



Synthesis, antioxidant and A β anti-aggregation properties of new ferulic, caffeic and lipoic acid derivatives obtained by the Ugi four-component reaction

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ABSTRACT

We report herein the synthesis antioxidant and A β anti-aggregation capacity of (*E*)-*N*-benzyl-*N*-[2-(benzylamino)-2-oxoethyl]-3-(aryl)acrylamides and related (*R*)-*N*-benzyl-*N*-(2-(benzylamino)-2-oxoethyl)-5-(1,2-dithiolan-3-yl)pentanamides 1–12. These compounds have been obtained, via Ugi four-component reaction, from modest to good yields. Their antioxidant analysis, using the DPPH and ORAC assays, allowed us to identify compounds 8 and 9, as potent antioxidant agents, showing also strong A β_{1-40} self-aggregation inhibition, two biological properties of interest in pathologies linked to the oxidative stress, such as Alzheimer's disease.

1. Introduction

Over the last decade, we have been actively working in the design, synthesis and pharmacological evaluation of diversely functionalized pyridine [1] and quinolone-substituted derivatives [2–4], as multitarget small molecules (MTSM) [5,6] for the potential treatment of Alzheimer's disease (AD) [7], using in some cases selected multicomponent reactions (MCR) [8–10], as suitable synthetic protocols.

Among the many neuropathological AD hallmarks [e.g. cholinergic deficit, intracellular neurofibrillary tangles and amyloid- β (A β) deposits], oxidative stress plays a fundamental role in the biological events leading to this neurodegenerative disease [11]. For instance, the histopathological data obtained from extracellular A β deposits have revealed that the release of reactive oxygen species (ROS) can heavily damage the mitochondria and other cellular contents of the neurons [12,13]. Taking into account these observations, the search for new and more efficient antioxidants for the therapy of AD has been very intense over the last years [14].

From a synthetic angle, MCRs have been now established as powerful synthetic tools for creating molecular complexity and diversity-oriented synthesis of various heterocycles [15–18]. Among the known MCRs, the isocyanide-based Ugi four-component reaction (U-4CR) [19] allows the condensation, in one-step, of isocyanides, aldehydes, primary amines and carboxylic acids (Scheme 1). Recently, D'Arrigo and colleagues have reported the multicomponent synthesis of polyphenol-based α -acylaminocarboxamides mimicking natural polyphenols [e.g. quercetin, resveratrol, (–)-epigallocatechin gallate] that were able to inhibit A β oligomerization [20]. The polyphenolic nature of these adducts allows the disruption of protein-protein interactions, thus hampering the aggregation of A β fibrils. One can also foresee that this set of polyphenol-based Ugi adducts could also have antioxidant activity, even if no data regarding this matter have been reported.

In this context, we have recently embarked in a project targeted to the synthesis of (*E*)-*N*-benzyl-*N*-[2-(benzylamino)-2-oxoethyl]-3-(aryl)acrylamides of type I bearing phenyl, pyridine and quinoline motifs (Scheme 1), assembled by U-4CR [19], where we have incorporated

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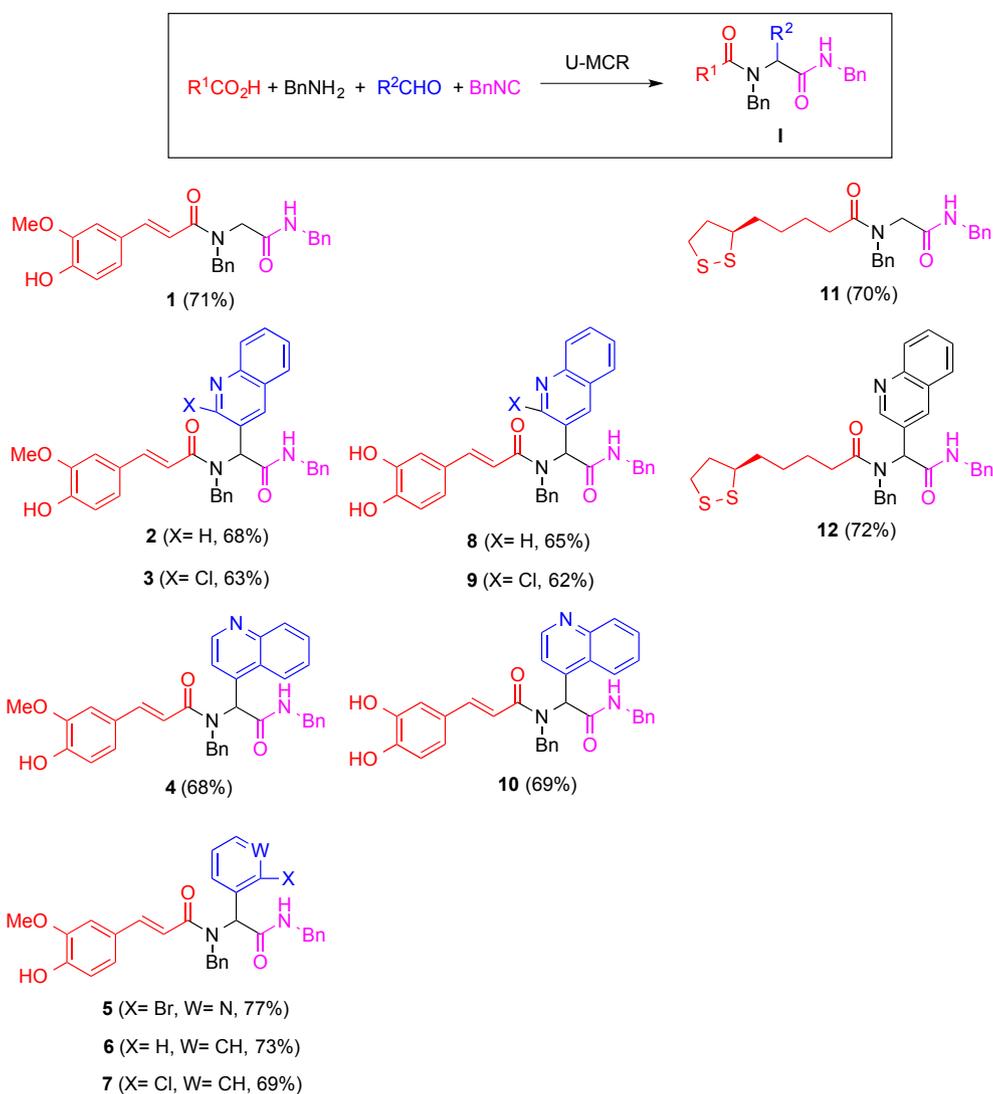
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Scheme 1. Structure of compounds of type I, prepared by U-4CR, and structure of molecules 1–12.

typical antioxidant motifs from the corresponding naturally-occurring ferulic (FA), caffeic (CA) and lipoic (LA) acids, that have been shown to be effective antioxidant agents [21,22]. In addition, we have incorporated halogens in our design due to their capacity to interact with a number of biological targets as hydrophobic moieties, to form halogen bonds with suitable functional motifs, [23,24] or increase their antioxidant activity [25], hypothesizing that the new synthetic derivatives would enhance the *in vitro* antioxidant effect compared to the parent acids.

Thus, in this report we describe the synthesis of adducts 1–12 (Scheme 1), and their antioxidant properties using the DPPH and ORAC assays. As a result, we have identified adducts 8 and 9 as potent antioxidant agents showing in addition strong $A\beta_{1-40}$ self-aggregation inhibition.

2. Results and discussion

2.1. Synthesis

As shown in scheme 1, product 1 was obtained from FA, formaldehyde, benzyl isocyanide, and benzylamine. The use of FA/benzyl isocyanide/benzylamine afforded molecules 2–7, when using quinoline-3-carbaldehyde, 2-chloroquinoline-3-carbaldehyde, quinoline-4-carbaldehyde, 2-bromonicotinaldehyde, benzaldehyde, and 2-

chlorobenzaldehyde, respectively. Similarly, the U-4CR of CA/benzyl isocyanide/benzylamine afforded the molecules 8–10, when quinoline-3-carbaldehyde, 2-chloroquinoline-3-carbaldehyde, and quinoline-4-carbaldehyde, respectively, have been reacted. Finally, selected and related analogues to adducts 1 and 8, such as 11 and 12 have been prepared by U-4CR between (*R*)-LA, formaldehyde (or quinoline-3-carbaldehyde), benzyl isocyanide, and benzylamine.

