



Design, synthesis and molecular modeling studies of new series of *s*-triazine derivatives as antimicrobial agents against multi-drug resistant clinical isolates

Nesreen Saied Haiba^a, Hosam H. Khalil^{b,*}, Mohamed Abdel Moniem^b, Marwa H. El-Wakil^c, Adnan A. Bekhit^{c,d}, Sherine Nabil Khattab^{b,d,*}

^a Department of Physics and Chemistry, Faculty of Education, Alexandria University, Alexandria, Egypt

^b Department of Chemistry, Faculty of Science, Alexandria University, Alexandria 21321, Egypt

^c Department of Pharmaceutical Chemistry, Faculty of Pharmacy, Alexandria University, Alexandria 21521, Egypt

^d Cancer Nanotechnology Research Laboratory (CNRL), Faculty of Pharmacy, Alexandria University, Alexandria 21521, Egypt

ARTICLE INFO

Keywords:

s-Triazine
Antimicrobial activity
MDR clinical isolates
SAR
In silico calculations
Molecular docking

ABSTRACT

Three novel series of *s*-triazine derivatives, including thirty-five new compounds **2a-d**, **3a-3p**, **4b-d**, **5b-d**, **6d-6d**, and **7a-7f** were synthesized comprising a diversity of substituents based on the structure of Astrazeneca arylaminotriazine DNA gyrase B inhibitor. The antimicrobial activity was determined for all compounds against *Staphylococcus aureus*, *Escherichia coli* and *Candida albicans* using the two-fold serial dilution technique and against reference standards Ampicillin for the antibacterial screening and Clotrimazole regarding the antifungal evaluation. The tested compounds showed strong to moderate antibacterial inhibitory action and weak antifungal activity. Compounds **3j** and **6b** were the most potent antibacterial agents against the tested strains and multi-drug resistant (MDR) clinical isolates of *Klebsiella pneumoniae* and methicillin resistant *Staphylococcus aureus* (MRSA) with minimal toxicity in comparison to the reference drugs. *In silico* molecular properties calculations and molecular docking study for **3j** and **6b** revealed that both compounds could be considered as promising antibacterial DNA gyrase B inhibitors.

1. Introduction

Worldwide, microbial infectious diseases caused by pathogenic microorganisms pose danger to the community [1,2]. The misuse of antibiotics had a profound impact on the development of multi-drug microbial resistance against the chemotherapeutic agents available [2]. The resistance developed by the microbial strains urged medicinal chemists to search for both novel antimicrobial agents and antibacterial targets to overcome this issue [3–5]. The *s*-triazine ring has been widely studied for its synthetic availability and diverse broad biological activities such as antiprotozoal [6], anticancer [7], antimalarial [3], antiviral [8], antifungal [9,10], antimicrobial [11], antileishmanial activity [12], MAO inhibitors [13] and drug delivery systems [14]. Cyanuric chloride is the starting reagent for synthesis of various *s*-triazine derivatives. It is a commercially available reagent that has attracted attention to its temperature-controlled substitution of the chlorine atoms by various nucleophiles [3,15–19].

Previous reports demonstrated that *s*-triazines linked to lipophilic

parts had remarkable antibacterial and antifungal activities [20–23]. Similarly, hydrazones are important as antimicrobial [24–27], antiviral [25,28], antioxidant [29], and antitumor [30,31] agents. Recently we have demonstrated a series of dimeric *s*-triazine hydrazone derivatives [32] and *s*-triazines containing *p*-aminobenzoic acid moiety [33] as antibacterial and antifungal agents (I and II, Fig. 1). Many investigations presented DNA gyrase enzyme as a pivotal antibacterial target. The enzyme is essential in overseeing the topological state of DNA during transcription and replication processes [34]. DNA gyrase inhibition results in the disruption of DNA synthesis and leads eventually to bacterial cell death. Astrazeneca arylaminotriazine (III, Fig. 1) has been identified as a promising DNA gyrase inhibitor with profound antibacterial activity [35].

2. Rationale and design

Taking into consideration the abovementioned data and in continuation to our dedicated program towards synthesis of new *s*-triazines

* Corresponding authors at: Faculty of Science, Alexandria University, Alexandria 21321, Ibrahimia, P.O. box 426, Egypt.

E-mail addresses: chemhosam1@yahoo.com (H.H. Khalil), sherinekhattab@alexu.edu.eg (S.N. Khattab).

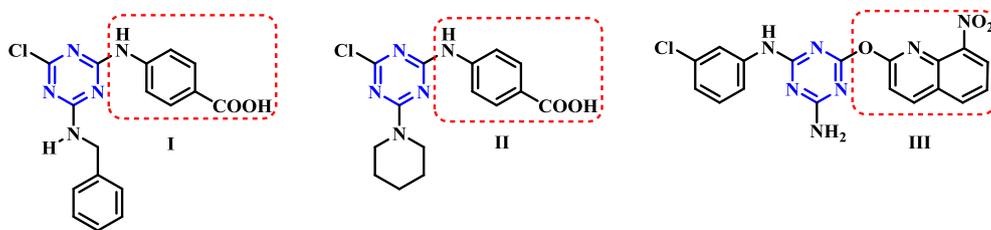


Fig. 1. Structures of reported antibacterial *s*-triazines (compounds I and II) and gyrase inhibitor Astrazeneca arylaminotriazine III.

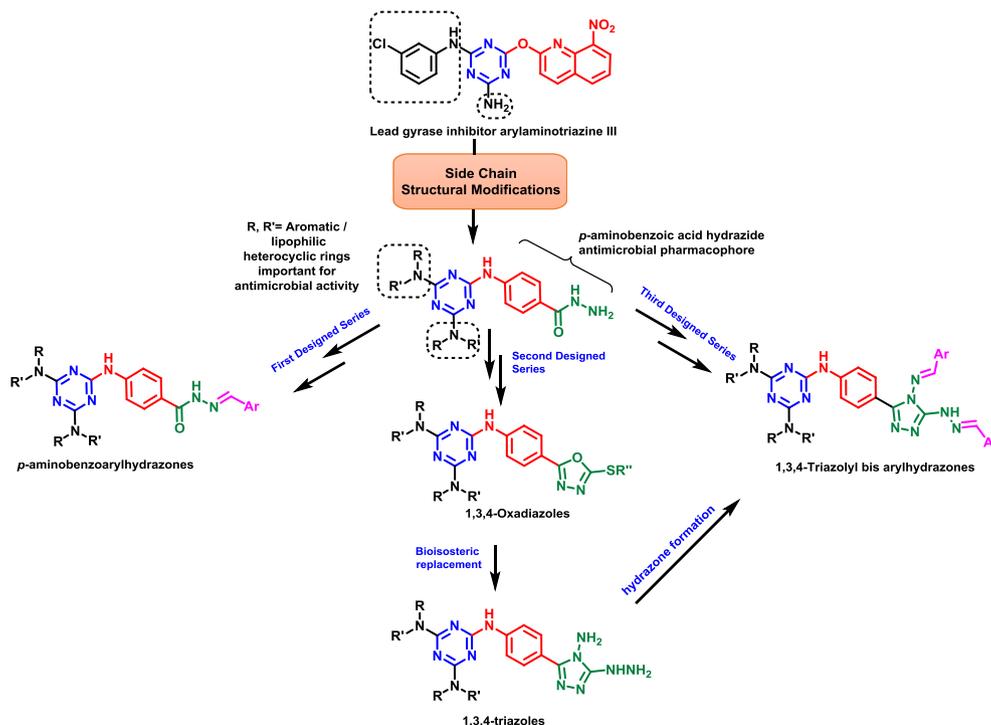


Fig. 2. The different design strategies for obtaining the target compounds based on the lead gyrase inhibitor arylaminotriazine III.

[32,33], we aimed at designing new *s*-triazine derivatives based on the important structural features of DNA gyrase B inhibitor arylaminotriazine III. The suggested candidates are designed to possess hydrophobic phenyl rings and/ or lipophilic amine moieties linked to the triazine core at the 4- and 6-positions. Such moieties are known to be essential for antimicrobial activity [33]. The triazine 2-position is planned to incorporate 2-aminobenzohydrazide linker which, first, was designed to be in the form of aryl hydrazones that are known to be found in different antimicrobial agents [36]. The second strategy involved cyclization of the benzohydrazide into 1,3,4-oxadiazole ring, followed by isosteric replacement into amino hydrazinyl-1,3,4-triazole. These heterocyclic rings are found in many antimicrobial derivatives [35,37]. Finally, design of different triazolyl bis aryl hydrazones was suggested to investigate their impact on the anticipated antimicrobial activity (Fig. 2). Synthesis of the target compounds were carried out according to outlined synthetic schemes and the derivatives were evaluated for their antibacterial and antifungal activities.

3. Results and discussion

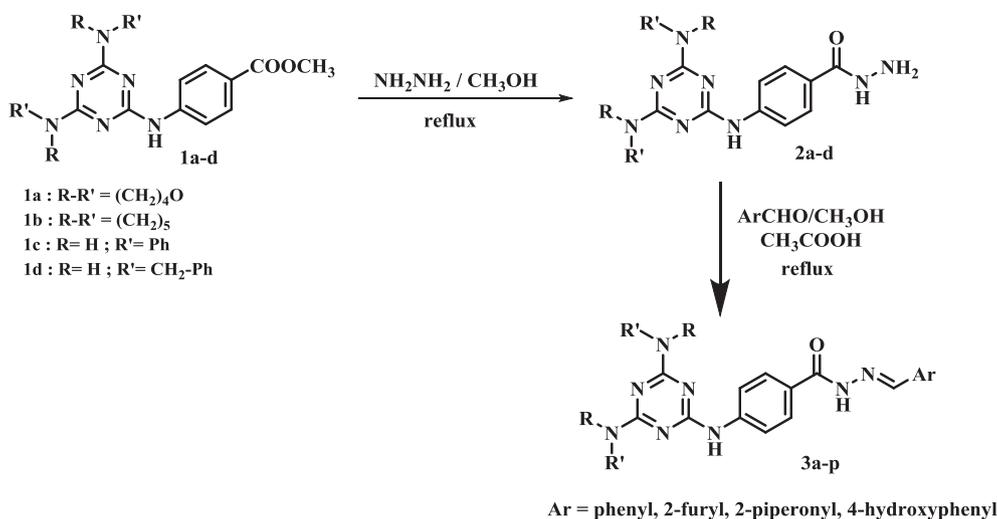
3.1. Chemistry

Synthesis of the new derivatives are outlined in Schemes 1, 2 and 3. Structures of the newly synthesized compounds were determined by spectroscopic methods (IR, ^1H NMR and ^{13}C NMR) and elemental analyses. Scheme 1 describes the reaction between 2-aminobenzamido methyl esters **1a-d** [33] and hydrazine hydrate in methanol to yield the

corresponding hydrazides **2a-d**. IR spectra of **2a-d** revealed presence of absorption bands due to NHNH_2 at $3427\text{--}3276\text{ cm}^{-1}$. While, the absorption bands of carbonyl group appeared at $1620\text{--}1604\text{ cm}^{-1}$. ^1H NMR spectra of **2a-d** showed two D_2O exchangeable singlets at δ 9.35–9.63 and 4.40–4.51 ppm corresponding to NHNH_2 protons, emphasizing the complete hydrazinolysis of the starting ester. Subsequent condensation of **2a-d** with different aromatic aldehydes in a mixture of methanol containing catalytic amount of glacial acetic acid afforded the corresponding hydrazones **3a-p**.

As observed from ^1H NMR spectra of **3a-p**, some hydrazones appeared as *syn* and *anti* isomers (Fig. 3). Taking ^1H NMR spectrum of **3c** as example, it showed that the ratio between the two isomers is nearly 5:1. The two isomers differ in the chemical shift and integration of N–H and methine ($\text{CH}=\text{N}$) proton. The eight methylene protons of morpholine moieties appeared as multiplet at δ 3.60–3.67 ppm, while the methylene protons of piperonal appeared as singlet at δ 6.06 ppm. Furthermore, the aromatic protons were observed as doublet at δ 6.95, singlet at δ 7.27 and multiplet at δ 7.12–7.13 and 7.80–7.88 ppm. The methine proton appeared as two singlets at δ 8.32 and 8.34 ppm, while the hydrazono NH proton appeared as two D_2O exchangeable singlets at δ 9.45 and 9.61 ppm. In addition, two D_2O exchangeable singlets were observed at δ 11.59 and 11.62 ppm corresponding to aminobenzoyl NH proton.

