39.14 ± 6.56	876.12 ± 205.87	
2l	35.81	0.9731	42.51	0.9506	551.05	0.9815	41.39 ± 9.26	40.72 ± 15.84	463.86 ± 93.07	
2m	12.60	0.9548	20.94	0.9773	795.33	0.9690	14.66 ± 3.52	19.24 ± 5.23	683.08 ± 157.41	
2n	14.03	0.9942	18.04	0.9930	626.01	0.9882	13.20 ± 2.95	22.51 ± 6.08	547.06 ± 150.38	
AZA*	171.13	0.9425	181.55	0.9709	–	–	178.11 ± 40.03	194.12 ± 68.83	–	
TAC**	–	–	–	–	1116.74	0.9830	–	–	1005.02 ± 198.66	

\* Acetazolamide (AZA) was used as a standard inhibitor for both human carbonic anhydrases I, and II isoenzymes (hCA I, hCA II).

\*\* Tacrine (TAC) was used as a standard inhibitor for acetylcholinesterase (AChE) enzyme.

**Table 2**

Glide scores (kcal/mol) of pyrazole-3-carboxylic acids (**1a,b**) and pyrazole[3,4-*d*]pyridazine (**2a-n**) derivatives into the active catalytic pocket of hCA I, and hCA II, and AChE.

Compounds	hCA I		hCA II		AChE	
	Glide XP		Glide XP		Glide XP	
	Docking score	Glide score	Docking score	Glide score	Docking score	Glide score
1a	-3.520	-3.520	-4.033	-4.033	-7.078	-7.078
1b	-3.701	-3.701	-7.293	-7.293	-5.776	-5.776
2a	-4.507	-4.507	-5.098	-5.098	-8.481	-8.481
2b	-4.160	-4.422	-5.181	-5.443	-6.270	-6.881
2c	-4.215	-4.215	-5.658	-5.658	-8.218	-8.218
2d	-4.378	-4.378	-4.527	-4.527	-8.730	-8.730
2e	-4.678	-4.678	-4.474	-4.474	-7.658	-7.658
2f	-4.942	-4.942	-4.215	-4.215	-8.378	-8.378
2g	-4.985	-4.985	-4.252	-4.252	-9.009	-9.009
2h	-3.978	-3.978	-3.909	-3.909	-8.786	-8.786
2i	-4.555	-4.731	-4.344	-4.520	-7.797	-7.976
2j	-4.525	-4.525	-4.571	-4.571	-8.083	-8.083
2k	-4.704	-4.704	-4.155	-4.155	-9.083	-9.083
2l	-4.641	-4.641	-4.497	-4.497	-6.007	-6.007
2m	-4.205	-4.205	-4.499	-4.499	-8.638	-8.638
2n	-4.329	-4.329	-4.093	-4.093	-8.547	-8.547
AZA*	-7.497	8.331	-7.337	-8.178	–	–
TAC**	–	–	–	–	-8.977	-8.978

\* Acetazolamide (AZA) was used as standard inhibitor for human carbonic anhydrase isoenzymes I and II (hCA I and II).

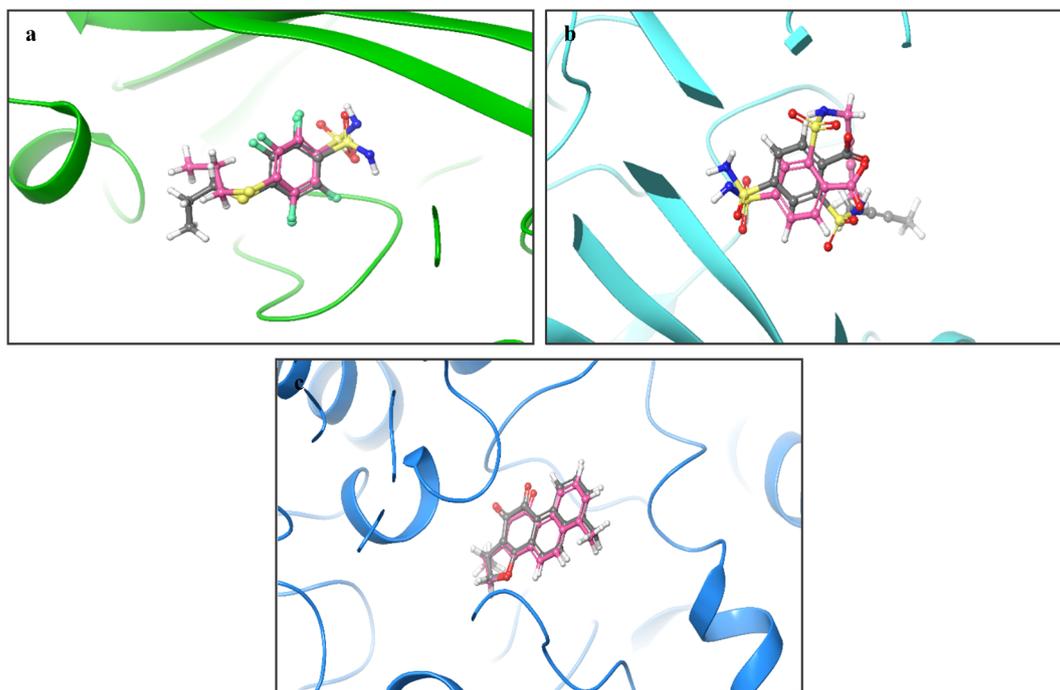
\*\* Tacrine (TAC) was used as standard inhibitor for acetylcholinesterase (AChE) enzyme.