As described below and following the general protocol for the U-4CR, by stirring the solution of reagents in methanol, at room temperature (rt) for 24 h, the Ugi adducts were easily isolated by recrystallization from suitable solvents or flash column chromatography. All new compounds gave analytical and spectroscopic data in good agreement with their structures (see Section 5).

2.2. Antioxidant analysis

With pure compounds 1–12 in hand, we next tested their antioxidant capacity using the 2,2-diphenyl-1-picrylhydrazyl radical (DPPH) [26] and the Oxygen Radical Capacity (ORAC) [27] methods. These assays are well-established techniques for determining the radical scavenging activity (RSA) of a compound. ROS, such as the hydroxyl or superoxide radicals, cause lipid peroxidation, protein oxidation and DNA damage which can lead to organelles malfunction and further cellular disorders.

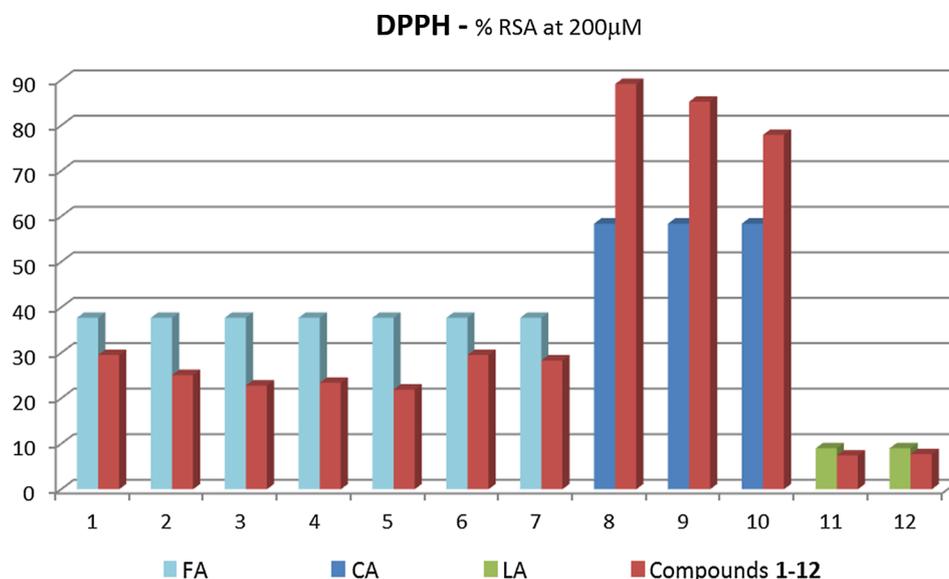


Fig. 1. DPPH (% RSA) at 200 μ M of compounds 1–12.

The antioxidant power of compounds 1–12 was then evaluated using the DPPH test, using FA, CA and LA as positive controls. As shown in Fig. 1, at 200 μ M dose, the investigated compounds showed a RSA percentage ranging from 7.4 for 11 to 89.1 for 8. Compounds 1–7 were more active than LA (9.1%), but still less active than FA and CA. Molecules 8–10, bearing a CA motif, showed a RSA percentage higher than 77%, being 8 the most active compound, 2.3 and 1.5-fold more active than FA and CA, respectively. Regarding structure-activity relationship (SAR), the most potent antioxidant compounds were those bearing CA followed by adducts bearing a feruloyl motif and, finally, those bearing a lipoyl moiety. For FA compound's, it seems that those bearing phenyl (6), 2-chlorophenyl (7) or without substituent (1) are better antioxidant than their analogues 2–5 while in the caffeic group the quinoline attached by its C3 (8) give the best effect and was thus the most antioxidant.

The ORAC test afforded the results shown in Fig. 2, expressed as Trolox equivalents (TE), in relation to radical scavenging properties of Trolox. Compounds 1–7 bearing a feruloyl motif showed good radical scavenging properties with ORAC values, ranging from 2.19 TE for compound 6 to 3.48 TE for product 7, but remaining less active than

FA. However, for caffeoyl-containing molecules 8–10, the ORAC values were equal to 5.48 TE, 6.86 TE and 4.98 TE, respectively, being more active than FA (3.7 TE). Lipoyl-containing compounds 11 and 12 gave low ORAC values as similarly observed in the DPPH assay. Regarding the SAR, the same trend observed in the DPPH test was found in the ORAC analysis, as the CA compounds are better antioxidants than those with FA and LA. From the FA group, compound 7, bearing 2-chlorobenzene motif, favourably compared to the other six ones, which did not show meaningful difference activity, regardless of the substituent incorporated. For the CA substituted quinolines 8–10 the presence of the chlorine atom at C2 in compound 9 is key to have a higher ORAC antioxidant effect.

2.3. Inhibition of $A\beta_{1-40}$ aggregation by compounds 7–10

The four best antioxidants 7–10 were then evaluated as inhibitors of self $A\beta_{1-40}$ aggregation at 20 μ M, curcumin being used as positive control. As shown in Fig. 3, the feruloyl-containing compound 7 has a very low activity as inhibitor of $A\beta_{1-40}$ self-aggregation, but the three molecules bearing a caffeoyl motif 8–10 displayed a strong inhibition

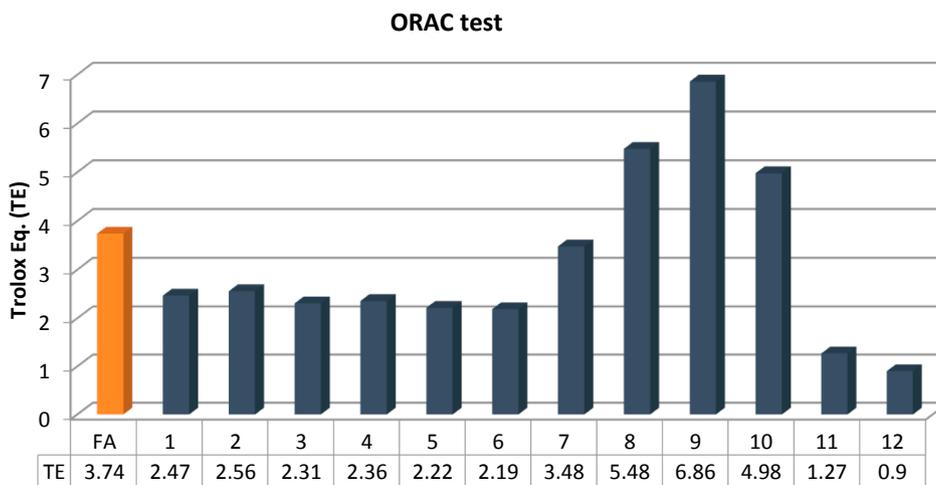
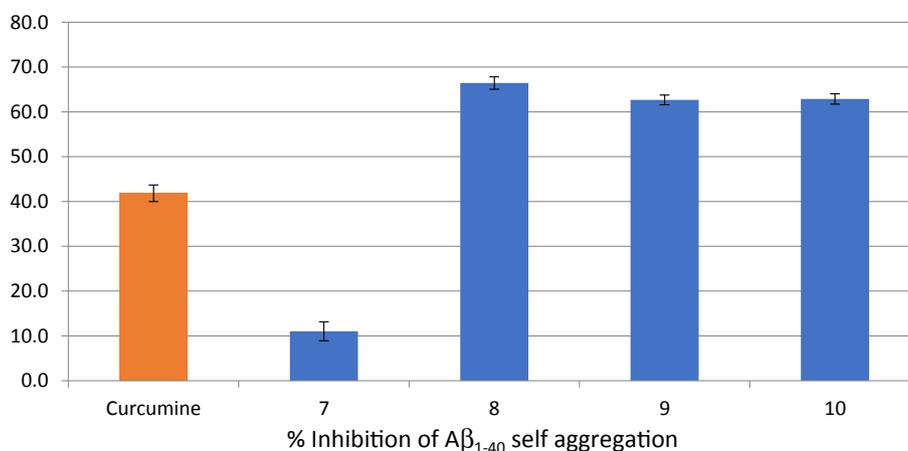


Fig. 2. ORAC values of compounds 1–12, data are expressed as TE and are the mean ($n = 3$) \pm SD.