Since formation of **3c** involves two geometric isomers as shown in Fig. 3. Therefore, it was of interest to model the compounds using molecular mechanics MM2 calculations. In addition, quantum chemical calculations were carried out with the GAUSSIAN 98 suite. Geometry



Scheme 1. Synthetic pathway for the preparation of the target compounds 2a-d and 3a-3p.

optimizations were carried out using the DFT level (B3LYP/6-31G) theory to assess the relative stability of the diastereomeric species. The calculated relative energies of 3c *syn* and *anti* conformers were 56.7272 kcal/mol and 56.7468 kcal/mol, respectively. The computed energies indicated that the *syn* isomer is 0.0196 kcal/mol more stable than the *anti* counterpart (Fig. 4).

Scheme 2 outlines the cyclization of the hydrazides 2b-d to the corresponding oxadiazole thiol derivatives 4b-d using carbon disulfide and aqueous KOH. IR spectra of 4b-d showed absorption bands due to N-H and SH at 3398–3288 cm⁻¹ and 2924–2852 cm⁻¹ respectively. ¹H NMR spectra of 4b-d lacked the signals due to NHH₂ and instead revealed presence of D₂O exchangeable singlet at δ 2.94–3.12 ppm corresponding to SH proton. Signals attributed to other aliphatic and aromatic protons appeared at their expected chemical shifts. Further methylation of the oxadiazole thiols 4b-d with methyl iodide in alcoholic NaOH solution efficiently afforded the corresponding methylsulfonyloxadiazoles 5b-d. IR spectra of 5b-d lacked the SH absorption band but showed only absorption bands attributed to N-H at 3402–3379 cm⁻¹. ¹H NMR spectra 5b-d lacked the signals due to SH protons while displayed signals at δ 2.79–3.33 ppm corresponding to SCH₃ protons. Heating 5b-d with excess hydrazine hydrate in methanol afforded the corresponding aminohydrazinyltriazole derivatives 6b-d. IR spectra of 6b-d showed absorption bands for NH at 3412–3340 cm⁻¹. ¹H NMR spectra of 6b-d lacked the signals of the methyl protons and instead displayed D₂O exchangeable singlets at δ

4.39–4.46 and 6.98–7.82 ppm corresponding to NH₂ protons.

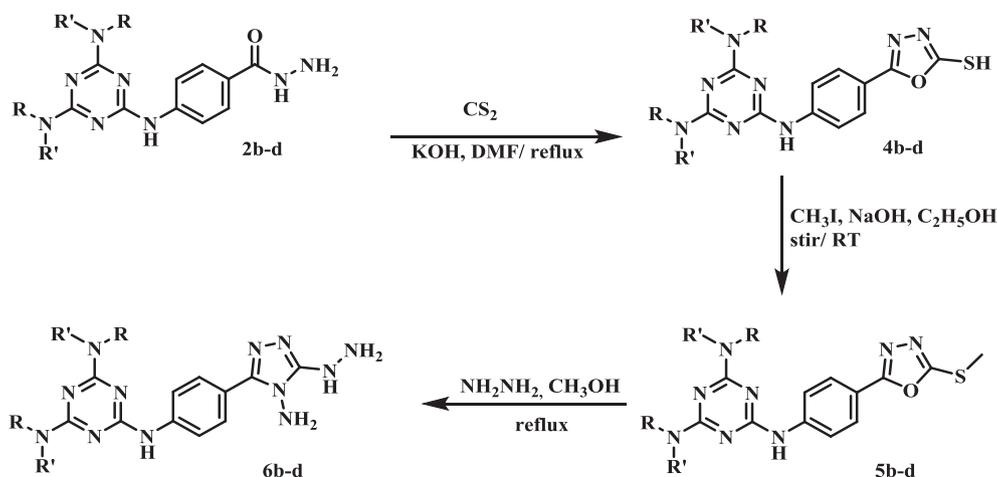
The target hydrazones 7a-f were obtained by reaction of the amino hydrazinyl triazoles 6b-d with different aromatic aldehydes in a refluxing mixture of methanol containing catalytic amount of glacial acetic acid (Scheme 3). IR spectra of 7a-f showed absorption bands due to NH at 3428–3395 cm⁻¹. ¹H NMR of 7a-f lacked the signals attributed to NH₂ and in turn revealed singlets at δ 8.27–8.84 ppm corresponding to methine protons.

3.2. Biological evaluation

3.2.1. In vitro antimicrobial activity

All synthesized compounds were screened for their *in vitro* antimicrobial activity against *Staphylococcus aureus* (*S. aureus* ATCC 19433) as example of Gram-positive bacteria, *Escherichia coli* (*E. coli* ATCC 25922) as example of Gram-negative bacteria and *Candida albicans* (*C. albicans*) as yeast-like fungus. Individual minimum inhibitory concentration (MIC, µg/mL) values of the screened compounds against the test microbes are listed in Table 1 along with MIC values of reference Ampicillin (for bacteria) and Clotrimazole (for yeast-like fungus).

In general, all compounds exhibited promising antibacterial activity against the bacterial strains tested and weaker antifungal activity in comparison to the reference drugs. Compounds 3j, 5b, 6b and 6c showed two-fold higher antibacterial activity against *E. coli* than ampicillin with MIC values 12.5 µg/mL. While compounds 3e, 4b, 4c, and



Scheme 2. Synthetic pathway for the preparation of the target compounds 4b-d, 5b-d and 6d-6d.

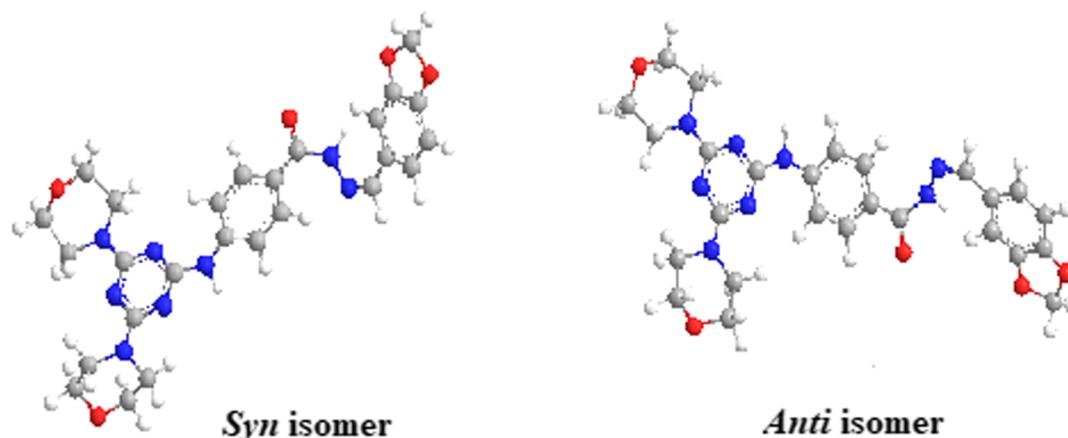


Fig. 4. The 3D structure of the *syn* and *anti* isomers of **3c**.

Table 1

Antibacterial and antifungal screening in terms of MIC ($\mu\text{g/mL}$) for the tested compounds in reference to Ampicillin and Clotrimazole, respectively.

Compound	R	R'	Ar	MIC ($\mu\text{g/mL}$)		
				<i>E. coli</i>	<i>S. aureus</i>	<i>C. albicans</i>
2a	(CH ₂) ₄ O			50	100	100
2b	(CH ₂) ₅			100	100	50
2c	H	Ph	–	50	50	50
2d	H	CH ₂ -Ph	–	100	100	50
3a	(CH ₂) ₄ O		Ph	100	100	> 200
3b	(CH ₂) ₄ O		2-furyl	100	100	100
3c	(CH ₂) ₄ O		2-piperonyl	> 200	> 200	50
3d	(CH ₂) ₄ O		4-OH-Ph	> 200	> 200	50
3e	(CH ₂) ₅		Ph	25	100	> 200
3f	(CH ₂) ₅		2-furyl	50	100	50
3g	(CH ₂) ₅		2-piperonyl	100	12.5	50
3h	(CH ₂) ₅		4-OH-Ph	100	100	50
3i	H	Ph	Ph	100	100	> 200
3j	H	Ph	2-furyl	12.5	6.25	50
3k	H	Ph	2-piperonyl	> 200	25	100
3l	H	Ph	4-OH-Ph	100	100	50
3m	H	CH ₂ -Ph	Ph	100	100	100
3n	H	CH ₂ -Ph	2-furyl	50	> 200	100
3o	H	CH ₂ -Ph	2-piperonyl	> 200	> 200	25
3p	H	CH ₂ -Ph	4-OH-Ph	50	50	100
4b	(CH ₂) ₅		–	25	6.25	100
4c	H	Ph	–	25	12.5	100
4d	H	CH ₂ -Ph	–	50	25	100
5b	(CH ₂) ₅		–	12.5	12.5	> 200
5c	H	Ph	–	50	100	100
5d	H	CH ₂ -Ph	–	50	100	> 200
6b	(CH ₂) ₅		–	12.5	6.25	> 200
6c	H	Ph	–	12.5	25	> 200
6d	H	CH ₂ -Ph	–	50	25	> 200
7a	(CH ₂) ₅		2-piperonyl	50	25	> 200
7b	H	Ph	2-furyl	100	100	> 200
7c	H	CH ₂ -Ph	2-piperonyl	25	100	> 200
7d	(CH ₂) ₅		2-furyl	50	100	> 200
7e	H	Ph	2-piperonyl	50	100	> 200
7f	H	CH ₂ -Ph	2-furyl	50	100	> 200
Ampicillin				25	12.5	–
Clotrimazole				–	–	12.5

The MIC values of the most active compounds are written in Bold.

3.5. Molecular docking study

It was of interest to elucidate the mechanism by which the two active compounds **3j** and **6b** persuaded their antibacterial activities. Since the compounds were designed based on the gyrase B inhibitor arylaminotriazine **III**, the co-crystallized structure of DNA gyrase B of *E. coli* (PDB code: **4DUH**) was downloaded from the protein data bank [43] coupled with its ligand inhibitor and used as reference for the

essential docking interactions. Docking was performed using Molecular Operating Environment (MOE) software version 2016.0802, Chemical Computing Group, Montreal, Canada [44], to evaluate the 2D and 3D binding modes of the selected derivatives in the enzyme active site. Enzyme preparation and docking protocol were carried out according to a previously reported method [45]. The 2D ligand interactions of the co-crystallized inhibitor (Fig. 5a) showed ionic interactions with amino acid residues Arg 76 and Arg 136. It also displayed hydrogen bond interactions with Gly 101 and Arg 136, in addition to other hydrophobic interactions. For comparison, docking of the gyrase B inhibitor arylaminotriazine **III** displayed free energy of binding ΔG -8.4542 kcal/mol and showed hydrogen bonding interactions with amino acid residues Gly 101 and His 99 in addition to π -H interaction with Asn 46 (Fig. 5b). Docking of **3j** and **6b** displayed ΔG -9.3159 and -9.6475 kcal/mol, respectively, better than arylaminotriazine **III** which suggests that the test compounds are embedded deeply in the enzyme active site. Compound **3j** exhibited hydrogen bond donor interaction between N at 2-position of triazine ring and Ala 100, hydrogen bond acceptor interaction between oxygen of the benzohydrazone and His 99 in addition to π -H interaction of N-phenyl ring with Lys 103 (Fig. 6). Whereas **6b** displayed mainly hydrogen bond acceptor interactions mainly between the two triazole N and Arg 136, besides π -H interaction between the phenyl ring and Pro 79 (Fig. 7). As a comparative study to relate binding mode to potency, docking of the moderately active derivative **7c** was considered. Although the estimated ΔG is -11.9675 kcal/mol, the 2D ligand interaction revealed only one hydrogen bond acceptor between the imine N and amino acid residue Lys 103 in addition to π -H interaction of the triazole ring also with Lys 103 (Fig. 8). It is worth mentioning that upon investigating the 3D interaction of **7c** with the enzyme active site it was obvious that the s-triazine core completely extended outside the binding pocket which would greatly be the cause of its weak interactions and decreased potency.