and reference inhibitor, **AZA**, occupy same binding pocket (data not shown).

It has been found that compound **2g** has the best scores with -4.985 kcal/mol towards hCA I. The best-scored compound is also most active inhibitor for hCA I as for the results of *in vitro* studies. The compound formed a hydrogen bond with His64 residue into the active catalytic pocket of hCA I and was also surrounded by hydrophobic amino acid residues including Phe91, Leu198, Pro202, and Tyr204 (Fig. 3a). To confirm of similarity between **2g** and reference inhibitor, **AZA** was docked into the active catalytic pocket of hCA I. Glide score of the best-scored pose of **AZA** was detected as -7.497 kcal/mol towards hCA I. In contrary to *in vitro* experiment, the score showed that **AZA** was theoretically more effective inhibitor than **2g**. **AZA** formed a hydrogen bond with Act304 and formed a salt bridge with Zn301. Additionally,

the reference inhibitor formed  $\pi$ - $\pi$  interaction with His200. It was detected that best-posed **2g** and **AZA** were surrounded by very similar hydrophobic amino acid residues (Fig. 3d). **2g** showed pretty similar interactions with former reported inhibitors [27,49,50].

**1b** with -7.293 kcal/mol score was determined as the best-scored compound towards hCA II with the computational study. However, *in vitro* studies revealed that **2g** was the most active inhibitor of hCA II. For that reason, the interaction and binding mechanism of **2g** with -4.252 kcal/mol score with active pocket residues of hCA II were analyzed in detail. **2g** well located into the active catalytic pocket of the receptor (Fig. 2b). **2g** formed a hydrogen bond with Gln92 residue, and exhibited hydrophobic interactions Leu60, Ile91, Pro202, Leu204, Val207, and Trp209 with amino acid residues (Fig. 3b). **AZA** acts as a standard inhibitor for HCAs. Thr199 and Asn67 are involved in the

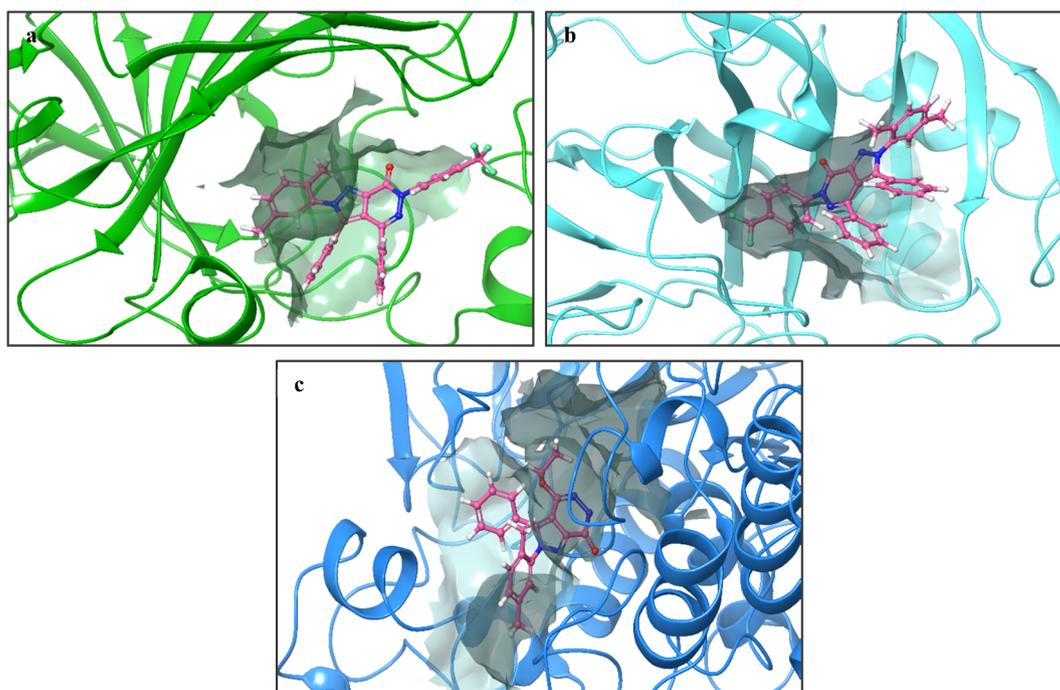


**Fig. 1.** Validation of docking. Receptors are shown in the ribbon model; blue: AChE; green: hCA I and cyan: hCA II. Disposed along poses of the crystalline ligand is represented by the gray ball and the ball is placed ligands magenta and bars with modeling is shown.