The Thioflavin-T fluorescence method was used. The values are expressed as the mean \pm SD of at least three independent measurements. All values were obtained at a compound concentration of 20 μ M.

Fig. 3. Inhibition of A β_{1-40} self-aggregation (%) by compounds 7–10. The Thioflavin-T fluorescence method was used. The values are expressed as the mean \pm SD of at least three independent measurements. All values were obtained at a compound concentration of 20 μ M.

activity, showing values equal to 66.4, 62.7 and 62.9%, respectively, better than the one shown by curcumine (41.8%).

2.4. Neuroprotective activity of compounds 8–10 on H₂O₂ (150 μ M)-induced cell death in SH-SY5Y cells

The ability to prevent the human neuroblastoma cell line SH-SY5Y cell death induced by hydrogen peroxide (H₂O₂) for the generation of exogenous free radicals, was then assayed to determine the neuroprotective capacity of the three most balanced antioxidants and A β_{1-40} self aggregation inhibitors, compounds 8–10, at non-toxic concentration 10 μ M, against SH-SY5Y. As shown in table 1, only compound 9 showed a significant neuroprotection around 25% against H₂O₂ insult at 10 μ M.

3. Conclusion

In this work we have reported for the first time the U-4CR synthesis of (*E*)-*N*-benzyl-*N*-[2-(benzylamino)-2-oxoethyl]-3-(aryl)acrylamides and (*R*)-*N*-benzyl-*N*-(2-(benzylamino)-2-oxoethyl)-5-(1,2-dithiolan-3-yl)pentanamides 1–12, from modest to good yields.

Their antioxidant analysis allowed us to identify (*E*)-*N*-benzyl-*N*-(2-(benzylamino)-2-oxo-1-(quinolin-3-yl)ethyl)-3-(3,4-dihydroxy phenyl)acrylamide (8) and (*E*)-*N*-benzyl-*N*-(2-(benzylamino)-1-(2-chloroquinolin-3-yl)-2-oxoethyl)-3-(3,4-dihydroxyphenyl)acrylamide (9) as potent antioxidant agents, in the DPPH and ORAC assays. In fact, compound 8 was 1.5-fold more active than CA, and compound 9 was 1.8-fold more active than FA. In addition, compounds 8 and 9 also

Table 1

Neuroprotective activity of compounds 8–10 on H₂O₂ (at 150 μ M)-induced cell death in SH-SY5Y cells.

Compounds	Dose	H ₂ O ₂ (%)
8	10 μ M	np
9	10 μ M	21.87 \pm 0.01*
10	10 μ M	np

Data are expressed as % neuroprotection \pm SEM of quadruplicates from at least three different cultures.

np: not protective.

* $p < 0.05$, as compared to the control cultures (one-way ANOVA).

showed an interesting inhibitory activity of A β_{1-40} self-aggregation, being 1.6 and 1.5-fold more active than curcumine, at 20 μ M. Finally, compound 9 showed a significant neuroprotection effect against H₂O₂ induced cell death in SH-SY5Y cells at 10 μ M. Consequently, we conclude that these new Ugi adducts deserve further investigation for the treatment of pathologies related to oxidative stress.

4. Experimental part

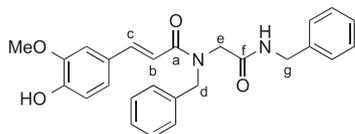
4.1. General methods

The starting materials were purchased from Sigma-Aldrich Chimie S.a.r.L (Saint-Quentin-Fallavier, France) or TCI Europe N.V. (Zwijndrecht, Belgium). Proton and Carbon JMOD NMR spectra were recorded on a Bruker AC 300 spectrometer (Bruker Biospin). The chemical shifts are expressed in parts per million (ppm) and the multiplicities are indicated by the following abbreviations: s, singlet; d, doublet; dd, double doublet; t, triplet; q, quadruplet; p, pentuplet; m, multiplet, and coupling constants are expressed in Hertz (Hz). Melting points (mp) were obtained on a Köfler apparatus and were not corrected. All reactions were monitored by Thin Layer Chromatography (TLC) using Alugram Sil G/UV₂₅₄ plates and visualized by UV light. Flash Column chromatography was carried out using silica gel 60 (70–230 mesh, Macherey-Nagel). Elemental analyses were performed on Thermo-Finnigan Flash EA 1112.

4.2. General procedure for the synthesis of Ugi adducts 1–12

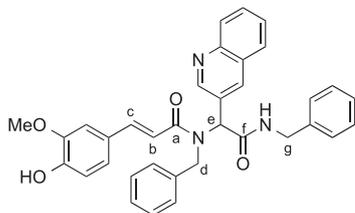
A solution of the benzylamine (1 equiv) and the corresponding aldehyde (1 equiv) in MeOH (1 M) was stirred for 15 min at room temperature. Ferulic, caffeic or (*R*)-lipoic acid (1 equiv) and benzyl isocyanide (1 equiv) were then added. The reaction mixture was stirred 24 h at 50 °C. The crude product was either purified by flash column chromatography or filtrated to afford the corresponding Ugi adduct. Regarding the ¹H and ¹³C spectra of compounds 1–12, at 298 K, some Ugi adducts appear as a mixture of invertomers. When each peak corresponding to the same proton of each invertomer can be identified, they are listed together.

4.2.1. (*E*)-*N*-Benzyl-*N*-(2-(benzylamino)-2-oxoethyl)-3-(4-hydroxy-3-methoxyphenyl)acrylamide (**1**)



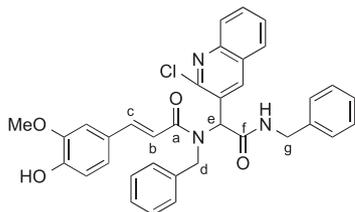
Following the general procedure, a solution of benzylamine (110 μ L, 1 mmol), paraformaldehyde (30 mg, 1 mmol), ferulic acid (194 mg, 1 mmol) and benzyl isocyanide (120 μ L, 1 mmol) in MeOH (1 mL), gave, after flash column chromatography [cyclohexane/AcOEt (7:3, v/v)], adduct **1** (305 mg, 71%) as a white foam: mp 163–4 °C; IR (ATR) ν 3258, 3067, 3031, 1676, 1643, 1582, 1509 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 7.70 (d, $J = 15.0$ Hz, 1H, H-c), 7.36–7.09 (m, 10H), 7.03–7.00 (m, 2H), 6.92–6.86 (m, 2H), 6.69 and 6.54 (d, $J = 15.0$ Hz, 1H, H-b), 6.06 (bs, 1H, NH), 4.80 and 4.65 (s, 2H, H-d), 4.44 and 4.31 (s, 2H, H-g), 4.12 (s, 2H, H-e), 3.86 (s, 3H, OMe); ^{13}C NMR (CDCl_3 , 75 MHz) δ 169.3 (C-f), 168.5 (C-a), 148.0, 147.0, 145.0 (C-c), 136.4, 129.2, 128.8, 128.2, 127.7, 127.5, 126.9, 122.6, 115.0 (C-b), 113.8, 110.2, 56.2 (OMe), 52.8 (C-e), 51.5 (C-d), 43.6 (C-g). Anal. Calcd. for $\text{C}_{26}\text{H}_{26}\text{N}_2\text{O}_4$: C, 72.54; H, 6.09; N, 6.51. Found: C, 72.15; H, 5.97; N, 6.66.