4. Conclusion

The present study describes the design and synthesis of thirty-five new s-triazine derivatives as promising antimicrobial agents based on DNA gyrase inhibitor Astrazeneca arylaminotriazine **III**. Among the synthesized derivatives, nine compounds (**3e**, **3g**, **3j**, **4b**, **4c**, **5b**, **6b**, **6c**, **7c**) showed promising antibacterial activities against tested *E. coli* and *S. aureus* bacterial strains more potent than Ampicillin as reference. Further antibacterial screening of these active compounds against MDR clinical isolates of *K. pneumoniae* and *MRSA1* revealed compounds **3j** and **6b** as potent antibacterial agents with minimal toxicity. SAR study pointed out the importance of the s-triazine ring substituted with hydrophobic phenyl rings or lipophilic piperidine moieties to the

Table 2

Minimum inhibitory concentrations (MIC, $\mu\text{g/mL}$) of the most active compounds on MDR clinical isolates of *K. pneumoniae* and *MRSA1*.

MIC ($\mu\text{g/mL}$) for MDR clinical isolates		
Compound	<i>K. pneumoniae</i>	<i>MRSA1</i>
3e	50	–
3g	–	25
3j	12.5	3.125
4b	25	12.5
4c	50	12.5
5b	12.5	25
6b	12.5	3.125
6c	12.5	25
7c	50	–
Ampicillin	–	–

The MIC values of the most active compounds are written in Bold.

Table 3

CC_{50} values of the most active compounds against normal VERO cells and their selectivity indices.

Compound	$\text{CC}_{50}^{\text{a}}$	<i>K. pneumoniae</i>		<i>MRSA1</i>	
		MIC^{b}	SI^{c}	MIC	SI
3e	250	50	5	–	–
3g	250	–	–	25	10
3j	125	12.5	10	3.125	40
4b	125	25	5	12.5	10
4c	500	50	10	12.5	40
5b	250	12.5	20	25	10
6b	62.5	12.5	5	3.125	20
6c	125	12.5	10	25	5
7c	250	50	5	–	–

The MIC values of the most active compounds are written in Bold.

^a CC_{50} is the concentration at which 50% of cells survive.

^b MIC ($\mu\text{g/mL}$) is the minimum concentration that inhibits bacterial cell growth.

^c SI is the selectivity index regarding antimicrobial activity against *K. pneumoniae* and *MRSA1*; $\text{SI} = \log [\text{CC}_{50}/\text{MIC}]$. Positive value represents more selectivity against microorganism strains than Vero cells, and negative values indicate higher toxicity to Vero cells and low selectivity to the bacteria [39].

antibacterial potency. In addition, substitution with less steric hydrazone or amino hydrazinyl triazole moieties are essential for optimal antibacterial activities. *In silico* calculation of Mwt and $\text{cLog } P$ values emphasized their impact on the increased antibacterial activities of **3j** and **6b**. Furthermore, docking study of these two derivatives in order to discern the possible mechanism for their antibacterial potency revealed strong binding interactions in DNA gyrase active site. Thus, both **3j** and **6b** could be considered as potential leads for further optimization and development in future work.

5. Experimental section

5.1. Materials and methods

All reagents were purchased from commercial suppliers and the solvents used were of HPLC grade. Melting points were determined with a Mel-Temp apparatus and are uncorrected. Infrared (IR) spectra were recorded on a Perkin-Elmer 1600 series Fourier transform instrument using KBr cell and measured by cm^{-1} scale. Nuclear Magnetic resonance spectra (^1H NMR and ^{13}C NMR spectra) were recorded on 500 MHz JEOL and 400 MHz BRUKER spectrometers. Chemical shifts were reported in δ values (ppm) relative to trimethylsilane (TMS) as an internal standard and are referenced relative to residual solvent (e.g. CHCl_3 at δ H 7.26 ppm for CDCl_3 and DMSO at δ H 2.50 ppm for $\text{DMSO-}d_6$). Splitting patterns were designated as follows: s: singlet; br.s: broad singlet; d: doublet; m: multiplet and the coupling constants (J) in hertz. Elemental analyses were performed on Perkin-Elmer 2400 elemental analyzer, and the values found were within $\pm 0.3\%$ of the theoretical values. Follow-up of the reactions and checking the purity of the compounds were done by TLC on silica gel coated aluminum sheets (Type 60 GF254, Merck) and the spots were detected by exposure to UV-lamp at λ 254 nm for few seconds.

5.1.1. General procedure for the preparation of 4-[(4,6-disubstituted-1,3,5-triazin-2-yl)amino]benzohydrazide (**2a-d**)

To a solution of the appropriate ester **1a-d** (2 mmol) in methanol (20 mL), hydrazine hydrate (2 mL) was added. The reaction mixture was heated under reflux for 6 h. The precipitated hydrazide products were filtered off and washed with methanol and recrystallized from ethanol.

5.1.1.1. 4-[(4,6-Dimorpholino-1,3,5-triazin-2-yl)amino]benzohydrazide (**2a**). White powder. Yield: 0.75 g (94%); mp 218–220 °C. IR (KBr, ν_{max}

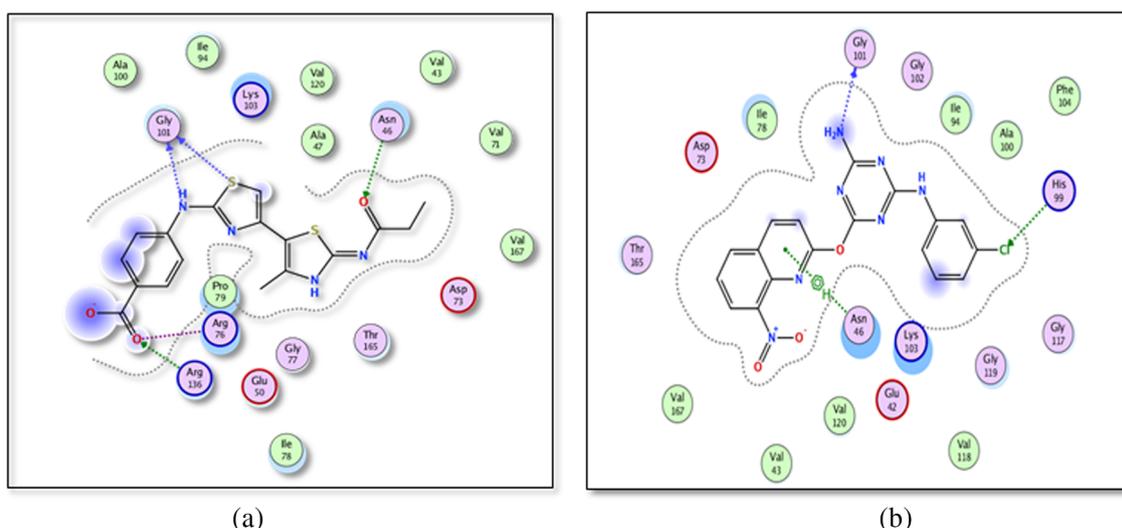


Fig. 5. The 2D binding mode of (a) the co-crystallized ligand; (b) arylaminotriazine **III** in the binding site of *E. coli* DNA gyrase B (PDB code: 4DUH). Hydrogen bond donor is indicated by blue dotted arrow and hydrogen bond acceptor is indicated by green dotted arrow.

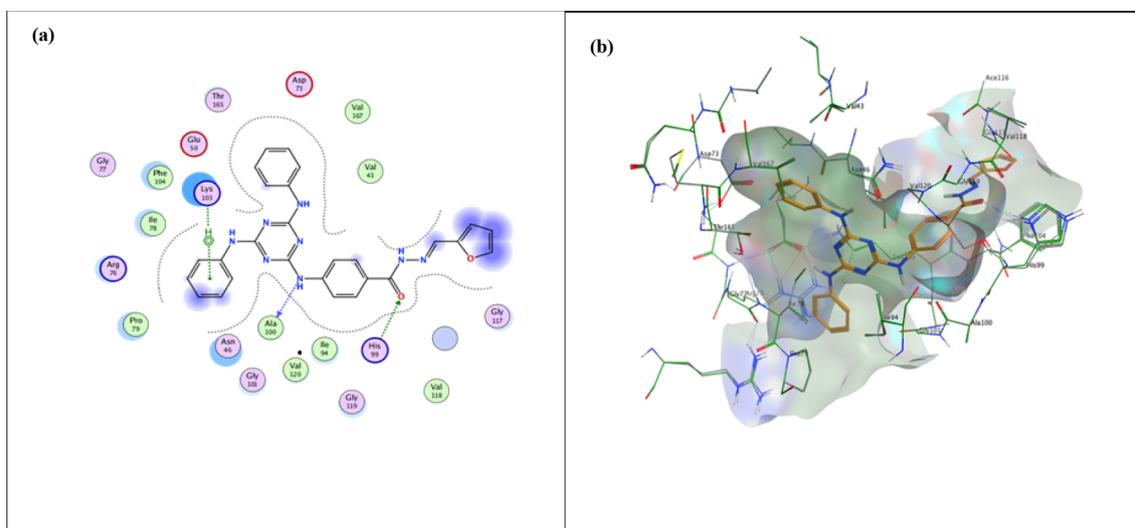


Fig. 6. (a) The 2D binding mode and (b) the 3D binding mode of **3j** in the active site of *E. coli* DNA gyrase B (PDB code: 4DUH).

cm^{-1}): 3410, 3311 (NH), 1608 (C=O, amide). ^1H NMR (500 MHz, $\text{DMSO}-d_6$): δ 3.60–3.74 (m, 16H, 8CH_2), 4.40 (br.s, 2H, NH_2 , D_2O exchangeable), 7.70–7.74 (m, 4H, Ar–H), 9.35 (s, 1H, NH, D_2O exchangeable.), 9.53 (m, 1H, NH, D_2O exchangeable). ^{13}C NMR (125 MHz, $\text{DMSO}-d_6$): δ 43.75 (4 C–N), 66.42 (4 C–O), 119.39, 119.62, 128.1 (2 C), 132.11, 148.86, 163.03, 164.43, 165.09. Anal. Calcd. for $\text{C}_{18}\text{H}_{24}\text{N}_8\text{O}_3$: C, 53.99; H, 6.04; N, 27.98. Found: C, 52.78; H, 5.93; N, 28.11.

5.1.1.2. 4-[(4,6-Di(piperidine eridin-1-yl)-1,3,5-triazin-2-yl)amino] benzohydrazide (**2b**). White powder. Yield: 0.7 g (88%); mp 190–192 °C. IR (KBr, ν_{max} cm^{-1}): 3296 (NH), 1620 (C=O, amide). ^1H NMR (500 MHz, $\text{DMSO}-d_6$): δ 1.47–1.58 (m, 12H, 6 CH_2 , piperidine), 3.34–3.74 (m, 8H, 4 CH_2 , piperidine), 4.43 (br.s, 2H, NH_2 , D_2O exchangeable), 7.70–7.74 (m, 4H, Ar–H), 9.43 (s, 1H, NH, D_2O exchangeable), 9.51–9.59 (m, 1H, NH, D_2O exchangeable). ^{13}C NMR (125 MHz, $\text{DMSO}-d_6$): δ 24.55, 25.72, 44.25, 118.99, 119.39, 125.59, 128.03, 148.64, 163.06, 164.85, 165.97. Anal. Calcd. for $\text{C}_{20}\text{H}_{28}\text{N}_8\text{O}$: C, 60.59; H, 7.12; N, 28.26. Found: C, 60.23; H, 7.01; N, 28.50.

5.1.1.3. 4-[(4,6-Bis(phenylamino)-1,3,5-triazin-2-yl)amino] benzohydrazide (**2c**). White powder. Yield: 0.7 g (85.4%); mp 180–182 °C. IR (KBr, ν_{max} cm^{-1}): 3289 (NH), 1604 (C=O, amide).

^1H NMR (500 MHz, $\text{DMSO}-d_6$): δ 4.45 (br.s, 2H, NH_2 , D_2O exchangeable), 6.97–7.00 (m, 2H, Ar–H), 7.27–7.30 (m, 4H, Ar–H), 7.72–7.96 (m, 8H, Ar–H), 9.31–9.47 (m, 3H, NH, D_2O exchangeable), 9.63 (s, 1H, NH, D_2O exchangeable). ^{13}C NMR (125 MHz, $\text{DMSO}-d_6$): δ 119.51, 121.06, 122.70, 126.75, 127.89, 128.83, 132.82, 140.23, 143.20, 164.5, 164.61, 166.22. Anal. Calcd. for $\text{C}_{22}\text{H}_{20}\text{N}_8\text{O}$: C, 64.07; H, 4.89; N, 27.17. Found: C, 64.92; H, 4.97; N, 27.02.