formation of hydrogen bonds with amino acid residues. In addition, it forms a salt bridge with Zn265 for the hCA II isoenzyme. (Fig. 3e). Best-posed **2g** and **AZA** were surrounded with polar binding pocket. It has been noticed that Gln92 amino acid is a key residue for hCA II isoenzyme. The results agreed with formerly reported inhibitors [49,51,52].

According to docking results, almost all of pyrazole[3,4-*d*]pyridazine derivatives exhibited high binding affinity ranges between

–5.776 and –9.083 kcal/mol into AChE receptor as TAC standard inhibitor. While **2k** has the best binding affinity towards AChE, **2i** is the most active inhibitor for AChE *in vitro* studies and exhibited well binding affinity with –7.976 kcal/mol towards the receptor (Table 2). Thus, it was analyzed interaction and binding mechanism between compound **2i** and catalytic active pocket residues of AChE. **2i** well positioned into the active catalytic pocket of the receptor (Fig. 2c). **2i** formed a hydrogen bond with Phe295 and Arg296 residue and with a



**Fig. 2.** The lowest energy conformation of ligands into AChE, hCA I and hCA II receptors, respectively. (a) Best-pose of **2g** into the catalytic active pocket of hCA I, (b) Best-pose of **2g** into the catalytic active pocket of hCA II, (c) Best-pose of **2i** into the catalytic active pocket of AChE. Receptors were shown in the ribbon model; green: hCA I, cyan: hCA II, blue: AChE. Ligands represent the magenta color with carbon balls and rods. Catalytic active pockets are represented as solid surfaces.