4.2.2. (*E*)-*N*-Benzyl-*N*-(2-(benzylamino)-2-oxo-1-(quinolin-3-yl)ethyl)-3-(4-hydroxy-3-methoxyphenyl)acrylamide (**2**)



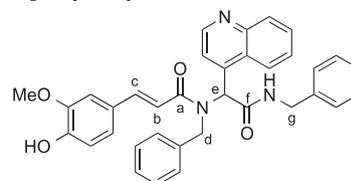
Following the general procedure, a solution of benzylamine (110 μ L, 1 mmol), 3-quinoline carboxaldehyde (157 mg, 1 mmol), ferulic acid (194 mg, 1 mmol) and benzyl isocyanide (120 μ L, 1 mmol) in MeOH (1 mL), gave, after recrystallization with Et_2O with few drops of MeOH, adduct **2** (317 mg, 68%) as a white solid: mp 183–4 °C; IR (ATR) ν 3260, 3058, 1646 1576, 1512 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 8.79 (d, $J = 2.2$ Hz, 1H), 8.29 (bs, 1H, OH), 8.0 (d, $J = 8.0$ Hz, 1H), 7.73–7.66 (m, 3H + H-c), 7.52 (t, $J = 7.0$ Hz, 1H), 7.28–7.26 (m, 5H), 7.12–7.03 (m, 5H + H-b), 6.94 (dd, $J = 1.5$ Hz, $J = 8.0$ Hz, 1H), 6.84 (d, $J = 8.0$ Hz, 1H), 6.77 (d, $J = 1.5$ Hz, 1H), 6.59 (d, $J = 15.4$ Hz, 1H, H-b), 6.34 (bs, 1H, NH), 5.99 (s, 1H, H-e), 4.92 (d, $J = 17.4$ Hz, 1H, H-d), 4.77 (d, $J = 17.4$ Hz, 1H, H-d), 4.57 (dd, $J = 14.1$ Hz, $J = 5.4$ Hz, 1H, H-g), 4.47 (dd, $J = 14.1$ Hz, $J = 5.4$ Hz, 1H, H-g), 3.81 (s, 3H, OMe); ^{13}C NMR (CDCl_3 , 75 MHz) δ 169.2 (C-f), 169.0 (C-a), 151.6, 148.1, 147.8, 147.0, 145.1 (C-c), 137.8, 137.0, 130.2, 129.4, 129.0, 128.9, 128.3, 128.0, 127.8, 127.6, 127.2, 126.7, 122.9, 115.3, 115.0 (C-b), 110.1, 61.0 (C-e), 56.2 (OMe), 50.5 (C-d), 44.1 (C-g). Anal. Calcd. for $\text{C}_{35}\text{H}_{31}\text{N}_3\text{O}_4$: C, 75.39; H, 5.60; N, 7.54. Found: C, 75.20; H, 5.54; N, 7.41.

4.2.3. (*E*)-*N*-Benzyl-*N*-(2-(benzylamino)-2-oxo-1-(2-chloroquinolin-3-yl)ethyl)-3-(4-hydroxy-3-methoxyphenyl)acrylamide (**3**)



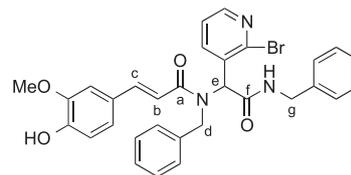
Following the general procedure, a solution of benzylamine (110 μ L, 1 mmol), 2-chloro-3-quinolinecarboxaldehyde (191 mg, 1 mmol), ferulic acid (194 mg, 1 mmol) and benzyl isocyanide (120 μ L, 1 mmol) in MeOH, gave after recrystallization with $\text{Et}_2\text{O}/\text{CH}_2\text{Cl}_2$ (8:2, v/v), adduct **3** (352 mg, 63%) as a white solid: mp 192–4 °C; IR (ATR) ν 3224, 3065, 1658, 1644, 1574, 1515, 1405 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 8.56 (s, 1H, OH), 7.86 (d, $J = 9.0$ Hz, 1H), 7.74–7.68 (m, 3H + H-c), 7.54 (t, $J = 7.0$ Hz, 1H), 7.26 (m, 5H), 7.03 (m, 5H + H-b), 6.95–6.90 (m, 2H), 6.85–6.79 (m, 2H), 6.67–6.62 (m, 2H + NH), 5.82 (s, 1H), 4.94 (d, $J = 17.1$ Hz, 1H, H-d), 4.77 (d, $J = 17.1$ Hz, 1H, H-d), 4.60–4.54 (dd, $J = 17.1$, $J = 6.3$ Hz Hz, 1H, H-g), 4.46–4.40 (dd, $J = 17.1$ Hz, $J = 6.3$ Hz, H-g), 3.83 (s, 3H, OMe); ^{13}C NMR (CDCl_3 , 75 MHz): $\delta = 169.2$ (C-f), 168.9 (C-a), 148.1, 147.5, 147.0, 145.2, 137.5, 137.3, 131.3, 129.0, 128.8, 128.4, 128.2, 128.0, 127.9, 127.8, 127.6, 127.6, 127.5, 127.2, 127.0, 126.78, 126.7, 126.7, 122.9, 115.3, 115.0 (C-b), 110.2, carbon signal of C-e missing, 56.2 (OMe), 50.2 (C-d), 44.2 (C-g). Anal. Calcd. for $\text{C}_{35}\text{H}_{30}\text{ClN}_3\text{O}_4$: C, 71.00; H, 5.11; N, 7.10. Found: C, 70.84; H, 5.02; N, 7.21.

4.2.4. (*E*)-*N*-Benzyl-*N*-(2-(benzylamino)-2-oxo-1-(quinolin-4-yl)ethyl)-3-(4-hydroxy-3-methoxyphenyl)acrylamide (**4**)



Following the general procedure, a solution of benzylamine (110 μ L, 1 mmol), 4-quinolinecarboxaldehyde (157 mg, 1 mmol), ferulic acid (194 mg, 1 mmol) and benzyl isocyanide (120 μ L, 1 mmol) in MeOH, gave, after recrystallization in AcOEt and *n*-hexane, adduct **4** (383 mg, 68%) as a white solid: mp 155–7 °C; IR (ATR) ν 3450, 3242, 3070, 1673, 1636, 1554, 1509, 1422 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 8.77 (d, $J = 4.4$ Hz, 1H), 8.02–7.97 (m, 2H), 7.75 (d, $J = 15.0$ Hz, H-c), 7.69–7.64 (m, 2H), 7.57–7.52 (t, $J = 7.5$ Hz, 1H), 7.32–7.19 (m, 6H), 6.94–6.82 (m, 5H), 6.75 (s, 1H), 6.70–6.68 (m, 2H), 6.59–6.54 (d, $J = 15.0$ Hz, 1H), 6.44–6.41 (t, $J = 6.0$ Hz, 1H, NH), 5.97 (bs, 1H, OH), 4.94 (d, $J = 18.1$ Hz, 1H, H-d), 4.77 (d, $J = 18.1$ Hz, 1H, H-d), 4.64–4.57 (dd, $J = 15.0$ Hz, $J = 6.0$ Hz, 1H, H-g), 4.48–4.41 (dd, $J = 15.0$ Hz, $J = 6.0$ Hz, 1H, H-g), 3.80 (s, 3H, OMe); ^{13}C NMR (CDCl_3 , 75 MHz) δ 169.3 (C-f), 168.7 (C-a), 150.0, 148.6, 148.1, 147.0, 145.1 (C-c), 140.7, 138.0, 137.5, 130.7, 130.0, 129.0, 128.3, 128.0, 127.9, 127.8, 127.6, 127.1, 126.4, 126.2, 123.6, 123.0, 122.2, 115.2, 115.0 (C-b), 109.9, carbon signal of C-e missing, 56.2 (OMe), 49.9 (C-d), 44.2 (C-g). Anal. Calcd. for $\text{C}_{35}\text{H}_{31}\text{N}_3\text{O}_4$: C, 75.39; H, 5.60; N, 7.54. Found: C, 75.35; H, 5.64; N, 7.46.