5.1.1.4. 4-[(4,6-Bis(benzylamino)-1,3,5-triazin-2-yl)amino] benzohydrazide (**2d**). White powder. Yield: 0.75 g (85.2%); mp 216–218 °C. IR (KBr, ν_{max} cm^{-1}): 3427, 3276 (NH), 1613 (C=O, amide). ^1H NMR (500 MHz, $\text{DMSO}-d_6$): δ 4.47–4.51 (m, 6H, 2CH_2 & NH_2 , D_2O exchangeable), 7.19–7.31 (m, 8H, Ar–H & NH, D_2O exchangeable), 7.69–7.84 (m, 8H, Ar–H & NH, D_2O exchangeable), 9.60–9.63 (m, 2H, 2NH , D_2O exchangeable.). ^{13}C NMR (125 MHz, $\text{DMSO}-d_6$): δ 43.92, 118.85, 126.92, 127.74, 128.57, 141.0, 148.26, 164.09, 164.66, 166.43. Anal. Calcd. for $\text{C}_{24}\text{H}_{24}\text{N}_8\text{O}$: C, 65.44; H, 5.49; N, 25.44. Found: C, 65.89; H, 5.86; N, 25.51.

5.1.2. General procedure for the preparation of the target compounds (**3a–3p**)

The appropriate hydrazide **2a–d** (1 mmol) and aromatic aldehyde (1 mmol) in a mixture of methanol (60 mL) and acetic acid (2 mL) was refluxed for 8 h. Excess solvent was removed under vacuum. The

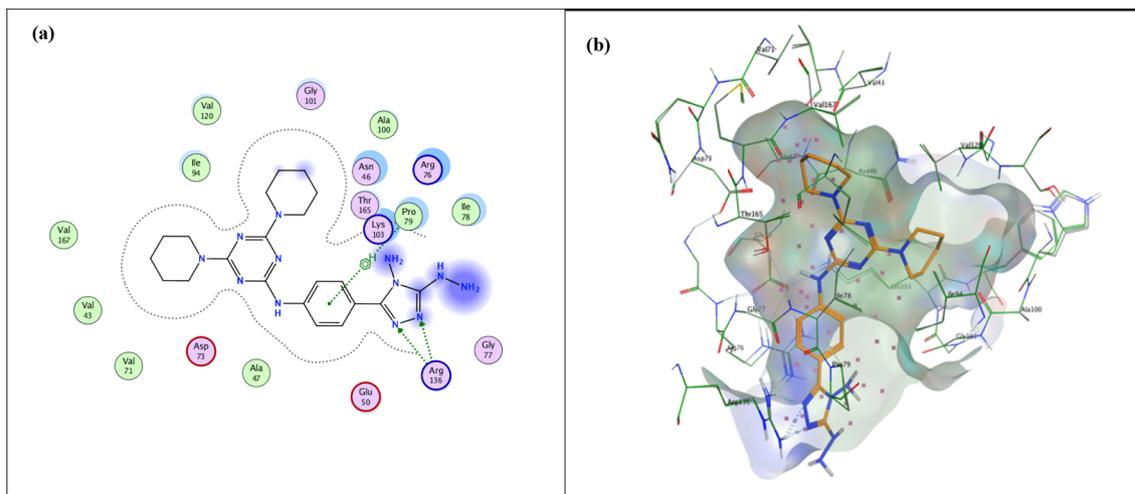


Fig. 7. (a) The 2D binding mode and (b) the 3D binding mode of **6b** in the active site of *E. coli* DNA gyrase B (PDB code: 4DUH).

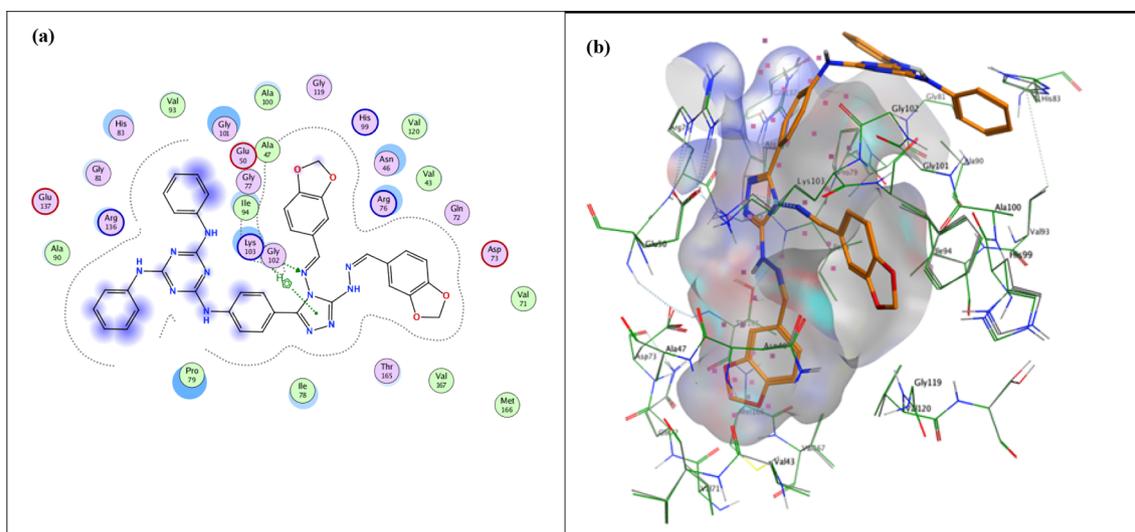


Fig. 8. (a) The 2D binding mode and (b) the 3D binding mode of **7c** in the active site of *E. coli* DNA gyrase B (PDB code: 4DUH).

remaining residue was diluted with ice water and the precipitated product was filtered off and washed with water to afford the pure products.

5.1.2.1. N'-Benzylidene-4-[(4,6-dimorpholino-1,3,5-triazin-2-yl)amino] Benzohydrazide (3a). White powder. Yield: 0.41 g (83.7%); mp 286–288 °C. IR (KBr, ν_{\max} cm^{-1}): 3218 (NH), 1646 (C=O, amide). ^1H NMR (500 MHz, DMSO- d_6) (**Isomer A: 80%**): δ 3.60–3.69 (m, 16H, 8 CH₂), 7.42–7.43 (m, 3H, Ar–H), 7.69–7.89 (m, 6H, Ar–H), 8.42 (s, 1H, CH=N), 9.46 (s, 1H, NH, D₂O exchangeable), 11.69 (s, 1H, NH, D₂O exchangeable). (**Isomer B: 20%**) δ 3.60–3.69 (m, 16H, 8 CH₂), 7.42–7.43 (m, 3H, Ar–H), 7.69–7.89 (m, 6H, Ar–H), 8.44 (s, 1H, CH=N), 9.62 (s, 1H, NH, D₂O exchangeable), 11.69 (s, 1H, NH, D₂O exchangeable). ^{13}C NMR (125 MHz, DMSO- d_6): δ 43.74, 66.44, 119.20, 127.73, 128.89, 129.32, 130.86, 134.30, 144.09, 148.85, 164.10, 164.41, 165.05. Anal. Calcd. for C₂₅H₂₈N₈O₃: C, 61.46; H, 5.78; N, 22.94. Found: C, 61.95; H, 5.17; N, 22.83.

5.1.2.2. 4-[(4,6-Dimorpholino-1,3,5-triazin-2-yl)amino]-N'-(2-furylmethylidene)benzohydrazide (3b). Yellow powder. Yield: 0.38 g (79%); mp 204–206 °C. IR (KBr, ν_{\max} cm^{-1}): 3410 (NH), 1609 (C=O, amide). ^1H NMR (500 MHz, DMSO- d_6): (**Isomer A: 71%**) δ 3.61–3.68 (m, 16H, 8 CH₂), 6.61 (s, 1H, Ar–H), 6.89 (s, 1H, Ar–H), 7.71–7.88 (m, 5H, Ar–H), 8.30 (s, 1H, CH=N), 9.45 (s, 1H, NH, D₂O exchangeable), 11.63 (s, 1H, NH, exchangeable), (**Isomer B: 29%**) δ 3.61–3.68 (m, 16H, 8 CH₂), 6.61 (s, 1H, Ar–H), 6.89 (s, 1H, Ar–H), 7.71–7.88 (m, 5H, Ar–H), 8.32 (s, 1H, CH=N), 9.62 (s, 1H, NH, D₂O exchangeable), 11.68 (s, 1H, NH, exchangeable). ^{13}C NMR (125 MHz, DMSO- d_6): δ 43.82, 66.44, 112.75, 113.32, 113.82, 119.23, 119.65, 128.91, 132.25, 144.06, 148.82, 163.80, 164.48, 165.17. Anal. Calcd. for C₂₃H₂₆N₈O₄: C, 57.73; H, 5.48; N, 23.42. Found: C, 57.38; H, 5.03; N, 23.65.

5.1.2.3. 4-[(4,6-Dimorpholino-1,3,5-triazin-2-yl)amino]-N'-[(2H-1,3-benzodioxol-5-yl)methylidene]benzohydrazide (3c). White powder. Yield: 0.41 g (77%); mp 254–256 °C. IR (KBr, ν_{\max} cm^{-1}): 3369 (NH), 1656 (C=O, amide). ^1H NMR (500 MHz, DMSO- d_6): (**Isomer A: 83%**) δ 3.60–3.67 (m, 16H, 8 CH₂), 6.06 (s, 2H, piperonal CH₂), 6.95 (d, 1H, $J = 7.6$ Hz, Ar–H), 7.12–7.13 (m, 1H, Ar–H), 7.27 (br.s, 1H, Ar–H), 7.80–7.88 (m, 4H, Ar–H), 8.32 (s, 1H, CH=N), 9.45 (s, 1H, NH, D₂O exchangeable), 11.59 (d, 1H, NH, D₂O exchangeable). (**Isomer B: 17%**) δ 3.60–3.67 (m, 16H, 8 CH₂), 6.06 (s, 2H, piperonal CH₂), 6.95 (d, 1H, $J = 7.6$ Hz, Ar–H), 7.12–7.13 (m, 1H, Ar–H), 7.27 (br.s, 1H, Ar–H), 7.80–7.88 (m, 4H, Ar–H), 8.34 (s, 1H, CH=N), 9.61 (s, 1H, NH, D₂O exchangeable), 11.62 (d, 1H, NH, D₂O exchangeable). ^{13}C NMR

(125 MHz, DMSO- d_6): δ 43.88, 66.47, 101.96, 105.59, 108.89, 118.96, 119.39, 123.67, 126.20, 126.67, 128.74, 129.38, 143.78, 144.12, 147.44, 148.44, 149.43, 163.14, 164.49, 165.20. Anal. Calcd. for C₂₈H₂₈N₈O₅: C, 60.42; H, 5.07; N, 20.13. Found: C, 59.02; H, 5.52; N, 20.01.

5.1.2.4. 4-((4,6-Dimorpholino-1,3,5-triazin-2-yl)amino)-N'-(4-hydroxybenzylidene) benzohydrazide (3d). Brown powder. Yield: 0.41 g (82%); mp 250–252 °C. IR (KBr, ν_{\max} cm^{-1}): 3349 (OH), 3255 (NH), 1647 (C=O, amide). ^1H NMR (500 MHz, DMSO- d_6): δ 3.60–3.69 (m, 16H, 8 CH₂), 6.79 (d, 1H, $J = 7$ Hz, Ar–H), 7.07 (d, 1H, $J = 7$ Hz, Ar–H), 7.16–7.32 (m, 2H, Ar–H), 7.80–7.84 (m, 4H, Ar–H), 8.31 (s, 1H, CH=N), 9.46, 9.64 (2s, each 1H, 2NH, D₂O exchangeable), 11.64 (s, 1H, OH, D₂O exchangeable). ^{13}C NMR (125 MHz, DMSO- d_6): δ 43.93, 66.47, 113.27, 119.02, 119.12, 119.43, 126.24, 126.71, 128.84, 130.27, 136.25, 143.80, 144.14, 147.76, 158.10, 163.30, 164.53, 165.27. Anal. Calcd. for C₂₅H₂₈N₈O₄: C, 59.51; H, 5.59; N, 22.21. Found: C, 59.99; H, 5.98; N, 22.30.