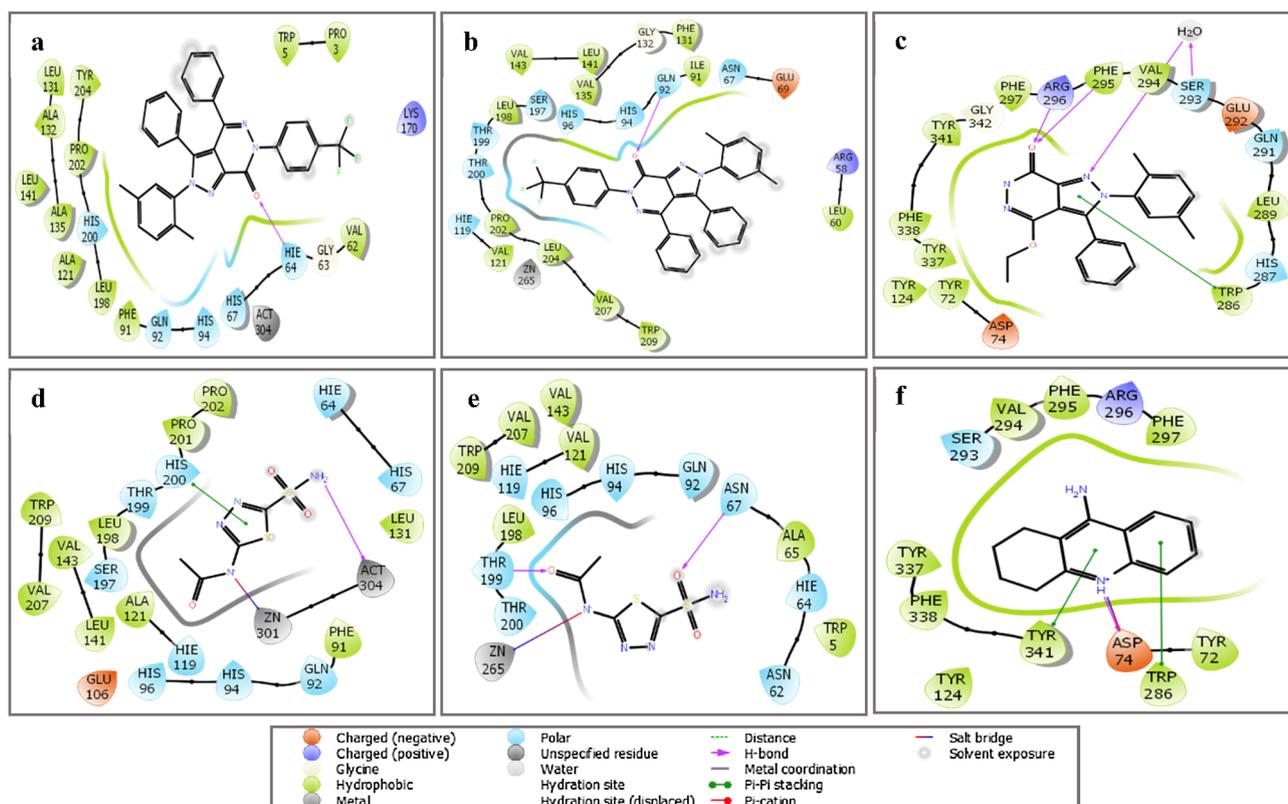


Fig. 3. 2D interaction profile between receptor and ligand; (a) 2g-hCA I, (b) 2g-hCA II, (c) 2i-AChE, (d) AZA-hCA I, (e) AZA-hCA II, (f) TAC-AChE.

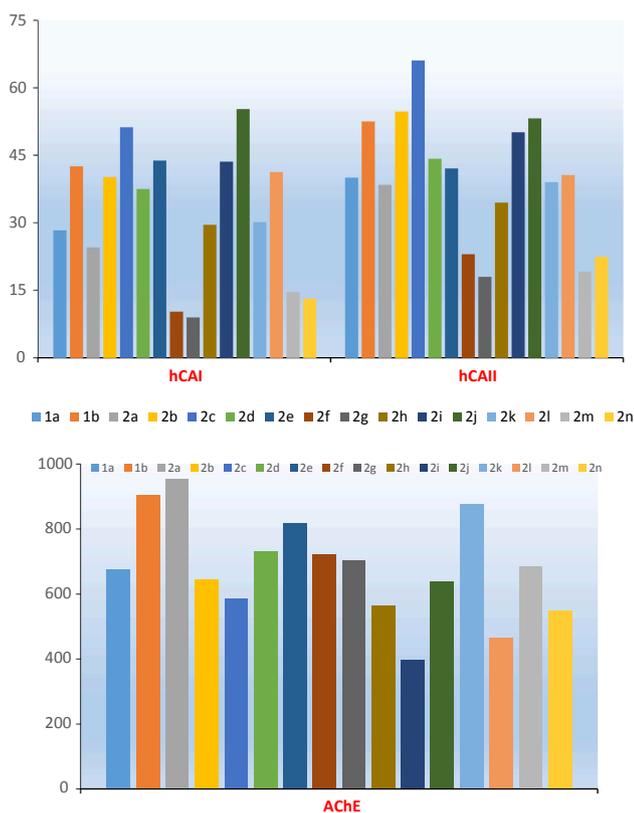


Fig. 4.  $IC_{50}$  and  $K_i$  values of pyrazole-3-carboxylic acids (1a and b) and pyrazole[3,4-d]pyridazine (2a-n) against human carbonic anhydrase I and II isoforms and acetylcholinesterase (AChE) enzyme.

HOH715 water molecule, as well. Furthermore, a  $\pi$ - $\pi$  stacking interaction between the 2i compound and Trp286 amino acid residue were observed. Compound 2i was surrounded by several residues of hydrophobic amino acid including residues of Tyr72, Val294, Leu289, Phe195, Tyr337, Phe338, Trp286, Phe297, and Tyr341 (Fig. 3c). The binding pocket of TAC and 2i is very similar to one another due to the hydrophobic amino acid residues. (Phe195, Val294, Try72, Tyr337, Phe297, Phe338, and Tyr341). Compound 2i shown pretty similar interaction as former reported inhibitors [53–55].

#### 4. Conclusion

Finally, a series of pyrazole-3-carboxylic acids (1a and b) and pyrazole[3,4-d]pyridazine (2a-n) derivatives were synthesized. These compounds were analyzed for AChE inhibition properties and hCAs I and II isoforms. Pharmacophores have gained significant importance in many different synthetic drug designs due to their various biological activities and the scaffolds of various bioactive natural compounds. Among the scaffolds of these compounds, thiazole and coumarin have recently attracted attention as the leading pharmacophore since they have a wide variety of pharmacological activities in the field of medicinal chemistry. Some of the biological properties have anticholinesterase activity (AChE and BChE), carbonic anhydrase I, II and IX, and potential antioxidant inhibition and aflatoxigenic activities. These molecules exhibited inhibition behavior at nanomolar concentration level against those enzymes. The discovery of novel inhibitors of AChE, one of the cholinesterase enzymes, is particularly important for Alzheimer's disease. To see this inhibition in the nanomolar scale from pyrazole derivatives made the study important. According to docking results, hCA I, hCA II, and AChE enzymes inhibited by all compounds through interactions together with H-bonds, electrostatic and hydrophobic interactions.