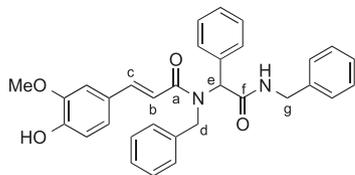
4.2.5. (*E*)-*N*-Benzyl-*N*-(2-(benzylamino)-1-(2-chlorophenyl)-2-oxoethyl)-3-(4-hydroxy-3-methoxyphenyl)acrylamide (**5**)



Following the general procedure, a solution of benzylamine (110 μ L, 1 mmol), 2-bromonicotinaldehyde (186 mg, 1 mmol), ferulic acid (194 mg, 1 mmol) and benzyl isocyanide (120 μ L, 1 mmol) in MeOH (1 mL), gave a solid that after washing with methanol (10 mL) and *n*-hexane (2 \times 20 mL), afforded adduct **5** (451 mg, 77%) as a white solid:

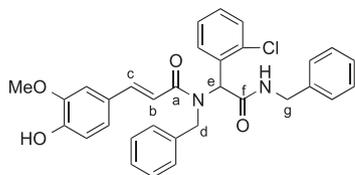
mp 169–171 °C; IR (ATR) ν 3220, 3045, 1667 1530, 1519 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 8.13–8.02 (m, 2H), 7.68 (d, $J = 15.0$ Hz, 1H, H-c), 7.26–7.11 (m, 11H), 6.93 (m, 2H), 6.84–6.79 (m, 2H), 6.61 (d, $J = 15.0$ Hz, 1H, H-b), 6.40 (bs, 1H, OH), 6.20 (bs, 1H, NH), 4.91 (d, $J = 17.0$ Hz, 1H, H-d), 4.72 (d, $J = 17.0$ Hz, 1H, H-d), 4.53–4.47 (m, 1H, H-g), 4.37–4.32 (m, 1H, H-g), 3.80 (s, 3H, OMe); ^{13}C NMR (CDCl_3 , 75 MHz) δ signal of C-f missing, 168.7 (C-a), 149.8, 148.0, 146.9, 145.9, 145.2 (C-c), 139.9, 137.8, 137.2, 132.4, 128.8, 128.7, 127.9, 127.6, 126.5, 122.8, 115.0, 114.8 (C-b), 110.1, 61.5 (C-e), 56.1 (OMe), 50.3 (C-d), 44.0 (C-g). Anal. Calcd. for $\text{C}_{31}\text{H}_{28}\text{BrN}_3\text{O}_4$: C, 63.49; H, 4.81; N, 7.16. Found: C, 63.20; H, 4.88; N, 7.03.

4.2.6. (*E*)-*N*-benzyl-*N*-(2-(benzylamino)-2-oxo-1-phenylethyl)-3-(4-hydroxy-3-methoxy phenyl)acrylamide (**6**)



Following the general procedure, a solution of benzylamine (244 μL , 1 mmol), benzaldehyde (212 mg, 2 mmol), ferulic acid (388 mg, 2 mmol) and benzyl isocyanide (240 μL , 2 mmol) in MeOH (2 mL), gave, after recrystallization from AcOEt and *n*-hexane, adduct **6** (737 mg, 73%) as a white solid: mp 159–160 °C; IR (ATR) ν 3478, 3172, 3029, 1643, 1566, 1515, 1468 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 7.67 (d, $J = 15.0$ Hz, 1H, H-c), 7.41–7.39 (m, 2H), 7.29–7.15 (m, 11H), 7.10–7.07 (m, 2H), 6.91 (d, $J = 7.9$ Hz, 1H), 6.83 (d, $J = 7.9$ Hz, 1H), 6.75 (m, 1H), 6.55 (d, $J = 15.0$ Hz, 1H, H-b), 6.23 (bs, 1H, OH), 6.11 (s, 1H, NH), 5.88–5.83 (m, 1H, H-e), 4.90 (d, $J = 17.1$ Hz, 1H, H-d), 4.68 (d, $J = 17.1$ Hz, 1H, H-d), 4.57–4.50 (dd, $J = 14.5$ Hz, $J = 6.3$ Hz, 1H, H-g), 4.49–4.42 (dd, $J = 14.5$ Hz, $J = 6.3$ Hz, 1H, H-g), 3.81 (s, 3H, OMe); ^{13}C NMR (CDCl_3 , 75 MHz) δ 169.9 (C-f), 168.8 (C-a), 147.8, 146.9, 144.2 (C-c), 138.4, 135.2, 130.0, 128.9 (2C), 128.8, 128.0 (2C), 127.6, 127.3, 122.7, 116.0, 114.9 (C-b), 109.9, 63.5 (C-e), 56.1 (OMe), 50.4 (C-d), 44.0 (C-g). Anal. Calcd. for $\text{C}_{32}\text{H}_{30}\text{N}_2\text{O}_4$: C, 75.87; H, 5.97; N, 5.53. Found: C, 75.61; H, 6.03; N, 5.49.

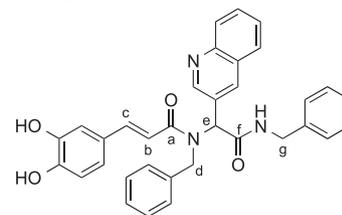
4.2.7. (*E*)-*N*-Benzyl-*N*-(2-(benzylamino)-1-(2-chlorophenyl)-2-oxoethyl)-3-(4-hydroxy-3-methoxyphenyl)acrylamide (**7**)



Following the general procedure, a solution of benzylamine (1.2 mL, 4.9 mmol), 2-chlorobenzaldehyde (689 mg, 4.9 mmol), ferulic acid (952 mg, 4.9 mmol) and benzyl isocyanide (588 μL , 4.9 mmol) in MeOH (5 mL), gave, after washing with methanol (10 mL), adduct **7** (2.65 g, 69%) as a white solid: mp 148–150 °C; IR (ATR) ν 3253, 3065, 1658, 1542, 1511 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 7.75 (d, $J = 15.0$ Hz, 1H, H-c), 7.63 (d, $J = 7.0$ Hz, 1H), 7.27–7.07 (m, 13H), 6.94 (d, $J = 8.0$ Hz, 1H), 6.85–6.79 (m, 2H), 6.60 (d, $J = 15.0$ Hz, 1H, H-b), 6.49 (bs, 1H, OH), 6.34 (m, 1H, NH), 6.08 (s, 1H, H-e), 4.91 (d, $J = 17.7$ Hz, 1H, H-d), 4.68 (d, $J = 17.7$ Hz, 1H, H-d), 4.58–4.51 (dd, $J = 15.4$ Hz, $J = 5.4$ Hz, 1H, H-g), 4.46–4.39 (dd, $J = 15.4$ Hz, $J = 5.4$ Hz, 1H, H-g), 3.82 (s, 3H, OMe); ^{13}C NMR (CDCl_3 , 75 MHz) δ 169.7 (C-f), 168.6 (C-a), 147.8, 146.9, 144.6 (C-c), 137.8, 136.0, 132.9, 131.2, 130.2, 130.0, 128.0, 128.5, 127.9, 127.8, 127.6, 127.2, 126.4, 122.7, 115.4, 114.9 (C-b), 109.4, 59.9 (C-e), 56.1 (OMe), 50.1 (C-d),

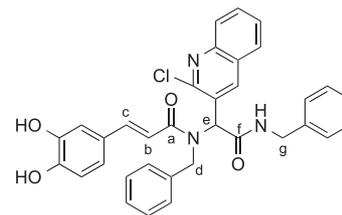
44.0 (C-g). Anal. Calcd. for $\text{C}_{32}\text{H}_{29}\text{ClN}_2\text{O}_4$: C, 71.04; H, 5.40; N, 5.18. Found: C, 71.18; H, 5.32; N, 5.23.

4.2.8. (*E*)-*N*-Benzyl-*N*-(2-(benzylamino)-2-oxo-1-(quinolin-3-yl)ethyl)-3-(3,4-dihydroxy phenyl)acrylamide (**8**)



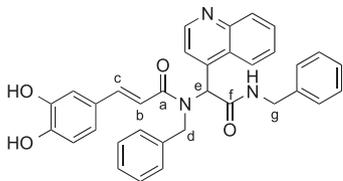
Following the General procedure, a solution of benzylamine (110 μL , 1 mmol), 3-quinolinecarboxaldehyde (157 mg, 1 mmol), caffeic acid (180 mg, 1 mmol) and benzyl isocyanide (120 μL , 1 mmol) in MeOH (1 mL), gave, after recrystallization from $\text{Et}_2\text{O}/\text{CH}_2\text{Cl}_2$ (8:2, v/v), adduct **8** (363 mg, 65%) as a white solid: mp 166–7 °C; IR (ATR) ν 3355, 3022, 1674, 1634, 1499, 1419 cm^{-1} ; ^1H NMR (acetone- d_6 , 300 MHz) δ 8.84 (s, 1H), 8.27–8.22 (m, 3H + 2OH), 8.03 (d, $J = 6.0$ Hz, 1H), 7.93 (d, $J = 8.1$ Hz, 1H), 7.83 (d, $J = 8.1$ Hz, 1H), 7.70 (dd, $J = 7$ Hz, $J = 1.2$ Hz, 1H), 7.63–7.52 (m, 2H, CH_{arom} + Hc), 7.31–7.19 (m, 5H), 7.08–6.99 (m, 6H), 6.89 (d, $J = 8.1$ Hz, 1H), 6.76 (m, 2H, CH_{arom} + Hb), 6.53 (s, 1H, H-e), 5.12 (d, $J = 20.0$ Hz, 1H, H-d), 4.86 (d, $J = 20.0$ Hz, 1H, H-d), 4.52 (dd, $J = 20.0$ Hz, $J = 6.2$ Hz, 1H, H-g), 4.42 (dd, $J = 20.0$ Hz, $J = 6.2$ Hz, 1H, H-g); ^{13}C NMR (acetone- d_6 , 75 MHz) δ 170.1 (C-f), 168.8 (C-a), 153.0, 148.5, 148.3, 146.3, 144.5 (C-c), 140.2, 139.9, 137.4, 137.4, 137.3, 137.3, 130.6, 130.5, 129.9, 129.3, 129.1, 128.7, 128.6, 128.5, 127.9, 127.6, 127.5, 122.3, 116.5, 116.4, 115.2 (C-b), carbon signal of C-e missing, 50.4 (C-d), 44.1 (C-g). Anal. Calcd. for $\text{C}_{34}\text{H}_{29}\text{N}_3\text{O}_4$: C, 75.12; H, 5.38; N, 7.73. Found: C, 75.29; H, 5.43; N, 7.79.