5.1.2.5. N'-Benzylidene-4-[(4,6-di(piperidin-1-yl)-1,3,5-triazin-2-yl) Amino]benzohydrazide (3e). White powder. Yield: 0.38 g (79%); mp 282–284 °C. IR (KBr, ν_{\max} cm^{-1}): 3233 (NH), 1650 (C=O, amide). ^1H NMR (500 MHz, DMSO- d_6): δ 1.51–1.60 (m, 12H, 6 CH₂), 3.34–3.74 (m, 8H, 4 CH₂), 7.42–7.43 (m, 3H, Ar–H), 7.69–7.85 (m, 7H, Ar–H & NH, D₂O exchangeable), 8.42 (s, 1H, CH=N), 11.73 (s, 1H, NH, D₂O exchangeable). ^{13}C NMR (125 MHz, DMSO- d_6): δ 24.33, 25.75, 45.10, 79.53, 119.77, 127.53, 129.31, 130.49, 134.85, 148.10. Anal. Calcd. for C₂₇H₃₂N₈O: C, 66.92; H, 6.66; N, 23.12. Found: C, 66.12; H, 6.11; N, 23.92.

5.1.2.6. 4-[(4,6-Di(piperidin-1-yl)-1,3,5-triazin-2-yl)amino]-N'-[(2-furyl)methylidene] benzohydrazide (3f). Yellow powder. Yield: 0.39 g (83%); mp 200–202 °C. IR (KBr, ν_{\max} cm^{-1}): 3418 (NH), 1651 (C=O, amide). ^1H NMR (500 MHz, DMSO- d_6): (**Isomer A: 70%**) δ 1.47–1.59 (m, 12H, 6 CH₂, piperidine), 3.69–3.77 (m, 8H, 4 CH₂, piperidine), 6.61 (s, 1H, Ar–H), 6.88 (s, 1H, Ar–H), 7.81–7.89 (m, 5H, Ar–H), 8.29 (s, 1H, CH), 9.31 (s, 1H, NH, D₂O exchangeable), 11.62 (s, 1H, NH, D₂O exchangeable), (**Isomer B: 30%**) δ 1.47–1.59 (m, 12H, 6 CH₂), 3.69–3.77 (m, 8H, 4CH₂), 6.61 (s, 1H, Ar–H), 6.88 (s, 1H, Ar–H), 7.81–7.89 (m, 5H, Ar–H), 8.32 (s, 1H, CH), 9.56 (s, 1H, NH, D₂O exchangeable), 11.67 (s, 1H, NH, D₂O exchangeable). ^{13}C NMR ((125 MHz, DMSO- d_6): δ 24.57, 25.79, 44.21, 56.83, 114.56, 119.40, 119.44, 128.84, 145.61, 149.50, 164.71, 164.74. Anal. Calcd. for C₂₅H₃₀N₈O₂: C, 63.27; H, 6.37; N, 23.61. Found: C, 63.01; H, 6.09; N,

23.84.

5.1.2.7. 4-[(4,6-Di(piperidin-1-yl)-1,3,5-triazin-2-yl)amino]-N'-[(2H-1,3-benzodioxol-5-yl)methylidene]benzohydrazide (**3g**). White powder. Yield: 0.40 g (75.7%); mp 230–232 °C. IR (KBr, ν_{\max} cm⁻¹): 3313 (NH), 1711 (C=O, amide). ¹H NMR (500 MHz, DMSO-*d*₆): δ 1.47–1.59 (m, 12H, 6 CH₂), 3.69–3.77 (m, 8H, 4 CH₂), 6.06 (s, 2H, CH₂ piperonal), 6.96 (d, 1H, *J* = 6.85 Hz, Ar–H), 7.11–7.13 (m, 1H, Ar–H), 7.27 (s, 1H, Ar–H), 7.72–7.90 (m, 4H, Ar–H), 8.32 (s, 1H, CH=N), 9.29, 11.57 (2 s, each 1H, 2NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): δ 24.83, 25.85, 44.16, 101.97, 105.57, 108.91, 118.63, 120.43, 123.57, 125.88, 127.93, 128.69, 129.44, 144.49, 147.28, 148.45, 149.41, 163.04, 164.61, 164.94. Anal. Calcd. for C₃₀H₃₂N₈O₃: C, 65.20; H, 5.84; N, 20.28. Found: C, 65.01; H, 5.65; N, 20.87.

5.1.2.8. 4-[(4,6-Di(piperidin-1-yl)-1,3,5-triazin-2-yl)amino]-N'-(4-hydroxybenzylidene)benzohydrazide (**3h**). White powder. Yield: 0.41 g (82%); mp 212–214 °C. IR (KBr, ν_{\max} cm⁻¹): 3600–3200 (OH), 3279 (NH), 1647 (C=O, amide) cm⁻¹. ¹H NMR (500 MHz, DMSO-*d*₆): δ 1.51–1.63 (m, 12H, 6 CH₂, piperidine), 3.73–3.82 (m, 8H, 4 CH₂, piperidine), 6.82–6.84 (m, 1H, Ar–H), 7.10–7.28 (m, 3H, Ar–H), 7.76–7.93 (m, 5H, Ar–H & NH, D₂O exchangeable), 8.37 (s, 1H, CH=N), 9.64 (s, 1H, NH, D₂O exchangeable), 11.64–11.75 (m, 1H, OH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): δ 24.79, 25.82, 44.22, 114.21, 118.33, 118.91, 119.16, 121.21, 125.91, 126.84, 128.86, 131.27, 136.98, 145.50, 148.98, 159.12, 164.66, 165.03. Anal. Calcd for C₂₇H₃₂N₈O₂: C, 64.78; H, 6.44; N, 22.38. Found: C, 64.69; H, 6.35; N, 22.29.

5.1.2.9. N'-Benzylidene-4-[(4,6-bis(phenylamino)-1,3,5-triazin-2-yl)amino]benzohydrazide (**3i**). White powder. Yield: 0.36 g (72%); mp is > 300 °C. IR (KBr, ν_{\max} cm⁻¹): 3399 (NH), 1647 (C=O, amide). ¹H NMR (500 MHz, DMSO-*d*₆): δ 7.22–7.40 (m, 5H, Ar–H), 7.70–7.97 (m, 14H, Ar–H), 8.16 (s, 1H, CH=N), 9.62 (br.s, 3H, 3NH, D₂O exchangeable), 11.72–11.74 (m, 1H, NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): δ 119.89, 121.12, 122.21, 125.36, 127.54, 127.89, 129.29, 129.47, 134.24, 141.28, 144.16, 148.22, 164.29, 164.43, 166.41. Anal. Calcd for C₂₉H₂₄N₈O: C, 69.59; H, 4.83; N, 22.39. Found: C, 69.76; H, 5.00; N, 22.11.

5.1.2.10. 4-[(4,6-Bis(phenylamino)-1,3,5-triazin-2-yl)amino]-N'-(2-furylmethylidene)benzohydrazide (**3j**). Brown powder. Yield: 0.37 g (75.5%); mp is 210–212 °C. IR (KBr, ν_{\max} cm⁻¹): 3283 (NH), 1646 (C=O, amide). ¹H NMR (500 MHz, DMSO-*d*₆): δ 6.62 (s, 1H, Ar–H), 6.90 (s, 1H, Ar–H), 6.98–7.00 (m, 1H, Ar–H), 7.17–7.29 (m, 3H, Ar–H), 7.78–8.07 (m, 11H, Ar–H), 8.33 (s, 1H, CH=N), 9.50–9.84 (m, 3H, 3NH, D₂O exchangeable), 11.68 (s, 1H, NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): δ 112.65, 113.61, 119.47, 120.23, 121.09, 122.79, 128.86, 140.18, 145.49, 150.07, 164.5, 164.62. Anal. Calcd. for C₂₇H₂₂N₈O₂: C, 66.11; H, 4.52; N, 22.84. Found: C, 66.01; H, 4.42; N, 22.97.

5.1.2.11. 4-[(4,6-Bis(phenylamino)-1,3,5-triazin-2-yl)amino]-N'-[(2H-1,3-benzodioxol-5-yl)methylidene]benzohydrazide (**3k**). Brown powder. Yield: 0.43 g (79.0%); mp is > 300 °C. IR (KBr, ν_{\max} cm⁻¹): 3294 (NH), 1712 (C=O, amide). ¹H NMR (400 MHz, DMSO-*d*₆): δ 6.10 (s, 2H, CH₂), 6.95–7.04 (m, 6H, Ar–H), 7.33–7.80 (m, 12H, 11 Ar–H & CH=N), 9.6 (s, 3H, 3 NH, D₂O exchangeable), 11.52 (s, 1H, 1 NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): δ 101.16, 106.24, 108.92, 115.98, 117.71, 118.93, 120.22, 120.45, 120.56, 121.21, 121.47, 128.80, 128.88, 129.08, 129.27, 143.63, 147.73, 148.50, 148.50, 148.95, 163.81, 163.86, 165.94. Anal. Calcd for C₃₂H₂₄N₈O₃: C, 67.60; H, 4.25; N, 19.71. Found: C, 67.55; H, 4.20; N, 19.96.

5.1.2.12. 4-[(4,6-Bis(phenylamino)-1,3,5-triazin-2-yl)amino]-N'-(4-

hydroxybenzylidene)benzohydrazide (**3l**). Grey powder. Yield: 0.40 g (78%); mp is 182–184 °C. IR (KBr, ν_{\max} cm⁻¹): 3500–3215 (OH), 3385 (NH), 1576 (C=O, amide). ¹H NMR (500 MHz, DMSO-*d*₆): δ 6.82 (s, 1H, Ar–H), 7.00–7.09 (m, 3H, Ar–H), 7.20–7.29 (m, 6H, Ar–H), 7.78–7.98 (m, 8H, Ar–H), 8.36 (s, 1H, CH=N), 9.35 (s, 1H, NH, D₂O exchangeable), 9.58–9.65 (m, 3H, 3NH, D₂O exchangeable), 11.70 (s, 1H, OH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): δ 113.34, 115.16, 117.85, 119.20, 119.56, 121.17, 122.33, 122.91, 126.64, 126.85, 128.87, 130.33, 130.75, 136.19, 140.06, 143.82, 148.08, 150.32, 158.05, 163.58, 164.55, 164.67. Anal. Calcd for C₂₉H₂₄N₈O₂: C, 67.43; H, 4.68; N, 21.69. Found: C, 67.58; H, 4.83; N, 21.84.

5.1.2.13. N'-Benzylidene-4-[(4,6-bis(benzylamino)-1,3,5-triazin-2-yl)amino]benzohydrazide (**3m**). White powder. Yield: 0.39 g (73%); mp is 290–292 °C. IR (KBr, ν_{\max} cm⁻¹): 3417 (NH), 1618 (C=O, amide). ¹H NMR (500 MHz, DMSO-*d*₆): (**Isomer A: 72%**): δ 4.54–4.41 (m, 4H, 2 CH₂), 7.22–7.50 (m, 12H, Ar–H), 7.70–7.86 (m, 7H, Ar–H), 8.43 (s, 1H, CH=N), 9.2–9.61 (m, 3H, 3 NH, D₂O exchangeable), 10.70–11.71 (s, 1H, NH, D₂O exchangeable). (**Isomer B: 28%**): δ 4.41–4.54 (m, 4H, 2 CH₂), 7.22–7.50 (m, 12H, Ar–H), 7.70–7.86 (m, 7H, Ar–H), 8.47 (s, 1H, CH=N), 9.2–9.61 (m, 3H, 3 NH, D₂O exchangeable), 10.70 (s, 1H, NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): δ 43.94, 119.13, 119.88, 125.15, 127.63, 128.71, 128.88, 129.26, 130.39, 144.33, 148.38, 164.64, 165.32, 166.02. Anal. Calcd for C₃₁H₂₈N₈O: C, 70.44; H, 5.34; N, 21.20. Found: C, 70.29; H, 5.19; N, 21.55.

5.1.2.14. 4-[(4,6-Bis(benzylamino)-1,3,5-triazin-2-yl)amino]-N'-(2-furylmethylidene)benzohydrazide (**3n**). Yellow powder. Yield: 0.41 g (78.8%); mp is 206–208 °C. IR (KBr, ν_{\max} cm⁻¹): 3283 (NH), 1643 (C=O, amide). ¹H NMR (500 MHz, DMSO-*d*₆): δ 4.41–4.53 (m, 4H, 2 CH₂), 6.60 (s, 1H, Ar–H), 6.89 (s, 1H, Ar–H), 7.18–7.32 (m, 7H, Ar–H), 7.51–7.95 (m, 8H, Ar–H), 8.29 (s, 1H, CH=N), 9.21–9.98 (br.s, 3H, NH, D₂O exchangeable), 11.61–11.69 (m, 1H, 1 NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): δ 43.91, 112.73, 114.11, 114.23, 119.03, 119.34, 120.28, 127.12, 127.52, 128.69, 128.94, 138.03, 140.61, 145.56, 149.81, 163.63, 164.49, 164.71. Anal. Calcd for C₂₉H₂₆N₈O₂: C, 67.17; H, 5.05; N, 21.61. Found: C, 67.29; H, 5.17; N, 21.13.