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## Declaration of Competing Interest

The authors declared that there is no conflict of interest.

## References

- [1] A. Çetin, İ. Bildirici, A study on the synthesis and antimicrobial activity of 4-acylpyrazoles, *J. Saudi Chem. Soc.* 22 (3) (2018) 279–296.
- [2] F.K. Keter, J. Darkwa, Perspective: the potential of pyrazole-based compounds in medicine, *BioMetals* 25 (1) (2012) 9–21.
- [3] A.A. Bekhit, T. Abdel-Azimi, Design, synthesis and biological evaluation of some pyrazole derivatives as anti-inflammatory-antimicrobial agents, *Bioorg. Med. Chem.* 12 (8) (2004) 1935–1945.
- [4] R. Nagamallu, B. Srinivasan, M.B. Ningappa, A.K. Kariyappa, Synthesis of novel coumarin appended bis(formylpyrazole) derivatives: Studies on their antimicrobial and antioxidant activities, *Bioorg. Med. Chem. Lett.* 26 (2) (2016) 690–694.
- [5] R.H.O. Montes, R.M. Dornellas, L.A.J. Silva, A.L. Squizzato, E.M. Richter, R.A.A. Munoz, Amperometric determination of the insecticide fipronil using batch injection analysis: comparison between unmodified and Carbon-nanotube-modified electrodes, *J. Solid State Electrochem.* 20 (9) (2016) 2453–2459.
- [6] P.W. Walker, P.G. Story, G.C. Hose, Comparative effects of pesticides, fenitrothion, and fipronil, applied as ultra-low volume formulations for locust control, on non-target invertebrate assemblages in Mitchell grass plains of south-west Queensland, Australia, *Crop Prot.* 89 (2016) 38–46.
- [7] V.C. Diculescu, A.-M. Chiorcea-Paquim, A.M. Oliveira-Brett, Applications of a DNA-electrochemical biosensor, *TrAC Trends Anal. Chem.* 79 (May 2016) 23–36.
- [8] O.I. El-Sabbagh, M.M. Baraka, S.M. Ibrahim, C. Pannecouque, G. Andrei, R. Snoeck, J. Balzarini, A.A. Rashad, et al., Synthesis and antiviral activity of new pyrazole and thiazole derivatives, *Eur. J. Med. Chem.* 44 (9) (2009) 3746–3753.
- [9] A. Özdemir, M. Altuntop, Z. Kaplançıklı, Ö. Can, Ü. Demir Özkay, G. Turan-Zitouni, Synthesis and evaluation of new 1,5-diaryl-3-[4-(methyl-sulfonyl)phenyl]-4,5-dihydro-1H-pyrazole derivatives as potential antidepressant agents, *Molecules* 20 (2) (2015) 2668–2684.
- [10] B.P. Bandgar, S.S. Gawande, R.G. Bodade, N.M. Gawande, C.N. Khobragade, Synthesis and biological evaluation of a novel series of pyrazole chalcones as anti-inflammatory, antioxidant and antimicrobial agents, *Bioorg. Med. Chem.* 17 (24) (Dec. 2009) 8168–8173.
- [11] R. Pérez-Fernández, P. Goya, J. Elguero, A review of recent progress (2002–2012) on the biological activities of pyrazoles, *Arkivoc* 2014 (2) (2013) 233.
- [12] N.K. Terrett, A.S. Bell, D. Brown, P. Ellis, Sildenafil (VIAGRAM), a potent and selective inhibitor of type 5 cGMP phosphodiesterase with utility for the treatment of male erectile dysfunction, *Bioorg. Med. Chem. Lett.* 6 (15) (1996) 1819–1824.
- [13] T.D. Penning, et al., Synthesis and biological evaluation of the 1,5-diarylpyrazole class of cyclooxygenase-2 inhibitors: identification of 4-[5-(4-methylphenyl)-3-(trifluoromethyl)-1H-pyrazol-1-yl]benzenesulfonamide (SC-58635, celecoxib), *J. Med. Chem.* 40 (9) (1997) 1347–1365.
- [14] D. Cluck, P. Lewis, B. Stayer, J. Spivey, J. Moorman, Ceftolozane-tazobactam: a new-generation cephalosporin, *Am. J. Heal. Pharm.* 72 (24) (Dec. 2015) 2135–2146.
- [15] "Small-Molecule Drug Discovery Suite | Schrödinger".
- [16] A. Cetin, A. Korkmaz, I. Bildirici, A novel oligo-pyrazole-based thin film: synthesis, characterization, optical and morphological properties, *Colloid Polym. Sci.* 296 (7) (2018) 1249–1257.
- [17] N. Menges, İ. Bildirici, Synthesis and evaluation of aromaticity and tautomerization of pyrazolepyridazin(on)es, *J. Chem. Sci.* 129 (6) (2017) 741–752.
- [18] R. Kasimoğulları, B. Seçkin Arslan, Synthesis and characterization of some pyrazole derivatives of 1,5-diphenyl-1H-pyrazole-3,4-dicarboxylic acid, *J. Heterocycl. Chem.* 47 (5) (2010) 1040–1048.