4.2.9. (*E*)-*N*-Benzyl-*N*-(2-(benzylamino)-1-(2-chloroquinolin-3-yl)-2-oxoethyl)-3-(3,4-dihydroxyphenyl)acrylamide (**9**)



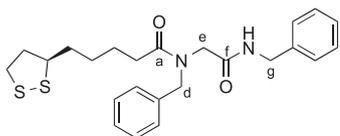
Following the general procedure, a solution of benzylamine (110 μL , 1 mmol), 2-chloro-3-quinolinecarboxaldehyde (191 mg, 1 mmol), caffeic acid (180 mg, 1 mmol) and benzyl isocyanide (120 μL , 1 mmol) in MeOH (1 mL), gave, after washing with methanol (2 mL) Et_2O (10 mL), and *n*-hexane (10 mL), adduct **9** (358 mg, 62%) as a white solid: mp 159–161 °C; IR (ATR) ν 3244, 3030, 1677, 1638, 1564, 1510, 1413 cm^{-1} ; ^1H NMR (acetone- d_6 , 300 MHz) δ 8.35 (s, 1H), 8.22–8.14 (m, 1H), 7.92 (m, 1H), 7.79–7.72 (m, 2H), 7.67–7.59 (m, 2H), 7.32–7.22 (m, 4H), 7.08–6.91 (m, 7H), 6.84–6.79 (m, 3H), 5.13 (d, $J = 19.2$ Hz, 1H), 4.89 (d, $J = 19.2$ Hz, 1H), 4.58–4.50 (dd, $J = 15.2$ Hz, $J = 7.1$ Hz, 1H, H-g), 4.34–4.27 (dd, $J = 15.2$ Hz, 7.1 Hz, 1H, H-g); ^{13}C NMR (acetone- d_6 , 75 MHz) δ 169.8 (C-f), 168.6 (C-a), 152.2, 148.3, 148.0, 146.3, 144.7 (C-c), 140.7, 140.7, 131.7, 129.7, 129.3, 129.0, 128.9, 128.7, 128.7, 128.2, 127.9, 127.4, 127.2, 122.4, 116.5, 116.3, 115.2 (C-b), 60.2 (C-e), 50.4 (C-d), 44.1 (C-g). Anal. Calcd. for $\text{C}_{34}\text{H}_{28}\text{ClN}_3\text{O}_4$: C, 70.64; H, 4.88; N, 7.27. Found: C, 70.69; H, 4.92; N, 7.33

4.2.10. (*E*)-*N*-Benzyl-*N*-(2-(benzylamino)-2-oxo-1-(quinolin-4-yl)ethyl)-3-(3,4-dihydroxy phenyl)acrylamide (**10**)



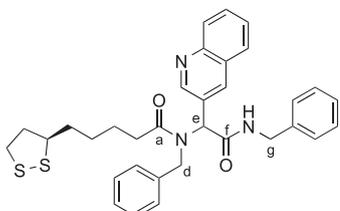
Following the general procedure, a solution of benzylamine (110 μ L, 1 mmol), 4-quinolinecarboxaldehyde (157 mg, 1 mmol), ferulic acid (194 mg, 1 mmol) and benzyl isocyanide (120 μ L, 1 mmol) in MeOH (1 mL), gave, after washing with methanol (2 mL), CH_2Cl_2 (10 mL), and *n*-hexane (10 mL), adduct **10** (280 mg, 69%) as a white solid: mp 187–8 °C; IR (ATR) ν 3494, 3271, 3030, 1694, 1636, 1575, 1508, 1427 cm^{-1} ; ^1H NMR (DMSO- d_6 , 300 MHz) δ 9.38 (bs, 1H, OH), 9.15 (bs, 1H, OH), 8.97 (t, $J = 5.6$ Hz, 1H), 8.73 (d, $J = 4.6$ Hz, 1H), 8.03 (d, $J = 8.3$ Hz, 1H), 7.88 (d, $J = 8.3$ Hz, 1H), 7.72–7.60 (m, 2H), 7.55 (d, $J = 15.0$ Hz, 1H), 7.40 (d, $J = 4.4$ Hz, 1H), 7.35–7.23 (m, 6H), 6.90–6.62 (m, 9H), 5.11 (d, $J = 17.0$ Hz, 1H, H-d), 4.78 (d, $J = 17.0$ Hz, 1H, H-d), 4.39 (dd, $J = 15.1$ Hz, $J = 6.1$ Hz, 1H, H-g), 4.27 (dd, $J = 15.1$ Hz, $J = 6.1$ Hz, 1H, H-g); ^{13}C NMR (DMSO- d_6 , 75 MHz) δ 169.0 (C-f), 167.0 (C-a), 149.6, 147.7, 147.5, 145.3, 143.5 (C-c), 141.89, 138.7, 137.9, 129.4, 129.0, 128.1, 127.3, 127.2, 126.9, 126.7, 126.6, 126.2, 125.9, 125.5, 123.6, 121.4, 120.9, 115.5, 114.3 (C-b), 56.0 (C-e), 48.5 (C-d), 42.4 (C-g). Anal. Calcd. for $\text{C}_{34}\text{H}_{29}\text{N}_3\text{O}_4$: C, 75.12; H, 5.38; N, 7.73. Found: C, 75.05; H, 5.32; N, 7.80.

4.2.11. (*R*)-*N*-Benzyl-*N*-(2-(benzylamino)-2-oxoethyl)-5-(1,2-dithiolan-3-yl)pentanamide (**11**)



Following the general procedure, a solution of benzylamine (110 μ L, 1 mmol), paraformaldehyde (30 mg, 1 mmol), lipoic acid (206 mg, 1 mmol) and benzyl isocyanide (120 μ L, 1 mmol) in MeOH (1 mL), gave, after flash column chromatography [cyclohexane/AcOEt (7:3, v/v)] adduct **11** (310 mg, 70%) as a yellow solid: mp 81–4 °C; IR (ATR): 3296, 3061, 2923, 1639, 1538, 1358 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 7.38–7.14 (m, 10H), 6.68 and 6.60 (t, $J = 5.4$ Hz, 1H, NH), 4.66 and 4.61 (s, 2H, H-b), 4.41 and 4.32 (d, $J = 5.4$ Hz, 2H, H-g), 4.00 and 3.93 (s, 2H, H-e), 3.53 (p, $J = 7.2$ Hz, 1H), 3.21–3.05 (m, 2H), 2.47–2.25 (m, 3H), 1.94–1.81 (m, 1H), 1.70–1.61 (m, 4H), 1.42 (m, 2H). ^{13}C NMR (CDCl_3 , 75 MHz): $\delta = 174.4$ (C-f), 169.1 (C-a), 136.0, 129.2, 128.9, 128.7, 128.1, 128.0, 127.8, 127.6, 126.7, 56.5, 52.8, 50.7, 43.6, 40.4, 38.7, 34.9, 32.9, 29.1, 25.1. Anal. Calcd. for $\text{C}_{24}\text{H}_{30}\text{N}_2\text{O}_2\text{S}_2$: C, 65.12; H, 6.83; N, 6.33. Found: C, 65.21; H, 6.77; N, 6.40.