5.1.2.15. 4-[(4,6-Bis(benzylamino)-1,3,5-triazin-2-yl)amino]-N'-[(2H-1,3-benzodioxol-5-yl)methylidene]benzohydrazide (**3o**). White powder. Yield: 0.47 g (82.1%); mp is 204–206 °C. IR (KBr, ν_{\max} cm⁻¹): 3412 (NH), 1687 (C=O, amide). ¹H NMR (400 MHz, DMSO-*d*₆): δ 4.53–4.61 (m, 4H, 2 CH₂), 6.11 (s, 2H, CH₂, piperonal), 7.01–7.03 (m, 2H, Ar–H), 7.15–7.17 (m, 11H, Ar–H), 7.78–7.85 (m, 4H, Ar–H), 8.37 (s, 1H, CH=N), 8.56 (s, 2H, 2 NH, D₂O exchangeable), 8.97–9.04 (m, 1H, NH, D₂O exchangeable), 11.37–11.41 (m, 1H, NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): δ 44.02, 101.93, 102.12, 105.73, 106.70, 108.98, 118.93, 123.35, 125.29, 126.94, 127.63, 128.57, 128.95, 129.61, 140.45, 148.48, 149.42, 150.48, 160.66, 164.71, 166.51. Anal. Calcd for C₃₄H₂₈N₈O₃: C, 68.44; H, 4.73; N, 18.78. Found: C, 68.38; H, 4.66; N, 18.71.

5.1.2.16. 4-[(4,6-Bis(benzylamino)-1,3,5-triazin-2-yl)amino]-N'-(4-hydroxybenzylidene)benzohydrazide (**3p**). Brown powder. Yield: 0.42 g (77.7%); mp is 164–166 °C. IR (KBr, ν_{\max} cm⁻¹): 3500–3200 (OH), 3409 (NH), 1647 (C=O, amide). ¹H NMR (500 MHz, DMSO-*d*₆): δ 4.43–4.55 (m, 4H, 2 CH₂), 6.80 (s, 1H, Ar–H), 7.06–7.30 (m, 13H, Ar–H), 7.74–7.89 (m, 5H, Ar–H & NH, D₂O exchangeable), 8.34 (br.s, 1H, CH=N), 9.18–9.26 (m, 1H, NH, D₂O exchangeable), 9.60–9.69 (m, 2H, OH & NH, D₂O exchangeable), 11.59–11.71 (m, 1H, NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): δ 43.89, 113.13, 117.73, 118.78, 119.13, 120.38, 126.98, 127.46, 128.63, 130.32, 136.24, 140.95, 148.11, 158.10, 161.75, 164.60, 166.33. Anal. Calcd. for C₃₁H₂₈N₈O₂: C, 68.37; H, 5.18; N, 20.58. Found: C, 68.24; H, 5.05; N, 20.45.

5.1.3. General procedure for the preparation of 1,3,4-oxadiazole-2-thiol derivatives (**4b-d**)

To a solution of the appropriate **2b-d** (2.5 mmol) in dimethylformamide (15 mL), carbon disulphide (0.15 mL, 2.5 mmol) and potassium hydroxide (0.14 g, 2.5 mmol, in 2 mL water) were added while stirring. The reaction mixture was then heated under reflux for 12 h. Excess solvent was removed under vacuum, and the remaining residue was diluted with ice water. The crude products obtained were filtered off, washed with water, dried, and recrystallized from ethanol.

5.1.3.1. 5-[4-((4,6-Di(piperidin-1-yl)-1,3,5-triazin-2-yl)amino)phenyl]-1,3,4-oxadiazole-2-thiol (4b**).** White powder, Yield: 0.80 g (72.7%); mp 290–292 °C. IR (KBr, ν_{\max} cm⁻¹): 3288 (NH), 2852 (SH), 1606 (C=N). ¹H NMR (500 MHz, DMSO-*d*₆): δ 1.57–1.68 (m, 12H, 6 CH₂, piperidine), 3.12 (s, 1H, SH, D₂O exchangeable), 3.78–3.86 (m, 8H, 4 CH₂, piperidine), 7.80–7.97 (m, 4H, Ar–H), 9.23–9.49 (m, 1H, NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): δ 25.81, 44.21, 52.06, 119.65, 122.15, 128.56, 129.17, 131.89, 145.84, 161.31, 164.94, 166.59. Anal. Calcd. for C₂₁H₂₆N₈OS: C, 57.51; H, 5.98; N, 25.55; O, 3.65. Found: C, 57.42; H, 5.76; N, 25.43.

5.1.3.2. 5-(4-((4,6-Bis(phenylamino)-1,3,5-triazin-2-yl)amino)phenyl)-1,3,4-oxadiazole-2-thiol (4c**).** White powder. Yield: 0.85 g (77.25%); mp 160–162 °C. IR (KBr, ν_{\max} cm⁻¹): 3398 (NH), 2924 (SH), 1604 (C=N). ¹H NMR (500 MHz, DMSO-*d*₆): δ 2.94 (s, 1H, SH, D₂O exchangeable), 6.97–7.01 (m, 2H, Ar–H), 7.27–7.50 (m, 4H, Ar–H), 7.63–8.10 (m, 9H, Ar–H & 1NH, D₂O exchangeable), 9.29–9.68 (m, 2H, 2NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): δ 120.35, 121.16, 122.83, 127.04, 128.13, 128.84, 129.78, 131.96, 140.16, 164.62. Anal. Calcd. for C₂₃H₁₈N₈OS: C, 60.78; H, 3.99; N, 24.65. Found: C, 60.61; H, 3.80; N, 24.36.

5.1.3.3. 5-(4-((4,6-Bis(benzylamino)-1,3,5-triazin-2-yl)amino)phenyl)-1,3,4-oxadiazole-2-thiol (4d**).** White powder. Yield: 0.88 g (85.4%); mp 138–140 °C. IR (KBr, ν_{\max} cm⁻¹): 3398 (NH), 2924 (SH), 1609 (C=N). ¹H NMR (500 MHz, DMSO-*d*₆): δ 3.01 (s, 1H, SH, D₂O exchangeable), 4.40–4.47 (m, 4H, 2CH₂), 7.17–7.29 (m, 10H, Ar–H), 7.63–8.00 (m, 6H, Ar–H & 2NH, D₂O exchangeable), 9.35–9.41 (m, 1H, NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): δ 43.92, 118.57, 126.14, 126.92, 127.74, 128.63, 151.57, 159.63, 164.63, 164.42, 165.06. Anal. Calcd. for C₂₅H₂₂N₈OS: C, 62.22; H, 4.60; N, 23.22. Found: C, 62.02; H, 4.40; N, 23.01.

5.1.4. General procedure for the preparation of target derivatives (**5b-d**)

Sodium hydroxide (0.096 g, 2.4 mmol) was added while stirring at room temperature to a solution of the appropriate **4b-d** (1 mmol) in ethanol (20 mL). Methyl iodide was then added (0.22 mL, 3.5 mmol) over a period of 1 h and the reaction mixture was stirred overnight at room temperature. The precipitated solid was filtered off, washed with water and dried to afford the pure products.

5.1.4.1. N-{4-[5-(Methylsulfanyl)-1,3,4-oxadiazol-2-yl]phenyl}-4,6-di(piperidin-1-yl)-1,3,5-triazin-2-amine (5b**).** White powder. Yield: 0.41 g (95.3%); mp 234–236 °C. IR (KBr, ν_{\max} cm⁻¹): 3379 (NH), 1607 (C=N). ¹H NMR (500 MHz, DMSO-*d*₆): δ 1.47–1.58 (m, 12H, 6 CH₂, piperidine), 3.23–3.33 (m, 3H, CH₃), 3.68 (s, 8H, 4 CH₂, piperidine), 7.81–7.87 (m, 4H, Ar–H), 9.40 (s, 1H, NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): δ 15.95, 24.73, 24.81, 44.25, 118.82, 119.41, 126.26, 127.88, 128.49, 130.36, 164.71, 165.16, 168.14. Anal. Calcd. for C₂₂H₂₈N₈OS: C, 58.39; H, 6.24; N, 24.76. Found: C, 58.28; H, 6.13; N, 24.65.

5.1.4.2. N²-{4-[5-(Methylsulfanyl)-1,3,4-oxadiazol-2-yl]phenyl}-N⁴,N⁶-diphenyl-1,3,5-triazine-2,4,6-triamine (5c**).** White powder. Yield: 0.36 g (80%); mp 170–172 °C. IR (KBr, ν_{\max} cm⁻¹): 3397 (NH), 1588 (C=N). ¹H NMR (500 MHz, DMSO-*d*₆): δ 2.79 (s, 3H, CH₃), 7.02–7.06 (m, 2H,

Ar–H), 7.30–7.34 (m, 3H, Ar–H), 7.71–7.81 (m, 7H, Ar–H) 8.07 (s, 2H, Ar–H), 9.10–9.15 (m, 2H, 2NH, D₂O exchangeable), 9.47 (s, 1H, 1NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): δ 14.91, 116.63, 120.59, 121.33, 122.84, 127.28, 128.77, 140.19, 144.05, 164.63, 164.80, 165.73. Anal. Calcd. for C₂₄H₂₀N₈OS: C, 61.52; H, 4.30; N, 23.92. Found: C, 61.64; H, 4.42; N, 40.04.

5.1.4.3. N²,N⁴-Dibenzyl-N⁶-{4-[5-(methylsulfanyl)-1,3,4-oxadiazol-2-yl]phenyl}-1,3,5-triazine-2,4,6-triamine (5d**).** White powder, Yield: 0.40 g (87%); mp 254–256 °C. IR (KBr, ν_{\max} cm⁻¹): 3402 (NH), 1572 (C=N). ¹H NMR (500 MHz, DMSO-*d*₆): δ 2.99 (s, 3H, CH₃), 4.39–4.45 (m, 4H, 2CH₂), 7.18–7.27 (m, 12H, Ar–H), 7.66–7.90 (m, 4H, Ar–H & 2 NH, D₂O exchangeable), 9.12–9.22 (m, 1H, NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): δ 14.88, 39.67, 127.24, 127.43, 127.73, 128.02, 128.61, 128.86, 140.84, 164.53, 166.11. Anal. Calcd. for C₂₆H₂₄N₈OS: C, 62.89; H, 4.87; N, 22.56. Found: C, 62.61; H, 4.79; N, 22.38.

5.1.5. General procedure for the preparation of target derivatives (**6b-d**)

A solution of the appropriate **5b-d** (1 mmol) and hydrazine hydrate (3 mL) in absolute ethanol (5 mL) was heated under reflux for 6 h. The reaction mixture was then cooled to room temperature and the precipitated product was filtered off, washed with cold ethanol to afford the pure products.

5.1.5.1. N-{4-[4-Amino-5-hydrazinyl-4H-1,2,4-triazol-3-yl]phenyl}-4,6-di(piperidin-1-yl)-1,3,5-triazin-2-amine (6b**).** White powder. Yield: 0.42 g (93.3%); mp 140–142 °C. IR (KBr, ν_{\max} cm⁻¹): 3340 (NH), 1600 (C=N). ¹H NMR (500 MHz, DMSO-*d*₆): δ 1.46–1.59 (m, 12H, 6 CH₂, piperidine), 3.68 (br.s, 8H, 4 CH₂, piperidine), 4.39 (s, 2H, NH₂, D₂O exchangeable), 7.69–7.71 (m, 6H, Ar–H & NH₂, D₂O exchangeable), 9.20 (s, 1H, NH, D₂O exchangeable), 9.52 (s, 1H, NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): 24.75, 25.81, 44.19, 118.65, 118.78, 128.57, 130.53, 164.62, 164.97. Anal. Calcd. for C₂₁H₃₀N₁₂: C, 55.98; H, 6.71; N, 37.31. Found: C, 56.12; H, 6.85; N, 37.55.