- [19] S. Ökten, M. Ekiz, U.M. Koçyiğit, A. Tutar, İ. Çelik, M. Akkurt, M. Gökalp, P. Taslimi, İ. Gulçin, et al., Synthesis, characterization, crystal structures, theoretical calculations and biological evaluations of novel substituted tacrine derivatives as cholinesterase and carbonic anhydrase enzymes inhibitors, *J. Mol. Struct.* 1175 (2019) 906–915.
- [20] F. Turkan, A. Cetin, P. Taslimi, İ. Gulçin, Some pyrazoles derivatives: potent carbonic anhydrase,  $\alpha$ -glycosidase, and cholinesterase enzymes inhibitors, *Arch. Pharm. (Weinheim)* 351 (10) (2018) 1800200.
- [21] B. Yiğit, M. Yiğit, D. Barut Celepci, Y. Gök, A. Aktaş, M. Aygün, P. Taslimi, İ. Gulçin, et al., Novel benzyl substituted imidazolium, tetrahydropyrimidinium and tetrahydrodiazepinium salts: potent carbonic anhydrase and acetylcholinesterase inhibitors, *ChemistrySelect* 3 (27) (2018) 7976–7982.
- [22] İ. Gulçin, P. Taslimi, Sulfonamide inhibitors: a patent review 2013-present, *Exp. Opin. Ther. Pat.* 28 (7) (2018) 541–549.
- [23] F. Türker, D. Barut Celepci, A. Aktaş, P. Taslimi, Y. Gök, M. Aygün, İ. Gulçin, et al., Meta-cyanobenzyl substituted benzimidazolium salts: synthesis, characterization, crystal structure and carbonic anhydrase,  $\alpha$ -glycosidase, butyrylcholinesterase, and acetylcholinesterase inhibitory properties, *Arch. Pharm. (Weinheim)* 351 (7) (2018) 1800029.
- [24] A. Behçet, T. Çağlılar, D. Barut Celepci, A. Aktaş, P. Taslimi, Y. Gök, M. Aygün, R. Kaya, İ. Gulçin, Synthesis, characterization, and crystal structure of 2-(4-hydroxyphenyl)ethyl and 2-(4-nitrophenyl)ethyl substituted benzimidazole bromide salts: their inhibitory properties against carbonic anhydrase and acetylcholinesterase, *J. Mol. Struct.* 1170 (2018) 160–169.
- [25] F. Erdemir, D. Barut Celepci, A. Aktaş, P. Taslimi, Y. Gök, H. Karabiyik, İ. Gulçin, et al., 2-Hydroxyethyl substituted NHC precursors: synthesis, characterization, crystal structure and carbonic anhydrase,  $\alpha$ -glycosidase, butyrylcholinesterase, and acetylcholinesterase inhibitory properties, *J. Mol. Struct.* 1155 (2018) 797–806.
- [26] P. Taslimi, İ. Gulçin, Antioxidant and anticholinergic properties of olivetol, *J. Food Biochem.* 42 (3) (2018) e12516.
- [27] A. Akıncioğlu, A. Kocaman, H. Akıncioğlu, R.E. Salmas, S. Durdağı, İ. Gülçin, C.T. Supuran, S. Göksu, et al., The synthesis of novel sulfamides derived from  $\beta$ -benzylphenethylamines as acetylcholinesterase, butyrylcholinesterase and carbonic anhydrase enzymes inhibitors, *Bioorg. Chem.* 74 (2017) 238–250.
- [28] İ. Gülçin, A. Scozzafava, C.T. Supuran, H. Akıncioğlu, Z. Koksall, F. Turkan, S. Alwasel, et al., The effect of caffeic acid phenethyl ester (CAPE) on metabolic enzymes including acetylcholinesterase, butyrylcholinesterase, glutathione S-transferase, lactoperoxidase, and carbonic anhydrase isoenzymes I, II, IX, and XII, *J. Enzyme Inhib. Med. Chem.* 31 (6) (2016) 1095–1101.
- [29] H. Akıncioğlu, İ. Gülçin, S.H. Alwasel, Investigation of the inhibitory effect of hemic acid on acetylcholinesterase and butyrylcholinesterase enzymes, *Fresen. Environ. Bull.* 26 (6) (2017) 3733–3739.
- [30] İ. Gülçin, A. Scozzafava, C.T. Supuran, Z. Koksall, F. Turkan, S. Çetinkaya, Z. Bingöl, Z. Huyut, S.H. Alwasel, et al., Rosmarinic acid inhibits some metabolic enzymes including glutathione S-transferase, lactoperoxidase, acetylcholinesterase, butyrylcholinesterase, and carbonic anhydrase isoenzymes, *J. Enzyme Inhib. Med. Chem.* 31 (6) (2016) 1698–1702.
- [31] J.A. Verpoorte, S. Mehta, J.T. Edsall, Esterase activities of human carbonic anhydrases B and C, *J. Biol. Chem.* 242 (18) (1967) 4221–4229.