4.2.12. *N*-Benzyl-*N*-(2-(benzylamino)-2-oxo-1-(quinolin-3-yl)ethyl)-5-(*R*)-1,2-dithiolan-3-yl)pentanamide (**12**)



Following the general procedure, a solution of benzylamine (110 μ L,

1 mmol), 3-quinolinecarboxaldehyde (30 mg, 1 mmol), lipoic acid (206 mg, 1 mmol) and benzyl isocyanide (120 μ L, 1 mmol) in MeOH (1 mL), gave, after recrystallization from Et₂O with few drops of MeOH, adduct **12** (407 mg, 72%) as a white solid: mp 124–5 °C; IR (ATR) ν 3296, 3061, 2923, 1639, 1538, 1358 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 8.73 (d, $J = 2.4$ Hz, 1H), 8.24 (s, 1H), 8.00 (d, $J = 8.7$ Hz, 1H), 7.71–7.67 (m, 2H), 7.55–7.50 (t, $J = 8.7$ Hz, 1H), 7.34–7.24 (m, 5H), 7.12–7.08 (m, 3H), 6.97–6.95 (m, 2H), 6.43 (t, $J = 6.0$ Hz, 1H, NH), 6.14 (s, 1H, H-e), 4.77 (d, $J = 17.0$ Hz, 1H, H-d), 4.63 (d, $J = 17.0$ Hz, 1H, H-d), 4.56–4.50 (dd, $J = 14.8$ Hz, 6.0 Hz, 1H, H-g), 4.48–4.41 (dd, $J = 14.8$ Hz, 6.0 Hz, 1H, H-g), 3.56–3.47 (q, $J = 6.7$ Hz, 1H), 3.20–3.05 (m, 2H), 2.48 (m, 3H), 1.91–1.80 (m, 1H), 1.72–1.58 (m, 4H), 1.45–1.36 (m, 2H); ^{13}C NMR (CDCl_3 , 75 MHz) δ 175.0 (C-f), 169.0 (C-a), 151.4, 147.5, 137.7, 136.9, 130.1, 129.0, 128.7, 128.6, 128.0, 127.7, 127.6, 127.3, 127.1, 126.1, 121.9, 118.1, 60.5, 56.3, 50.2, 43.8, 40.2, 38.4, 34.6, 33.7, 29.7, 28.8, 24.9. Anal. Calcd. for $\text{C}_{33}\text{H}_{35}\text{N}_3\text{O}_2\text{S}_2$: C, 69.56; H, 6.19; N, 7.37. Found: C, 69.49; H, 6.25; N, 7.30.

4.3. DPPH radical scavenging activity

The ability of the newly synthesized adducts to scavenge the “stable” free radical DPPH was performed using the method described by Hatano and colleagues [26]. Different concentrations of compounds (0.3 mL) were added to methanolic solution containing DPPH radicals ($1.5 \cdot 10^{-4}$ mol·L⁻¹, 2.7 mL). The mixture was shaken vigorously and kept for 2 h in the dark. The reduction of the DPPH radical was determined by measuring the absorption at 517 nm. The RSA was calculated as a percentage of DPPH discoloration using the equation:

$$\% \text{ RSA} = [(A_{\text{DPPH}} - A_{\text{S}}) / A_{\text{DPPH}}] \times 100,$$

where A_{S} is the absorbance of the solution when the compound has been added at a particular level and A_{DPPH} is the absorbance of the DPPH solution. Mean values from three independent samples were calculated for each compound and standard deviations were less than 5%.

4.4. Oxygen radical absorbance capacity assay

The antioxidant power of compounds **1–12** was performed according the ORAC-FL method described by Dávalos and colleagues [27] using fluorescein as a fluorescent probe with minor modification. Briefly, the antioxidant and fluorescein were incubated in a black 96-well microplate (Nunc) for 15 min at 37 °C. The 2,2'-azobis(amidinopropane) dihydrochloride (AAPH) was then added quickly using the built-in injector of Varioskan Flash plate reader (Thermo Scientific). The fluorescence was measured at 485 nm (excitation wavelength) and 535 nm (emission wavelength) every minute for 60 min. The microplate was automatically shaken prior to each reading. All the reactions were made in triplicate and at least three different assays were performed for each sample. Antioxidant curves (fluorescence versus time) were first normalized to the curve of the blank (without antioxidant) and then, the area under the fluorescence decay curve (AUC) was calculated as: $\text{AUC} = 1 + \sum(f_i/f_0)$, Where f_0 is the initial fluorescence reading at 0 min and f_i is the fluorescence value at time i .

The net AUC corresponding to a sample was calculated as follows:

$$\text{Net AUC} = \text{AUC}_{\text{antioxidant}} - \text{AUC}_{\text{blank}}.$$

Regression equations were calculated by plotting the net AUC against the antioxidant concentration. The ORAC value was obtained by dividing the slope of the latter curve between the slope of the Trolox curve obtained in the same assay. Final ORAC values were expressed as trolox equivalents (μmol of Trolox equivalent/ μmol of adduct). Data are expressed as means \pm SD.

4.5. Inhibition of A β _{1–40} aggregation

For the inhibition of self-mediated A β _{1–40} aggregation experiment, the A β stock solution was diluted with 50 mM phosphate buffer (pH 7.4) to 50 μ M before use. A mixture of the peptide (10 μ L, 25 μ M, final concentration) with or without the tested compound (10 μ L, 20 μ M, final concentration) was incubated at 37 °C for 48 h. Blanks using 50 mM phosphate buffer (pH 7.4) instead of A β with or without inhibitors were also carried out. The sample was diluted to a final volume of 200 μ L with 50 mM glycine–NaOH buffer (pH 8.0) containing thioflavin T (5 μ M). Then the fluorescence intensities were recorded five minutes later (excitation, 450 nm; emission, 485 nm). The percent inhibition of aggregation was calculated by the expression:

$$(1 - \text{IFi/IFc}) \times 100\%$$

in which IFi and IFc are the fluorescence intensities obtained for A β in the presence and absence of inhibitors after subtracting the background, respectively.

4.6. Effect of compounds 8–10 on H₂O₂ (150 μ M)-induced cell death in SH-SY5Y cells

SH-SY5Y cells were seeded in 96-well culture plates at a density of 8×10^4 cells per well in DMEM/F12 (1:1) medium supplemented with 10% fetal bovine serum, $1 \times$ non-essential amino acids, 100 units/mL penicillin and 10 mg/mL streptomycin (Dutscher, France). After 48 h of incubation, the cultures were treated with 100 μ L of the test compounds or DMSO (0.1%) in the same medium. Following 24 h, the cells were co-incubated with H₂O₂ (150 μ M) with or without the tested compounds at noncytotoxic concentrations for an additional period of 24 h. The percent of cell viability was measured using CellTiter 96 Aqueous Non-Radioactive Cell Proliferation (MTS) Assay (Promega, France).

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Appendix A. Supplementary material

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.bioorg.2018.12.029>.