5.1.5.2. N²-{4-[4-amino-5-hydrazinyl-4H-1,2,4-triazol-3-yl]phenyl}-N⁴,N⁶-diphenyl-1,3,5-triazine-2,4,6-triamine (6c**).** White powder. Yield: 0.46 g (92.5%); mp 152–154 °C. IR (KBr, ν_{\max} cm⁻¹): 3396 (NH), 1605 (C=N). ¹H NMR (500 MHz, DMSO-*d*₆): δ 4.40 (br., 2H, NH₂, D₂O exchangeable), 6.98 (s, 2H, NH₂, D₂O exchangeable), 7.26–7.27 (m, 5H, Ar–H), 7.71–7.76 (m, 6H, Ar–H), 7.87–7.98 (m, 3H, Ar–H), 9.31 (br.s, 2H, 2NH, D₂O exchangeable), 9.47, 9.64 (2br.s, each 1H, 2NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): 119.52, 121.07, 122.73, 127.94, 128.84, 140.21, 164.61. Anal. Calcd. for C₂₃H₂₂N₁₂: C, 59.22; H, 4.75; N, 36.03. Found: C, 59.10; H, 4.63; N, 35.91.

5.1.5.3. N₂-(4-(4-amino-5-hydrazinyl-4H-1,2,4-triazol-3-yl)phenyl)-N₄,N₆-dibenzyl-1,3,5-triazine-2,4,6-triamine (6d**).** White powder. Yield: 0.43 g (89.6%); mp 160–162 °C. IR (KBr, ν_{\max} cm⁻¹): 3412 (NH), 1611 (C=N). ¹H NMR (500 MHz, DMSO-*d*₆): δ 4.40–4.46 (m, 6H, 2CH₂ & NH₂, D₂O exchangeable), 7.19–7.28 (m, 12H, Ar–H), 7.61–7.82 (m, 6H, Ar–H, 2NH & NH₂, D₂O exchangeable), 9.08–9.17 (m, 1H, NH, D₂O exchangeable), 9.55 (br.s, 1H, NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): 43.02, 91.81, 123.30, 128.32, 129.10, 130.14, 132.07, 132.99, 134.66, 143.40, 161.46, 163.53, 163.81. Anal. Calcd. for C₂₅H₂₆N₁₂: C, 60.71; H, 5.30; N, 33.99. Found: C, 60.56; H, 5.15; N, 33.84.

5.1.6. General procedure for the preparation of target derivatives (**7a-7f**)

A solution of the appropriate **6a-f** (0.2 g, 0.44 mmol) and the aromatic aldehyde (0.44 mmol) in a mixture of methanol (20 mL) and acetic acid (2 mL) was heated under reflux for 10 h. Excess solvent was removed under vacuum and the crude product was then diluted with ice water. The precipitated solid was filtered off and washed with water

to afford the pure products.

5.1.6.1. *N*-{4-[4-((2*H*-1,3-Benzodioxol-5-ylmethylidene)amino)-5-(2-(2*H*-1,3-benzodioxol-5-ylmethylidene)hydrazinyl)-4*H*-1,2,4-triazol-3-yl]phenyl]-4,6-di(piperidin-1-yl)-1,3,5-triazine-2-amine (7a)}. White powder. Yield: 0.25 g (79.6%); mp 204–206 °C. IR (KBr, ν_{\max} cm⁻¹): 3422 (NH), 1499 (C=N). ¹H NMR (500 MHz, DMSO-*d*₆): δ 1.47–1.58 (m, 12H, 6 CH₂, piperidine), 3.68 (s, 8H, 4 CH₂, piperidine), 6.08 (s, 4H, 2 CH₂, piperonal), 7.00–7.02 (m, 3H, Ar-H), 7.31–7.37 (m, 4H, Ar-H & NH, D₂O exchangeable), 7.70–7.80 (m, 4H, Ar-H), 8.56 (s, 2H, 2 CH=N) 9.30 (br.s, 1H, NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): 24.83, 25.85, 44.17, 102.17, 106.53, 109.04, 118.63, 125.51, 128.82, 148.45, 150.50, 161.04, 164.94. Anal. Calcd. for C₃₇H₃₈N₁₂O₄: C, 62.17; H, 5.36; N, 23.52. Found: C, 62.33; H, 5.52; N, 23.78.

5.1.6.2. *N*-{4-[4-((2-Furylmethylidene)amino)-5-(2-(furan-2-ylmethylene)hydrazinyl)-4*H*-1,2,4-triazol-3-yl]phenyl]-4,6-di(piperidin-1-yl)-1,3,5-triazine-2-amine (7b)}. White powder. Yield: 0.22 g (81.5%); mp 98–100 °C. IR (KBr, ν_{\max} cm⁻¹): 3428 (NH), 1497 (C=N). ¹H NMR (500 MHz, DMSO-*d*₆): δ 1.47–1.59 (m, 12H, 6 CH₂, piperidine), 3.91 (br.s, 8H, 4 CH₂, piperidine), 6.54–7.11 (m, 4H, Ar-H), 7.54–7.93 (m, 6H, Ar-H), 8.54 (s, 2H, 2 CH=N), 9.11–9.54 (m, 2H, 2 NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): 25.82, 44.17, 61.91, 74.54, 80.49, 114.33, 117.97, 1189.69, 121.41, 129.33, 133.34, 145.44, 149.36, 164.90. Anal. Calcd. for C₃₁H₃₄N₁₂O₂: C, 61.37; H, 5.65; N, 27.70. Found: C, 61.20; H, 5.48; N, 27.53.

5.1.6.3. *N*²-{4-[4-((2*H*-1,3-Benzodioxol-5-ylmethylidene)amino)-5-(2-(2*H*-1,3-benzodioxol-5-ylmethylidene)hydrazinyl)-4*H*-1,2,4-triazol-3-yl]phenyl]-*N*⁴,*N*⁶-diphenyl-1,3,5-triazine-2,4,6-triamine (7c)}. White powder. Yield: 0.19 g (95%); mp 237–239 °C. IR (KBr, ν_{\max} cm⁻¹): 3395 (NH), 1495 (C=N). ¹H NMR (500 MHz, DMSO-*d*₆): δ 6.06–6.08 (m, 4H, 2 CH₂, piperonal), 6.96–6.99 (m, 4H, Ar-H), 7.14–7.15 (m, 1H, Ar-H), 7.29–7.30 (m, 6H, Ar-H), 7.77–7.83 (m, 7H, Ar-H), 7.96 (br.s, 2H, Ar-H), 8.35, 8.56 (2s, each 1H, 2 CH=N), 9.34 (s, 2H, 2 NH, D₂O exchangeable), 9.56, 11.63 (2s, each 1H, 2NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): 101.98, 102.16, 105.61, 106.53, 108.94, 109.04, 119.48, 121.08, 122.77, 123.63, 125.51, 126.71, 128.85, 129.40, 140.19, 147.61, 148.45, 149.45, 150.50, 161.04, 164.51, 164.62. Anal. Calcd. for C₃₉H₃₀N₁₂O₄: C, 64.10; H, 4.14; N, 23.00. Found: C, 63.89; H, 3.93; N, 22.79.

5.1.6.4. *N*²-{4-[4-((2-Furylmethylidene)amino)-5-(2-(2-furylmethylidene)hydrazinyl)-4*H*-1,2,4-triazol-3-yl]phenyl]-*N*⁴,*N*⁶-diphenyl-1,3,5-triazine-2,4,6-triamine (7d)}. White powder. Yield: 0.17 g (85%); mp 140–142 °C. IR (KBr, ν_{\max} cm⁻¹): 3399 (NH), 1495 (C=N). ¹H NMR (500 MHz, DMSO-*d*₆): δ 6.62 (s, 2H, Ar-H), 6.97–6.98 (m, 2H, Ar-H), 7.24–7.32 (m, 6H, Ar-H), 7.76–7.82 (m, 10H, Ar-H), 8.27–8.36 (m, 2H, 2 CH=N), 9.22–9.56 (m, 3H, 3NH, D₂O exchangeable), 11.61–11.69 (m, 1H, NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): 113.11, 117.50, 117.77, 119.46, 121.07, 122.81, 128.86, 140.14, 145.88, 147.05, 149.56, 150.86, 164.61. Anal. Calcd. for C₃₃H₂₆N₁₂O₂: C, 63.66; H, 4.21; N, 26.99. Found: C, 63.41; H, 4.06; N, 26.84.

5.1.6.5. *N*²-{4-[4-((2*H*-1,3-Benzodioxol-5-ylmethylidene)amino)-5-(2-(2*H*-1,3-benzodioxol-5-ylmethylidene)hydrazinyl)-4*H*-1,2,4-triazol-3-yl]phenyl]-*N*⁴,*N*⁶-dibenzyl-1,3,5-triazine-2,4,6-triamine (7e)}. White powder. Yield: 0.27 g (81.8%); mp 220–222 °C. IR (KBr, ν_{\max} cm⁻¹): 3415 (NH), 1502 (C=N). ¹H NMR (500 MHz, DMSO-*d*₆): δ 4.42–4.47 (m, 4H, 2CH₂), 6.06–6.08 (m, 4H, 2 CH₂, piperonal), 7.00–7.02 (m, 3H, Ar-H), 7.18–7.20 (m, 5H, Ar-H), 7.31–7.33 (m, 8H, Ar-H), 7.37 (s, 3H, Ar-H), 7.74–7.86 (m, 3H, Ar-H & 2 NH, D₂O exchangeable), 8.56 (s, 2H, 2 CH=N), 9.15–9.30 (m, 1H, NH, D₂O exchangeable), 11.68 (s, 1H, NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): 40.33, 102.14, 106.53, 109.07, 118.83, 125.61, 127.02, 127.62, 128.13, 128.65,

148.41, 161.11. Anal. Calcd. for C₄₁H₃₄N₁₂O₄: C, 64.90; H, 4.52; N, 22.15. Found: C, 65.10; H, 4.72; N, 22.35.

5.1.6.6. *N*²,*N*⁴-Dibenzyl-*N*⁶-(4-((2-furylmethylidene)amino)-5-(2-(2-furylmethylidene)hydrazinyl)-4*H*-1,2,4-triazol-3-yl)phenyl-1,3,5-triazine-2,4,6-triamine (7f)}. White powder. Yield: 0.19 g (72.2%); mp 186–188 °C. IR (KBr, ν_{\max} cm⁻¹): 3414 (NH), 1511 (C=N). ¹H NMR (500 MHz, DMSO-*d*₆): δ 4.37–4.47 (m, 4H, 2CH₂), 6.60 (s, 2H, Ar-H), 6.87 (s, 2H, Ar-H), 7.20–7.29 (m, 13H, Ar-H), 7.64–7.92 (m, 5H, Ar-H & 2NH, D₂O exchangeable), 8.32, 8.84 (2s, each 1H, 2 CH=N), 9.12–9.28 (m, 1H, NH, D₂O exchangeable), 11.68 (s, 1H, NH, D₂O exchangeable). ¹³C NMR (125 MHz, DMSO-*d*₆): 43.89, 118.76, 126.96, 127.46, 127.62, 128.11, 128.61, 129.10, 131.96, 132.18, 140.44, 140.99, 164.60, 166.41, 167.41. Anal. Calcd. for C₃₅H₃₀N₁₂O₂: C, 64.60; H, 4.65; N, 25.83. Found: C, 64.36; H, 4.41; N, 25.69.

5.2. *In vitro* antimicrobial evaluation

The *in vitro* antimicrobial screening was carried out according to Hawkey et al. [46] and Murray et al. [47], Supporting Information (S1).

5.3. Molecular docking

Molecular operating environment (MOE) software package version 2016.0802 was used to perform docking simulations, Supporting Information (S2).

Declaration of Competing Interest

The authors declare that there is no conflict of interest.

Acknowledgments

The authors thank the Science and Technology Development Fund (STDF), Egypt, for funding this work through the Innovation Grant (Proposal ID 15053).

Appendix A. Supplementary material

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.bioorg.2019.103013>.