- [32] P. Taslimi, C. Çağlayan, F. Farzaliyev, O. Nابیev, A. Sujayev, F. Turkan, R. Kaya, İ. Gulçin, et al., Synthesis and discovery of potent carbonic anhydrase, acetylcholinesterase, butyrylcholinesterase, and  $\alpha$ -glycosidase enzymes inhibitors: the novel N,N'-bis-cyanomethylamine and alkoxyethylamine derivatives, *J. Biochem. Mol. Toxicol.* 32 (4) (2018) e22042.
- [33] P. Taslimi, E. Sujayev, F. Turkan, E. Garibov, Z. Huyut, F. Farzaliyev, S. Mamedova, İ. Gulçin, et al., Synthesis and investigation of the conversion reactions of pyrimidine-thiones with nucleophilic reagent and evaluation of their acetylcholinesterase, carbonic anhydrase inhibition, and antioxidant activities, *J. Biochem. Mol. Toxicol.* 32 (2) (2018) e22019.
- [34] P. Taslimi, A. Sujayev, S. Mamedova, P. Kalin, İ. Gulçin, N. Sadeghian, S. Beydemir, Ö. Küfrevioğlu, S.H. Alwasel, V. Farzaliyev, S. Mamedov, et al., Synthesis and bioactivity of several new hetaryl sulfonamides, *J. Enzyme Inhib. Med. Chem.* 32 (1) (2017) 137–145.
- [35] F. Topal, İ. Gulçin, A. Dastan, M. Guney, Novel eugenol derivatives: potent acetylcholinesterase and carbonic anhydrase inhibitors, *Int. J. Biol. Macromol.* 94 (2017) 845–851.
- [36] G.L. Ellman, K.D. Courtney, V. Andres, R.M. Featherstone, A new and rapid colorimetric determination of acetylcholinesterase activity, *Biochem. Pharmacol.* 7 (2) (1961) 88–95.
- [37] F. Turkan, Z. Huyut, P. Taslimi, İ. Gulçin, The effects of some antibiotics from cephalosporin groups on the acetylcholinesterase and butyrylcholinesterase enzymes activities in different tissues of rats, *Arch. Physiol. Biochem.* (2018) 1–7.
- [38] F. Turkan, Z. Huyut, P. Taslimi, İ. Gulçin, The *in vivo* effects of cefazolin, cefuroxime, and cefoperazone on the carbonic anhydrase in different rat tissues, *J. Biochem. Mol. Toxicol.* 32 (3) (2018) e22041.
- [39] E. Garibov, P. Taslimi, A. Sujayev, Z. Bingöl, S. Çetinkaya, İ. Gulçin, S. Beydemir, V. Farzaliyev, S.H. Alwasel, C.T. Supuran, et al., Synthesis of 4,5-disubstituted-2-thioxo-1,2,3,4-tetrahydropyrimidines and investigation of their acetylcholinesterase, butyrylcholinesterase, carbonic anhydrase I/II inhibitory and antioxidant activities, *J. Enzyme Inhib. Med. Chem.* 31 (sup3) (2016) 1–9.
- [40] H.E. Aslan, Y. Demir, M.S. Özasan, F. Turkan, Ş. Beydemir, Ö.I. Küfrevioğlu, The behavior of some chalcones on acetylcholinesterase and carbonic anhydrase activity, *Drug Chem. Toxicol.* (2018) 1–7.
- [41] B. Turan, K. Sendil, E. Sengul, M.S. Gultekin, P. Taslimi, İ. Gulçin, C.T. Supuran, et al., The synthesis of some  $\beta$ -lactams and investigation of their metal-chelating activity, carbonic anhydrase and acetylcholinesterase inhibition profiles, *J. Enzyme Inhib. Med. Chem.* 31 (sup1) (2016) 79–88.
- [42] F. Turkan, Z. Huyut, Y. Demir, F. Ertaş, Ş. Beydemir, The effects of some cephalosporins on acetylcholinesterase and glutathione S-transferase: an *in vivo* and *in vitro* study, *Arch. Physiol. Biochem.* (2018) 1–9.
- [43] P. Taslimi, S. Osmanova, C. Çağlayan, F. Turkan, S. Sardarova, F. Farzaliyev, E. Sujayev, N. Sadeghian, İ. Gulçin, et al., Novel amides of 1,1-bis-(carboxymethylthio)-1-arylethanes: synthesis, characterization, acetylcholinesterase, butyrylcholinesterase, and carbonic anhydrase inhibitory properties, *J. Biochem. Mol. Toxicol.* 32 (9) (2018) e22191.
- [44] N. Öztaşkın, Y. Çetinkaya, P. Taslimi, S. Göksu, İ. Gülçin, Antioxidant and acetylcholinesterase inhibition properties of novel bromophenol derivatives, *Bioorg. Chem.* 60 (2015) 49–57.
- [45] A. Akıncioğlu, H. Akıncioğlu, İ. Gülçin, S. Durdağı, C.T. Supuran, S. Göksu, Discovery of potent carbonic anhydrase and acetylcholinesterase inhibitors: novel sulfamoylcarbamates and sulfamides derived from acetophenones, *Bioorg. Med. Chem.* 23 (13) (2015) 3592–3602.
- [46] İ. Gulçin, P. Taslimi, A. Aygün, N. Sadeghian, E. Bastem, Ö. Küfrevioğlu, F. Turkan, F. Şen, et al., Antidiabetic and antiparasitic potentials: Inhibition effects of some