References

- [1] A. Samadi, M. Estrada, C. Pérez, M.I. Rodríguez-Franco, I. Iriepa, I. Moraleda, M. Chioua, J. Marco-Contelles, Pyridonepezils, new dual AChE inhibitors as potential drugs for the treatment of Alzheimer's disease: synthesis, biological assessment, and molecular modeling, *Eur. J. Med. Chem.* 57 (2012) 296–301, <https://doi.org/10.1016/j.ejmech.2012.09.030>.
- [2] L. Ismaili, A. Nadaradjane, L. Nicod, C. Guyon, A. Xicluna, J.-F. Robert, B. Refouvet, Synthesis and antioxidant activity evaluation of new hexahydro-pyrimido[5,4-c]quinoline-2,5-diones and 2-thioxohexahydro-pyrimido[5,4-c]quinoline-5-ones obtained by Biginelli reaction in two steps, *Eur. J. Med. Chem.* 43 (2008) 1270–1275, <https://doi.org/10.1016/j.ejmech.2007.07.012>.
- [3] I. Tomassoli, G. Herlem, F. Picaud, M. Benchekroun, O.M. Bautista-Aguilera, V. Luzet, M.-L. Jimeno, T. Gharbi, B. Refouvet, L. Ismaili, Synthesis, regioselectivity, and DFT analysis of new antioxidant pyrazolo[4,3-c]quinoline-3,4-diones, *Monatshfte Für Chem. – Chem. Mon.* 147 (2016) 1069–1079, <https://doi.org/10.1007/s00706-016-1660-7>.
- [4] M.-Y. Wu, G. Esteban, S. Brogi, M. Shionoya, L. Wang, G. Campiani, M. Unzeta, T. Inokuchi, S. Butini, J. Marco-Contelles, Donepezil-like multifunctional agents: Design, synthesis, molecular modeling and biological evaluation, *Eur. J. Med. Chem.* 121 (2016) 864–879, <https://doi.org/10.1016/j.ejmech.2015.10.001>.
- [5] R. León, A.G. Garcia, J. Marco-Contelles, Recent advances in the multitarget-directed ligands approach for the treatment of Alzheimer's disease, *Med. Res. Rev.* 33 (2013) 139–189, <https://doi.org/10.1002/med.20248>.
- [6] M.J. Oset-Gasque, J. Marco-Contelles, Alzheimer's disease, the "One-Molecule, One-Target" paradigm, and the multitarget directed ligand approach, *ACS Chem. Neurosci.* 9 (2018) 401–403, <https://doi.org/10.1021/acscchemneuro.8b00069>.
- [7] F. Belluti, A. Rampa, S. Gobbi, A. Bisi, Small-molecule inhibitors/modulators of amyloid- β peptide aggregation and toxicity for the treatment of Alzheimer's disease: a patent review (2010–2012), *Expert Opin. Ther. Pat.* 23 (2013) 581–596, <https://doi.org/10.1517/13543776.2013.772983>.
- [8] M. Benchekroun, A. Romero, J. Egea, R. León, P. Michalska, I. Buendía, M.L. Jimeno, D. Jun, J. Janockova, V. Sepsova, O. Soukup, O.M. Bautista-Aguilera, B. Refouvet, O. Ouari, J. Marco-Contelles, L. Ismaili, The antioxidant additive approach for Alzheimer's disease therapy: new ferulic (lipoic) acid plus melatonin modified tacrines as cholinesterases inhibitors, direct antioxidants, and nuclear factor (erythroid-derived 2)-like 2 activators, *J. Med. Chem.* 59 (2016) 9967–9973, <https://doi.org/10.1021/acs.jmedchem.6b01178>.
- [9] M. Benchekroun, L. Ismaili, M. Pudlo, V. Luzet, T. Gharbi, B. Refouvet, J. Marco-Contelles, Donepezil–ferulic acid hybrids as anti-Alzheimer drugs, *Future Med. Chem.* 7 (2015) 15–21, <https://doi.org/10.4155/fmc.14.148>.
- [10] I. Tomassoli, L. Ismaili, M. Pudlo, C. de los Ríos, E. Soriano, I. Colmena, L. Gandía, L. Rivas, A. Samadi, J. Marco-Contelles, B. Refouvet, Synthesis, biological assessment and molecular modeling of new dihydroquinoline-3-carboxamides and dihydroquinoline-3-carbohydrazide derivatives as cholinesterase inhibitors, and Ca channel antagonists, *Eur. J. Med. Chem.* 46 (2011) 1–10, <https://doi.org/10.1016/j.ejmech.2010.08.054>.
- [11] H.W. Querfurth, F.M. LaFerla, Alzheimer's disease, *N. Engl. J. Med.* 362 (2010) 329–344, <https://doi.org/10.1056/NEJMra0909142>.
- [12] M. Mancuso, F. Coppede, L. Migliore, G. Siciliano, L. Murri, Mitochondrial dysfunction, oxidative stress and neurodegeneration, *J. Alzheimers Dis. JAD.* 10 (2006) 59–73.
- [13] R. von Bernhardi, J. Eugenin, Alzheimer's disease: redox dysregulation as a common denominator for diverse pathogenic mechanisms, *Antioxid. Redox Signal.* 16 (2011) 974–1031, <https://doi.org/10.1089/ars.2011.4082>.
- [14] G.J. McBean, M.G. López, F.K. Wallner, Redox-based therapeutics in neurodegenerative disease, *Br. J. Pharmacol.* 174 (2017) 1750–1770, <https://doi.org/10.1111/bph.13551>.
- [15] I. Akritopoulou-Zanze, Isocyanide-based multicomponent reactions in drug discovery, *Curr. Opin. Chem. Biol.* 12 (2008) 324–331, <https://doi.org/10.1016/j.cbpa.2008.02.004>.
- [16] J.E. Biggs-Houck, A. Younai, J.T. Shaw, Recent advances in multicomponent reactions for diversity-oriented synthesis, *Curr. Opin. Chem. Biol.* 14 (2010) 371–382, <https://doi.org/10.1016/j.cbpa.2010.03.003>.
- [17] A. Dömling, W. Wang, K. Wang, Chemistry and biology of multicomponent reactions, *Chem. Rev.* 112 (2012) 3083–3135, <https://doi.org/10.1021/cr100233r>.
- [18] L. Ismaili, M. do Carmo Carreiras, Multicomponent reactions for multitargeted compounds for Alzheimer's disease, *Curr. Top. Med. Chem.* 17 (2017) (accessed July 4, 2018), <http://www.eurekaselect.com/159013/article>.
- [19] V. Nenajdenko, *Isocyanide Chemistry: Applications in Synthesis and Material Science*, John Wiley & Sons, 2012.
- [20] C. Lambruschini, D. Galante, L. Moni, F. Ferraro, G. Gancia, R. Riva, A. Traverso, L. Banfi, C. D'Arrigo, Multicomponent, fragment-based synthesis of polyphenol-containing peptidomimetics and their inhibiting activity on beta-amyloid oligomerization, *Org. Biomol. Chem.* 15 (2017) 9331–9351, <https://doi.org/10.1039/C7OB02182H>.
- [21] J. Dai, R.J. Mumper, Plant phenolics: extraction, analysis and their antioxidant and anticancer properties, *Molecules* 15 (2010) 7313–7352, <https://doi.org/10.3390/molecules15107313>.
- [22] L. Packer, H.J. Tritschler, K. Wessel, Neuroprotection by the metabolic antioxidant α -lipoic acid, *Free Radic. Biol. Med.* 22 (1997) 359–378, [https://doi.org/10.1016/S0891-5849\(96\)00269-9](https://doi.org/10.1016/S0891-5849(96)00269-9).
- [23] R. Wilken, M.O. Zimmermann, A. Lange, A.C. Joerger, F.M. Boeckler, Principles and applications of halogen bonding in medicinal chemistry and chemical biology, *J. Med. Chem.* 56 (2013) 1363–1388, <https://doi.org/10.1021/jm3012068>.
- [24] E. Parisini, P. Metrangola, T. Pilati, G. Resnati, G. Terraneo, Halogen bonding in halocarbon-protein complexes: a structural survey, *Chem. Soc. Rev.* 40 (2011) 2267–2278, <https://doi.org/10.1039/c0cs00177e>.
- [25] M. Chioua, D. Sucunza, E. Soriano, D. Hadjipavlou-Litina, A. Alcázar, I. Ayuso, M.J. Oset-Gasque, M.P. González, L. Monjas, M.I. Rodríguez-Franco, J. Marco-Contelles, A. Samadi, A-aryl-N-alkyl nitrones, as potential agents for stroke treatment: synthesis, theoretical calculations, antioxidant, anti-inflammatory, neuroprotective, and brain-blood barrier permeability properties, *J. Med. Chem.* 55 (2012) 153–168, <https://doi.org/10.1021/jm201105a>.
- [26] T. Hatano, H. Kagawa, T. Yasuhara, T. Okuda, Two new flavonoids and other constituents in licorice root: their relative astringency and radical scavenging effects, *Chem. Pharm. Bull. (Tokyo)* 36 (1988) 2090–2097, <https://doi.org/10.1248/cpb.36.2090>.
- [27] A. Dávalos, C. Gómez-Cordovés, B. Bartolomé, Extending applicability of the oxygen radical absorbance capacity (ORAC–fluorescein) assay, *J. Agric. Food Chem.* 52 (2004) 48–54, <https://doi.org/10.1021/jf0305231>.