References

- [1] N. Sunduru, M. Sharma, K. Srivastava, S. Rajakumar, S. Puri, J.K. Saxena, P.M.S. Chauhana, Synthesis of oxalamide and triazine derivatives as a novel class of hybrid 4-aminoquinoline with potent antiparasitic activity, *Bioorg. Med. Chem.* 17 (2009) 6451–6462.
- [2] N. Suree, M.E. Jung, R.T. Clubb, Recent advances towards new anti-infective agents that inhibit cell surface protein anchoring in *Staphylococcus aureus* and other gram-positive pathogens, *Mini-Rev. Med. Chem.* 7 (2007) 991–1000.
- [3] C. Zhou, J. Min, Z. Liu, A. Young, H. Deshazer, T. Gao, Y.-T. Chang, N.R. Kallenbach, Synthesis and biological evaluation of novel 1,3,5-triazine derivatives as antimicrobial agents, *Bioorg. Med. Chem. Lett.* 18 (2008) 1308–1311.
- [4] A. Solankee, K. Kapadia, A. Ciric, M. Sokovic, I. Doytchinova, A. Geronikaki, Synthesis of some new S-triazine based chalcones and their derivatives as potent antimicrobial agents, *Eur. J. Med. Chem.* 45 (2010) 510–518.
- [5] R.V. Patel, P. Kumari, K.H. Chikhaliya, Design, synthesis and antimicrobial screening of s-triazinyl piperazine and piperidine derivatives, *Int. J. Adv. Pharm. Sci.* 1 (2010) 395–403.
- [6] S. Melato, D. Prosperi, P. Coghi, N. Basilio, D. Monti, A Combinatorial Approach to 2, 4,6-trisubstituted triazines with potent antimalarial activity: combining conventional synthesis and microwave-assistance, *Chem. Med. Chem.* 3 (2008) 873–876.
- [7] Y.-Z. Xiong, F.-E. Chen, J. Balzarini, E. De Clercq, C. Pannecouque, Non-nucleoside HIV-1 reverse transcriptase inhibitors. Part 11: structural modulations of diaryl-triazines with potent anti-HIV activity, *Eur. J. Med. Chem.* 43 (2008) 1230–1236.
- [8] M. Saleh, S. Abbott, V. Perron, C. Lauzon, C. Penney, B. Zacharie, Synthesis and antimicrobial activity of 2-fluorophenyl-4,6-disubstituted[1,3,5]triazines, *Bioorg. Med. Chem. Lett.* 20 (2010) 945–949.
- [9] Y.K. Rao, S.-H. Fang, Y.-M. Tzeng, Differential effects of synthesized 2'-oxygenated chalcone derivatives: modulation of human cell cycle phase distribution, *Bioorg.*

- Med. Chem. 12 (10) (2004) 2679–2686.
- [10] L. Svetaz, A. Tapia, S.N. Lopez, R.L.E. Furlan, E. Petenatti, R. Pioli, G. Schmeda-Hirschmann, S.A. Zacchino, Antifungal chalcones and new caffeic acid esters from *Zuccagnia punctata* acting against soybean infecting fungi, *J. Agric. Food. Chem.* 52 (11) (2004) 3297–3300.
- [11] Z. Nowakowska, A review of anti-infective and anti-inflammatory chalcones, *Eur. J. Med. Chem.* 42 (2) (2007) 125–137.
- [12] Sh.N. Khattab, H.H. Khalil, A.A. Bekhit, M.M. Abd El-Rahman, B.G. de la Torre, A. El-Faham, F. Albericio, 1,3,5-triazino peptide derivatives: synthesis, characterization, and preliminary antileishmanial activity, *ChemMedChem* 13 (2018) 725–735.
- [13] Sh.N. Khattab, H.H. Khalil, A.A. Bekhit, M.M. Abd El-Rahman, A. El-Faham, F. Albericio, Synthesis and preliminary biological evaluation of 1,3,5-triazine amino acid derivatives to study their MAO inhibitors, *Molecules* 20 (2015) 15976–15988.
- [14] Sh.N. Khattab, S.E. Abdel Naim, M. El Sayed, A.A. El Bardan, A.O. Elzoghby, A.A. Bekhit, A. El-Faham, Design and synthesis of new *s*-triazine polymers and their application as nanoparticulate drug delivery systems, *New J. Chem.* 40 (2016) 9565–9578.
- [15] L. Ballell, R.A. Field, K. Duncan, R.J. Young, New small-molecule synthetic antimycobacterials, *Antimicrob. Agents Chemother.* 49 (2005) 2153–2163.
- [16] D.P. Walsh, Y.T. Chang, Chemical genetics, *Chem. Rev.* 106 (2006) 2476–2530.
- [17] G. Giacomelli, A. Porcheddu, L. De Luca, [1,3,5]-Triazine: a versatile heterocycle in current applications of organic chemistry, *Curr. Org. Chem.* 8 (2004) 1497–1519.
- [18] J.L. Silen, A.T. Lu, D.W. Solas, M.A. Gore, D. Maclean, N.H. Shah, J.M. Coffin, N.S. Bhinderwala, Y. Wang, K.T. Tsutsui, G.C. Look, D.A. Campbell, R.L. Hale, M. Navre, C.R. Deluca-Flaherty, Screening for novel antimicrobials from encoded combinatorial libraries by using a two-dimensional agar format, *Antimicrob. Agents Chemother.* 42 (1998) 1447–1453.
- [19] V.R. Avupati, R.P. Yejella, V.R. Parala, K.N. Killari, V.M.R. Papasani, P. Cheepurupalli, B. Gavalapu, V.R. Boddada, Synthesis, characterization and in vitro biological evaluation of some novel 1,3,5-triazine-Schiff base conjugates as potential antimycobacterial agents, *Bioorg. Med. Chem. Lett.* 23 (2013) 5968–5970.
- [20] S.N. Gavade, V.L. Markad, K.M. Kodam, M.S. Shingare, D.V. Mane, Synthesis and biological evaluation of novel 2,4,6-triazine derivatives as antimicrobial agents, *Bioorg. Med. Chem. Lett.* 22 (2012) 5075–5077.
- [21] C. Courme, N. Gresh, M. Vidal, C. Lenoir, C. Garbay, J.-C. Florent, E. Bertounesque, Synthesis of aryl phosphates based on pyrimidine and triazine scaffolds, *Eur. J. Med. Chem.* 45 (2010) 244–255.
- [22] P. Gahtori, S.K. Ghosh, B. Sing, U.P. Singh, H.R. Bhat, A. Uppal, Synthesis, SAR and antibacterial activity of hybrid chloro, dichloro-phenylthiazolyl-*s*-triazines, *Saudi Pharm. J.* 20 (2012) 35–43.
- [23] N.C. Desai, A.H. Makwana, K.M. Rajpara, Synthesis and study of 1,3,5-triazine based thiazole derivatives as antimicrobial agents, *J. Saudi Chem. Soc.* 20 (2016) S334–S341.
- [24] A. Jarrahpour, D. Khalili, E. De Clercq, C. Salmi, J.M. Brunel, Synthesis, antibacterial, antifungal and antiviral activity evaluation of some new bis-schiff bases of isatin and their derivatives, *Molecules* 12 (2007) 1720–1730.
- [25] P. Panneerselvam, B.A. Rather, D.R.S. Reddy, N.R. Kumar, Synthesis and antimicrobial screening of some Schiff bases of 3-amino-6,8-dibromo-2-phenylquinazolin-4(3*H*)-ones, *Eur. J. Med. Chem.* 44 (2009) 2328–2333.
- [26] K. Cheng, Q.-Z. Zheng, Y. Qian, L. Shi, J. Zhao, H.-L. Zhu, Synthesis, antibacterial activities and molecular docking studies of peptide and Schiff bases as targeted antibiotics, *Bioorg. Med. Chem.* 17 (2009) 7861–7871.
- [27] P.K. Panchal, P.B. Pansuriya, M.N. Patel, In-vitro biological evaluation of some ONS and NS donor Schiff's bases and their metal complexes, *J. Enz. Inhib. Med. Chem.* 21 (2006) 453–458.
- [28] K.V. Sashidhara, J.N. Rosaiah, G. Bhatia, J.K. Saxena, Novel keto-enamine Schiff's bases from 7-hydroxy-4-methyl-2-oxo-2*H*-benzo[*h*]chromene-8,10-dicarbaldehyde as potential antidyslipidemic and antioxidant agents, *Eur. J. Med. Chem.* 43 (2008) 2592–2596.
- [29] J. Vanco, O. Svajlenova, E. Ramanska, J. Muselik, J. Valentova, Antiradical activity of different copper(II) Schiff base complexes and their effect on alloxan-induced diabetes, *Trace Elem. Med. Biol.* 18 (2004) 155–161.
- [30] D. Sinha, A.K. Tiwari, S. Singh, G. Shukla, P. Mishra, H. Chandra, A.K. Mishra, Synthesis, characterization and biological activity of Schiff base analogues of indole-3-carboxaldehyde, *Eur. J. Med. Chem.* 43 (2008) 160–165.
- [31] G. Nasr, E. Petit, C.T. Supuran, Y. Winum, M. Barboiu, Carbonic anhydrase II-induced selection of inhibitors from a dynamic combinatorial library of Schiff's bases, *Bioorg. Med. Chem. Lett.* 19 (2009) 6014–6017.
- [32] D.R. Ramadan, A.A. Elbardan, A.A. Bekhit, A. El-Faham, Sh.N. Khattab, Synthesis and characterization of novel dimeric *s*-triazine derivatives as potential antibacterial agents against MDR clinical isolates, *New J. Chem.* 42 (2018) 10676–10688.
- [33] K.M. Al-Zaydi, H.H. Khalil, A. El-Faham, Sh.N. Khattab, Synthesis, characterization and evaluation of 1,3,5-triazine aminobenzoic acid derivatives for their antimicrobial activity, *Chem. Cent. J.* 11 (2017) 39–50.
- [34] J.I. Manshester, D.D. Dussault, J.A. Rose, P.A. Boriack-Sjodin, M. Uria-Nickelsen, G. Ioannidis, S. Bist, P. Fleming, K.G. Hull, Discovery of a novel azaindole class of antibacterial agents targeting the ATPase domains of DNA gyrase and topoisomerase IV, *Bioorg. Med. Chem. Lett.* 22 (2012) 5150–5156.
- [35] Z. Jakopin, J. Ilas, M. Barancokova, M. Brvar, P. Tammela, M.S. Dolenc, T. Tomasic, D. Kikelj, Discovery of substituted oxadiazoles as a novel scaffold for DNA gyrase inhibitors, *Eur. J. Med. Chem.* 130 (2017) 171–184.
- [36] S.G. Komurcu, S. Rollas, M. Ulgen, J.W. Gorrod, A. Cevikbas, Evaluation of some arylhydrazones of *p*-aminobenzoic acid hydrazide as antimicrobial agents and their in vitro hepatic microsomal metabolism, *Boll. Chim. Farm.* 134 (7) (1995) 375–379.
- [37] M. Neelgundmath, O. Kotresh, Synthesis, evaluation and characterization of some 1,3,4-triazole-2-one derivatives as antimicrobial agents, *E-J. Chem.* 9 (4) (2011) 2407–2414.
- [38] T. Mosmann, Rapid colorimetric assay for cellular growth and survival: application to proliferation and cytotoxicity assays, *J. Immunol. Methods* 65 (1983) 55–63.
- [39] R.J. Case, S.G. Franzblau, Y. Wang, S.H. Cho, D.D. Soejarto, G.F. Pauli, Ethnopharmacological evaluation of the informant consensus model on anti-tuberculosis claims among the Manus, *J. Ethnopharmacol.* 106 (2006) 82–89.
- [40] R. Hanachi, S. Belaidi, A. Kerassa, S. Boughdiri, Structure activity/property relationships of pyrazole derivatives by MPO and QSAR methods for drug design, *Res. J. Pharm., Biol. Chem.* 6 (2015) 923–935.
- [41] C.G. De-Almeida, G.D. Garbois, L.M. Amaral, C.C. Diniz, M. Hayaric, Relationship between structure and antibacterial activity of lipophilic *N*-acyldiamines, *Biomed. Pharmacother.* 64 (2010) 287–290.
- [42] A. Daina, O. Michielin, V. Zoete, SwissADME: a free web tool to evaluate pharmacokinetics, drug-likeness and medicinal chemistry friendliness of small molecules, *Sci. Rep.* 7 (2017) 42717.
- [43] <https://www.rcsb.org/>.
- [44] C.C.G.U. Molecular Operating Environment (MOE), 1010 Sherbooke St. West, Suite #910, Montreal, QC, Canada, H3A 2R7, 2016.
- [45] S. Durdagi, M.T. Qamar, R.E. Salmas, Q. Tariq, F. Anwar, U.A. Ashfaq, Investigating the molecular mechanism of staphylococcal DNA gyrase inhibitors: a combined ligand-based and structure-based resources pipeline, *J. Mol. Graph. Model.* 85 (2018) 122–129.
- [46] P.M. Hawkey, D.A. Lewis, *Medical Bacteriology: A Practical Approach*, Oxford University Press, USA, 1994, pp. 181–194.
- [47] P.R. Murray, E.J. Baron, M.A. Pfaller, F.C. Tenover, R.H. Tenover, *Manual of clinical microbiology, Antimicrobial Agents and Susceptibility Testing*, American Society for Microbiology, Washington, DC, 1995, pp. 73–97.