- natural antioxidant compounds on  $\alpha$ -glycosidase,  $\alpha$ -amylase and human glutathione S-transferase enzymes, *Int. J. Biol. Macromol.* 119 (2018) 741–746.
- [47] G. Gondolova, P. Taslimi, A. Medjidov, F. Farzaliyev, A. Sujayev, M. Huseuinova, O. Şahin, B. Yalçın, F. Turkan, İ. Gülçin, et al., Synthesis, crystal structure and biological evaluation of spectroscopic characterization of Ni(II) and Co(II) complexes with N -salicyloyl- N'-maleoil-hydrazine as anticholinergic and antidiabetic agents, *J. Biochem. Mol. Toxicol.* 32 (9) (2018) e22197.
- [48] S. Göksu, A. Naderi, Y. Akbaba, P. Kalın, A. Akıncıoğlu, İ. Gülçin, S. Durdağı, R.E. Salmas, et al., Carbonic anhydrase inhibitory properties of novel benzylsulfamides using molecular modeling and experimental studies, *Bioorg. Chem.* 56 (2014) 75–82.
- [49] C. Yamali, H.I. Gul, A. Ece, P. Taslimi, I. Gulcin, Synthesis, molecular modeling, and biological evaluation of 4-[5-aryl-3-(thiophen-2-yl)-4,5-dihydro-1 H -pyrazol-1-yl] benzenesulfonamides toward acetylcholinesterase, carbonic anhydrase I and II enzymes, *Chem. Biol. Drug Des.* 91 (4) (2018) 854–866.
- [50] M. Ahmed, M.A. Qadir, A. Hameed, M.N. Arshad, A.M. Asiri, M. Muddassar, Sulfonamides containing curcumin scaffold: synthesis, characterization, carbonic anhydrase inhibition, and molecular docking studies, *Bioorg. Chem.* 76 (2018) 218–227.
- [51] M.F. Abo-Ashour, W.M. Eldehna, A. Nocentini, H.S. Ibrahim, S. Bua, S.M. Abou-Seri, C.T. Supuran, et al., Novel hydrazido benzenesulfonamides-isatin conjugates: Synthesis, carbonic anhydrase inhibitory activity, and molecular modeling studies, *Eur. J. Med. Chem.* 157 (2018) 28–36.
- [52] T. Fattah, A. Saeed, P.A. Channar, F.A. Larik, M. Hassan, H. Raza, Q. Abbas, S.Y. Seo, et al., Synthesis and Molecular Docking Studies of (E)-4-(substituted-benzylideneamino)-2H-chromen-2-one derivatives: entry to new carbonic anhydrase class of inhibitors, *Drug Res. (Stuttg)* 68 (7) (2018) 378–386.
- [53] S.M.D. Rizvi, S. Shaikh, D. Naaz, S. Shakil, A. Ahmad, M. Haneef, A.M. Abuzenadah, et al., Kinetics and molecular docking study of an anti-diabetic drug glimepiride as acetylcholinesterase inhibitor: implication for alzheimer's disease-diabetes dual therapy, *Neurochem. Res.* 41 (6) (2016) 1475–1482.
- [54] S.-H. Lu, J.W. Wu, H.L. Liu, J.H. Zhao, K.T. Liu, C.K. Chuang, H.Y. Lin, W.B. Tsai, Y. Ho, et al., The discovery of potential acetylcholinesterase inhibitors: a combination of pharmacophore modeling, virtual screening, and molecular docking studies, *J. Biomed. Sci.* 18 (1) (2011) 8.